APPENDIX F: CCR Certification Form (Suggested Format)

Consumer Confidence Report Certification Form

(to be submitted with a copy of the CCR)

(To certify electronic delivery of the CCR, use the certification form on the State Water Board's

website at http://wv	vw.swrcb.ca.gov/drinking water/certlic/drinkingwater/CCR.shtml)
Water System Name:	Groveland CSD
Water System Number:	5510009
on July1, 2022 to custome system certifies that the in	above hereby certifies that its Consumer Confidence Report was distributed ers (and appropriate notices of availability have been given). Further, the formation contained in the report is correct and consistent with the appropriate a previously submitted to the State Water Resources Control Board,
Certified by:	
Name: Greg Dunn	
Signature: XM/	
Title: CPO	
Phone number: 209.628.9	172
Date: 6/16/22	
	ery used and good-faith efforts taken, please complete the below by bly and fill-in where appropriate:
methods used: ⊠"Good faith" effor the following metho ⊠Posting the C	ts were used to reach non-bill paying consumers. Those efforts included ds: CR on the Internet at www.gcsd.org CR to postal patrons within the service area (attach zip codes used)

This form is provided as a convenience for use to meet the certification requirement of the California Code of Regulations, section 64483(c)

For investor-owned utilities: Delivered the CCR to the California Public Utilities Commission

E-mail: info@gcsd.org Visit our web site at: www.gcsd.org Phone: (209) 962-7161 Fax: (209) 962-4943

Groveland Community Services District Groveland, California, 95321 18966 Ferretti Road P.O. Box 350



2021 Water Quality Report

Groveland Community Services District



Water Conservation Tips

- Automatic dishwashers use 15 gallons for every cycle, regardless of how many dishes are loaded. Get a run for your money and load it to capacity.
- Turn off the tap when brushing your teeth
- Check every faucet in your home for leaks. Just a slow drip can waste 15 to 20 gallons a day and up to slow drip can wasic 7,300 gallons a year.
- food coloring in the tank. Watch for a few minutes see if the color shows up in the bowl. It is not uncommon to lose up to 100 gallons a day from an undetectable toilet leak. Fix it and you could save Check your toilets for leaks by putting a few drops of food coloring in the tank. Watch for a few minutes to more than 30,000 gallons a year.
- Use your water meter to detect hidden leaks. Sturn off all taps and water using appliances. The check the meter after 15 minutes. If it moved, s. Simply Then
- They are inexpensive and by replacing just one, the average family can save 2,900 gallons per year. By using less hot water, you'll also save on your energy bill. Replace shower heads with new, low flow models

District Toilet and Showerhead Rebates

The District currently offers rebates to its customers who replace high flow toilets and shower heads with new low flow ones. Water customers can receive a \$50 rebate per new low flow toilet and customers on Sewer and Water can receive \$100 per toilet, with a maximum of two (2) rebates per household.

Water and Sewer customers can also receive a \$20 rebate for installing a new 1.5 gallons per minute or less showerhead, with a maximum of two (2) per household.

excellent way to learn about water and wastewater issues that directly affect you and everyone in the Groveland, Big Oak Flat, and Pine Mountain Lake areas. Your participation is appreciated. Current information is available on our web site www.gcsd.org. You are invited to attend our regularly scheduled Board meetings held on the second Tuesday of each month, beginning at 10:00 a.m. in the Groveland Community Services District's boardroom, at 18966 Ferretti Road, Groveland, California. GCSD's Board meetings are an

Community Participation



tunnel AWS a safe The Pall Trailer is known as the Alternative Water Supply (AWS) treatment plant. The AWS was installed in 2008. It is capable of producing 600 gallons per minute of treated water drawn outage, or emergency situation, the AV treatment plant is capable of providing a s drinking water supply to all GCSD customers. Mountain Lake. During a The Pall Trailer is known as the Pine from

Sampling Results

The District routinely monitors for contaminants in your drinking water in accordance with federal and state laws. The results contained in this report are for the monitoring period of January 1, 2021, through December 31, 2021.

This report contains results from laboratory testing, excluding contaminants that were not detected, or that were detected at a level below the state's detection level for the purposes of reporting (DLR). This information has been compiled in the tables on the back of this pamphlet to show what these contaminants were

In order to ensure that tap water is safe to drink, the USEPA, and the State Water Resources Control Board prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. State Board regulations also establish limits for contaminants in bottled water that proved the same protection for public health.

The chart lists all of the drinking water contaminants that were detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The State Board allows us to monitor for certain contaminants less than once per year because concentrations of these contaminants do not change frequently. Some of the data, though representative of the water quality, are more than one year old. Any violation of an AL, MCL, MRDL, or TT is asterisked.

Terms Used In This Report

Action Level (Reconcentration of a triggers treatment of system must follow. Level (Regulatory Action ation of a contaminant which, treatment or other requirements Level):

one of the content of the c vel): The exceeded, at a water

MCL (Maximum Contaminant Level): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs (SMCLs) are set to protect the odor, taste and appearance of drinking water.

bottled water, and the bacterial counts approach zero. Because of the high quality of our source water, the District obtained a Filtration Avoidance permit (no filtration process required) on April 22, 1998, and during 2007 and 2008 began using disinfection-by-chloramination and ultraviolet disinfection to kill any pathogens, including disinfection to kill any pathogens, including Cryptosporidium and Giardia, that may be present in its surface water supply.



GCSD obtains the majority of its water from the San Francisco Public Utilities Commission's (SFPUC) Hetch Hetchy Reservoir supply by pumping from a deep conveyance tunnel southeast of town, known as the Mountain Tunnel. The water originates in Yosemite National Park as snow melt from a large pristine watershed in the High Sierra. With pristine watershed in the High Sierra. With controlled human contact and granite-type geology, the mineral content of this water is lower than most

Where Your Water Comes From

MCLG (Maximum Contaminant Level Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. EPA.

MRDL (Maximum Residual Disinfectant Level): level of a disinfectant added for water treatment that may exceeded at the consumer's tap.

which there is no known or MRDLGs are set by the U.S. EPA. MRDLG (Maximum Residual Disinfectant Level Goal): The level of a disinfectant added for water treatment below which there is no known or expected risk to health.

ND (Not Detected): Indicates that the substance was found by laboratory analysis.

NS: No standard.

the clarity, or turbidity, of water. Turb NTU is just noticeable to the average per NTU (Nephelometric Turbidity Units): Measurement the clarity, or turbidity, of water. Turbidity in excess of the clarity, or turbidity in excess of the clarity, or turbidity in excess of the clarity. ent of of 5

monitoring an requirements. PDWS (Primary Drinking Water Standard): MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment

PHG (**Public Health Goal**): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California EPA.

Ppb/μg (parts per billion): One part billion parts water (or micrograms per liter). of substance per

TT (Treatment Technique): A required process intended to reduce the level of a contaminant in drinking water.

Ppm/mg (parts per million): One part million parts water (or milligrams per liter).

of substance

Variances and Exemptions: State Board permission to exceed and MCL or not comply with a treatment or other requirements that a water system must follow.

ppm: parts per million of milligrams per liter (mg/L)

ppb: parts per billion of micorograms per liter (µg/L)

ppt: parts per trillion or nanograms per liter (ng/L)

Ppq: pars per quadrillion or pictogram per liter (pg/L)

pCi/L: picocuries per liter (a measure of radiation)

2021 WATER QUALITY DATA

Groveland Community Services District, Groveland, California

PRIMARY DRINKING WATER STANDARDS

Este informe contiene informacion muy importante sobre su agua potable. Traduzcalo o hable con alguien que lo entienda bien.

Substance or Parameter	Unit	MCL/MRDL	(MCLG)	Range	Average	Typical Sources in Drinking Water
CLARITY						
Turbidity—Raw Source Water	NTU	5	NS	0.14-0.91	0.2	Primarily related to soil runoff (erosion) which is made up of suspended matter that interferes with light
Turbidity—Finished Water	NTU	5	NS	0.13-0.97	0.3	Primarily related to soil runoff (erosion) which is made up of suspended matter that interferes with light
MICROBIOLOGICAL						
Total Coliform —Raw Water	#	(a)	0	<2-170	13.2	Naturally present in the environment from decomposition of organic matter; may be an indication of fecal waste
Fecal Coliform—Raw Water	#	(b)	0	<2-17	2.2	Related to human and animal waste
DISINFECTION BY PRODUCTS AND D	ISINFECTANT F	RESIDUALS				
Total Trihalomethanes (TTHMs)	μg/L	80	NS	14.1-15.9	14.7	Byproducts of drinking water disinfection using chlorine; upgrades to the treatment process have reduced TTHMs to below MCL's
Total Haloacetic Acids (HAAs)	μg/L	60	NS	19.9-35.8	27.3	Byproducts of drinking water disinfection using chlorine; upgrades to the treatment process have reduced HAAs to below MCL's
Chlorine	mg/L	4.0 (as Cl2)	4.0	1.3-3.1	2.7	Drinking water disinfectant added for treatment
Chloramines (Distribution System)*	mg/L	4.0 (as Cl2)	4.0	1.4-3.0	2.5	Drinking water disinfectant added for treatment
INORGANIC CHEMICALS				90th Percentile		
Copper (August 2020)	mg/L	1.3	0.17	ND076	0.02	Internal corrosion of household plumbing systems, erosion of natural rock/soil deposits, and leaching from wood preservatives
Lead [♥] (August 2020)	mg/L	15	2	ND	ND	Internal corrosion of household plumbing systems, erosion of natural rock/soil deposits, and discharges from industrial manufacturers
SECONDARY DRINKING WATER STA	NDARD					
Substance or Parameter						Typical Sources in Drinking Water
Color	unit	15	NS	3-4	3.5	From naturally occurring organic materials such as leaves, pine needles, and wood
		•	NS	0-3	1.5	Γ
	TON	3				From naturally occurring organic materials
Specific Conductance	μS/cm	900	NS	16-98.2	57.1	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water
Specific Conductance Total Dissolved Solids (TDS)	μS/cm mg/L	900 500	NS NS	16-98.2 10-67	57.1 38.5	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks)
Odor Specific Conductance Total Dissolved Solids (TDS) Sulfate	μS/cm	900	NS	16-98.2	57.1	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water
Specific Conductance Total Dissolved Solids (TDS) Sulfate OTHER	μS/cm mg/L	900 500	NS NS	16-98.2 10-67	57.1 38.5	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks)
Specific Conductance Total Dissolved Solids (TDS) Sulfate OTHER Substance or Parameter	μS/cm mg/L mg/L	900 500 250	NS NS NS	16-98.2 10-67 0-6	57.1 38.5 3	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks) Typical Sources in Drinking Water
Specific Conductance Total Dissolved Solids (TDS) Sulfate OTHER Substance or Parameter Alkalinity (as CaCO ₃)	μS/cm mg/L mg/L mg/L	900 500 250 NS	NS NS NS	16-98.2 10-67 0-6	57.1 38.5 3	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks) Typical Sources in Drinking Water From natural sources and dissolved minerals
Specific Conductance Total Dissolved Solids (TDS) Sulfate OTHER Substance or Parameter Alkalinity (as CaCO ₃) Hardness (as CaCO ₃)	μS/cm mg/L mg/L	900 500 250 NS	NS NS NS NS	16-98.2 10-67 0-6 8.2-32.1 0-24.9	57.1 38.5 3 24.3 12.5	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks) Typical Sources in Drinking Water From natural sources and dissolved minerals From naturally occurring dissolved substances (Ca ²⁺ , Mg ²⁺ , Sr ²⁺ , Fe ²⁺ , Mn ²⁺) that come in contact with water
Specific Conductance Total Dissolved Solids (TDS) Sulfate OTHER Substance or Parameter Alkalinity (as CaCO ₃) Hardness (as CaCO ₃) Chloride	μS/cm mg/L mg/L mg/L mg/L mg/L mg/L mg/L	900 500 250 NS NS 250	NS NS NS NS NS NS NS	16-98.2 10-67 0-6 8.2-32.1 0-24.9 ND-5.1	57.1 38.5 3 24.3 12.5 2.6	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks) Typical Sources in Drinking Water From natural sources and dissolved minerals From naturally occurring dissolved substances (Ca ²⁺ , Mg ²⁺ , Sr ²⁺ , Fe ²⁺ , Mn ²⁺) that come in contact with water From naturally occurring and dissolved minerals
Specific Conductance Total Dissolved Solids (TDS) Sulfate OTHER Substance or Parameter Alkalinity (as CaCO ₃) Hardness (as CaCO ₃) Chloride Sodium	μS/cm mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L unit	900 500 250 NS NS 250 NS	NS NS NS NS NS NS NS NS NS	16-98.2 10-67 0-6 8.2-32.1 0-24.9 ND-5.1 ND-3.74	57.1 38.5 3 24.3 12.5 2.6 1.9	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks) Typical Sources in Drinking Water From natural sources and dissolved minerals From naturally occurring dissolved substances (Ca ²⁺ , Mg ²⁺ , Sr ²⁺ , Fe ²⁺ , Mn ²⁺) that come in contact with water From naturally occurring and dissolved minerals From naturally occurring and dissolved minerals From naturally occurring and dissolved minerals
Specific Conductance Total Dissolved Solids (TDS) Sulfate OTHER Substance or Parameter Alkalinity (as CaCO ₃) Hardness (as CaCO ₃) Chloride Sodium Calcium	μS/cm mg/L mg/L mg/L mg/L mg/L mg/L unit mg/L	900 500 250 NS NS 250 NS NS	NS	16-98.2 10-67 0-6 8.2-32.1 0-24.9 ND-5.1 ND-3.74 ND-5.84	57.1 38.5 3 24.3 12.5 2.6 1.9 2.9	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks) Typical Sources in Drinking Water From natural sources and dissolved minerals From naturally occurring dissolved substances (Ca ²⁺ , Mg ²⁺ , Sr ²⁺ , Fe ²⁺ , Mn ²⁺) that come in contact with water From naturally occurring and dissolved minerals
Specific Conductance Total Dissolved Solids (TDS) Sulfate OTHER Substance or Parameter Alkalinity (as CaCO ₃) Hardness (as CaCO ₃) Chloride Sodium Calcium Ph (Distribution System)	μS/cm mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L unit	900 500 250 NS NS 250 NS NS	NS	16-98.2 10-67 0-6 8.2-32.1 0-24.9 ND-5.1 ND-3.74 ND-5.84 6.6-10.2	57.1 38.5 3 24.3 12.5 2.6 1.9 2.9 8.8	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks) Typical Sources in Drinking Water From natural sources and dissolved minerals From naturally occurring dissolved substances (Ca²+, Mg²+, Sr²+, Fe²+, Mn²+) that come in contact with water From naturally occurring and dissolved minerals Affected by alkaline sources, atmospheric CO₂, organic matter, and acidity from mineral sources—distilled water has a 7.0 pH
Specific Conductance Total Dissolved Solids (TDS) Sulfate OTHER Substance or Parameter Alkalinity (as CaCO ₃) Hardness (as CaCO ₃) Chloride Sodium Calcium Ph (Distribution System) Potassium	μS/cm mg/L mg/L mg/L mg/L mg/L unit mg/L unit mg/L	900 500 250 NS NS 250 NS NS NS NS	NS	16-98.2 10-67 0-6 8.2-32.1 0-24.9 ND-5.1 ND-3.74 ND-5.84 6.6-10.2 ND-2.09	57.1 38.5 3 24.3 12.5 2.6 1.9 2.9 8.8 1.0	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks) Typical Sources in Drinking Water From natural sources and dissolved minerals From naturally occurring dissolved substances (Ca ²⁺ , Mg ²⁺ , Sr ²⁺ , Fe ²⁺ , Mn ²⁺) that come in contact with water From naturally occurring and dissolved minerals From naturally occurring and dissolved minerals From naturally occurring and dissolved minerals Affected by alkaline sources, atmospheric CO ₂ , organic matter, and acidity from mineral sources—distilled water has a 7.0 pH From naturally occurring and dissolved mineral
Specific Conductance Total Dissolved Solids (TDS) Sulfate OTHER Substance or Parameter Alkalinity (as CaCO ₃) Hardness (as CaCO ₃) Chloride Sodium Calcium Ph (Distribution System) Potassium	mg/L mg/L mg/L mg/L mg/L mg/L mg/L unit mg/L unit	900 500 250 NS NS 250 NS NS	NS N	16-98.2 10-67 0-6 8.2-32.1 0-24.9 ND-5.1 ND-3.74 ND-5.84 6.6-10.2 ND-2.09 ND-2.5	57.1 38.5 3 24.3 12.5 2.6 1.9 2.9 8.8 1.0 1.3	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks) Typical Sources in Drinking Water From natural sources and dissolved minerals From naturally occurring dissolved substances (Ca ²⁺ , Mg ²⁺ , Sr ²⁺ , Fe ²⁺ , Mn ²⁺) that come in contact with water From naturally occurring and dissolved minerals From naturally occurring and dissolved minerals From naturally occurring and dissolved minerals Affected by alkaline sources, atmospheric CO ₂ , organic matter, and acidity from mineral sources—distilled water has a 7.0 pH From naturally occurring and dissolved mineral From naturally occurring and dissolved mineral
Specific Conductance Total Dissolved Solids (TDS)	μS/cm mg/L mg/L mg/L mg/L mg/L unit mg/L unit mg/L	900 500 250 NS NS 250 NS NS NS NS	NS	16-98.2 10-67 0-6 8.2-32.1 0-24.9 ND-5.1 ND-3.74 ND-5.84 6.6-10.2 ND-2.09	57.1 38.5 3 24.3 12.5 2.6 1.9 2.9 8.8 1.0	From naturally occurring dissolved solids that form ions in water, an indication of the dissolved mineral content of water From runoff and leaching from natural deposits (soil and rocks) From runoff and leaching from natural deposits (soil and rocks) Typical Sources in Drinking Water From natural sources and dissolved minerals From naturally occurring dissolved substances (Ca ²⁺ , Mg ²⁺ , Sr ²⁺ , Fe ²⁺ , Mn ²⁺) that come in contact with water From naturally occurring and dissolved minerals From naturally occurring and dissolved minerals From naturally occurring and dissolved minerals Affected by alkaline sources, atmospheric CO ₂ , organic matter, and acidity from mineral sources—distilled water has a 7.0 pH From naturally occurring and dissolved mineral

The tables above list all of the drinking water substances and parameters that were detected in 2021.

MCLs for Total and Fecal Coliform
(a) - For 40 samples/month: No more than 5.0% of monthly samples may be positive; for <40 samples/month; no more than 1 positive sample

(b) - A routine sample and repeat samples are total coliform positive, and one of these is also fecal coliform or E. Coli positive Results for total and fecal coliform are for raw water sources; they do not represent the drinking water concentrations of these substances <u>Water Hardness Classification</u> (Note: GCSD's water is soft) 0—75 mg/L = Soft

75—150 mg/L = Moderately hard

150—300 mg/L = Hard>300 mg/L = Very hard

* Results for TTHM and HAA samples are averaged over four quarters. Results indicate levels well below the MCL over many years may experience liver, kidney, or central nervous system problems, and may have an increased risk of getting cancer. Some people who drink water containing HAAs in excess of the MCL over many years may

Drinking water, including bottled water, may reasonably be expected to contaminants. The presence of contaminants and potential health effects can be obtained by calling the USEPA's Safe Drinking Water Hotline (1-800-426-4791). Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advise about drinking water from their health care providers. USEPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium or other microbial contaminants and potential health effects can be obtained by calling the same. If you have health issues concerning the consumption of this water, you may wish to consult your doctor.

Note: During 2021 we failed to pull nitrate samples at the Second Garrotte shaft location and therefore cannot be sure of nitrate quality during that time. However, after completion of the Second Garrotte Clearwell Rehabilitation Project in March 2022, we pulled nitrate samples with the result being Non-Detect.

Please share this information with all people who drink this water, especially those who may not have received this notice directly (for example, people in apartments, nursing homes, schools, and businesses). You can do this by posting this public notice in a public place or distributing copies by hand or mail.

For questions regarding this report, please contact the District's Chief Plant Operator Greg Dunn at 209-962-7161 Ext. 16.

have an increased risk of getting cancer.

Some people who use water containing chlorine well in excess of the MRDL could experience irritating effects to their eyes and nose. Some people who drink water containing chlorine in excess of the MRDL could experience stomach discomfort.

^{*} Some people who use water containing chloramines well in excess of the MRDL could experience irritating effects to their eyes and nose. Some people who drink water containing chloramines well in excess of the MRDL could experience stomach discomfort or anemia.

^{*20} samples were collected by the GCSD during August 2020 (the minimum required). No samples were positive for lead. If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. GCSD is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing or cooking. If you are concerned about lead in your water, you may wish to have it tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.