ANNUAL WATER Reporting Year 2022 Presented By **Tuolumne Utilities District** Este informe contiene información muy importante sobre su agua potable. Tradúzcalo o hable con alguien que lo entienda bien. PWS ID#: 11



Our Mission Continues

We are once again pleased to present our annual water quality report covering all testing performed between January 1 and December 31, 2022. Over the years, we have dedicated ourselves to producing drinking water that meets all state and federal standards. We continually strive to adopt new methods for delivering the best-quality drinking water to you. As new challenges to drinking water safety emerge, we remain vigilant in meeting the goals of source water protection, water conservation, and community education while continuing to serve the needs of all our water users. Please remember that we are always available should you ever have any questions or concerns about your water.

Thousands have lived without

love, not one without water."

-W.H. Auden

Where Does My Water Come From?

The most important factor in water quality is the source. There are two sources of water supply from which Tuolumne Utilities District (TUD) receives its water: surface water that originates from rainfall and runoff from snowpack in the Sierra Nevada Mountains and groundwater wells throughout our water systems. The district is comprised of 11 water service areas that

include 11 surface water treatment plants and 12 active wells. These produce and supply quality drinking water to their service areas.

Our surface water is delivered to TUD treatment plants from the South Fork of the Stanislaus River at Lyons Reservoir via the Tuolumne Main Canal by agreement

with Pacific Gas and Electric Company (PG&E), which owns and operates Pinecrest Lake, Lyons Reservoir, and the Tuolumne Main Canal. Approximately 96 percent of TUD's annual water needs is supplied by surface water from Lyons Reservoir and Pinecrest Lake; the other 4 percent comes from groundwater wells, either as a primary or a backup source. To learn more about our watershed online, visit U.S. EPA's How's My Waterway at https://www.epa.gov/waterdata/hows-my-waterway.

Important Health Information

Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants may be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. The U.S. EPA/CDC

(Centers for Disease Control and Prevention) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial con-

other microbial contaminants are available from the Safe Drinking Water Hotline at (800) 426-4791 or http://water.epa.gov/drink/hotline.

Protecting Your Water

Bacteria are a natural and important part of our world. There are around 40 trillion bacteria living in each of us; without them, we would not be able to live healthy lives. Coliform bacteria are common in the environment and generally not harmful themselves. The presence of this bacterial form in drinking water is a concern, however, because it indicates that the water may be

contaminated with other organisms that can cause disease.

In 2016 the U.S. EPA passed a regulation called the Revised Total Coliform Rule, which requires water systems to take additional steps to ensure the integrity of the drinking water distribution system by monitoring for the presence of bacteria

like total coliform and E. coli. The rule requires more stringent standards than the previous regulation, and it requires water systems that may be vulnerable to contamination to have procedures in place that will minimize the incidence of contamination. Water systems that exceed a specified frequency of total coliform occurrences are required to conduct an assessment and correct any problems quickly. The U.S. EPA anticipates greater public health protection under this regulation due to its more preventive approach to identifying and fixing problems that may affect public health.

Though we are fortunate in having the highest-quality drinking water, our goal is to eliminate all potential pathways of contamination into our distribution system, and this requirement helps us accomplish that goal.

Information on the Internet

The U.S. EPA (https://goo.gl/TFAMKc) and the Centers for Disease Control and Prevention (www.cdc.gov) websites provide a substantial amount of information on many issues relating to water resources, water conservation, and public health. Also, the Division of Drinking Water and Environmental Management website (https://goo.gl/kGepu4) provides complete and current information on water issues in California, including valuable information about our watershed.

QUESTIONS? For more information about this report, or any questions relating to your drinking water, please call Angel Tarango, Regulatory Compliance Specialist, at (209) 532-5536, extension 537.

What type of container is best for storing water?

Consumer Reports has consistently advised that glass or BPA-free plastics such as polyethylene are the safest choices. To be on the safe side, do not use any container with markings on the recycle symbol showing 7PC (that's code for BPA). You could also consider using stainless steel or aluminum with BPA-free liners.

How much emergency water should I keep?

Typically, one gallon per person per day is recommended. For a family of four, that would be 12 gallons for three days. Humans can survive without food for one month but can only survive one week without water.

How long can I store drinking water?

The disinfectant in drinking water will eventually dissipate, even in a closed container. If that container housed bacteria prior to filling up with the tap water, the bacteria may continue to grow once the disinfectant has dissipated. Some experts believe that water can be stored up to six months before needing to be replaced. Refrigeration will help slow the bacterial growth.

How long does it take a water supplier to produce one glass of treated drinking water?

It can take up to 45 minutes to produce a single glass of drinking water.

How many community water systems are there in the U.S.?

About 53,000 public water systems across the United States process 34 billion gallons of water per day for home and commercial use. Eighty-five percent of the population is served by these systems.

Which household activity wastes the most water?

Most people would say the majority of water use comes from showering or washing dishes; however, toilet flushing is by far the largest single use of water in a home (accounting for 40 percent of total water use). Toilets use about 4 to 6 gallons per flush, so consider an ultra-low-flow (ULF) toilet, which requires only 1.5 gallons.

Substances That Could Be in Water

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material and can pick up substances resulting from the presence of animals or from human activity.

In order to ensure that tap water is safe to drink, the U.S. Environmental Protection Agency (U.S. EPA) and the State Water Resources Control Board (State Board) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. The U.S. Food and Drug Administration regulations and California law also establish limits for contaminants in bottled water that provide the same protection for public health. Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk.

Contaminants that may be present in source water include:

Microbial Contaminants, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife;

Inorganic Contaminants, such as salts and metals, that can be naturally occurring or can result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming;

Pesticides and Herbicides, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses;

Organic Chemical Contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production and which can also come from gas stations, urban stormwater runoff, agricultural applications, and septic systems;

Radioactive Contaminants, which can be naturally occurring or can be the result of oil and gas production and mining activities.

More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline at (800) 426-4791.



What's a Cross-Connection?

Cross-connections that contaminate drinking water distribution lines are a major concern. A cross-connection is formed at any point where a drinking water line connects to equipment (boilers), systems containing chemicals (air-conditioning systems, fire sprinkler systems, irrigation systems), or water sources of questionable quality. Cross-connection contamination can occur when the pressure in the equipment or system is greater than the pressure inside the drinking water line (backpressure). Contamination can also occur when the pressure in the drinking water line drops due to fairly routine occurrences (main breaks, heavy water demand), causing contaminants to be sucked out from the equipment and into the drinking water line (back-siphonage).

Outside water taps and garden hoses tend to be the most common sources of cross-connection contamination at home. The garden hose creates a hazard when submerged in a swimming pool or attached to a chemical sprayer for weed killing. Garden hoses that are left lying on the ground may be contaminated by fertilizers, cesspools, or garden chemicals. Improperly installed valves in your toilet could also be a source of cross-connection contamination.

Community water supplies are continuously jeopardized by cross-connections unless appropriate valves, known as backflow prevention devices, are installed and maintained. We have surveyed industrial, commercial, and institutional facilities in the service area to make sure that potential cross-connections are identified and eliminated or protected by a backflow preventer. We also inspect and test backflow preventers to make sure that they provide maximum protection. For more information on backflow prevention, contact the Safe Drinking Water Hotline at (800) 426-4791.

Community Participation

The public is encouraged to attend the district's regularly scheduled board meetings, which occur on the second and fourth Tuesday of each month at 9:00 a.m. These meetings are held in accordance



with the federal Americans with Disabilities Act. During the board meeting, members of the public who wish to provide public comment will be invited to do so by the board president. Members of the public who are physically present will be called first. Members of the public who wish to provide remote public comment may do so via Zoom or by phone. The public may also view the board meetings at the district's website, https://tudwater.com/boards-of-directors/meeting-agenda-minutes-video/.

Tip Top Tap

The most common signs that your faucet or sink is affecting the quality of your drinking water are discolored water, sink or faucet stains, a buildup of particles, unusual odors or tastes, and a reduced flow of water. The solutions to these problems may be in your hands.

Kitchen Sink and Drain

Handwashing, soap scum buildup, and the handling of raw meats and vegetables can contaminate your sink. Clogged drains can lead to unclean sinks and backed-up water in which bacteria (i.e., pink or black slime growth) can grow and contaminate the sink area and faucet, causing a rotten egg odor. Disinfect and clean the sink and drain area regularly and flush with hot water.

Faucets, Screens, and Aerators

Chemicals and bacteria can splash and accumulate on the faucet screen and aerator, which are located on the tip of faucets and can collect particles like sediment and minerals, resulting in a decreased flow from the faucet. Clean and disinfect the aerators or screens on a regular basis.

Check with your plumber if you find particles in the faucet screen, as they could be pieces of plastic from the hot water heater dip tube. Faucet gaskets can break down and cause black, oily slime. If you find this slime, replace the faucet gasket with a higher-quality product. White scaling or hard deposits on faucets and showerheads may be caused by water with high levels of calcium carbonate. Clean these fixtures with vinegar or use water softening to reduce the calcium carbonate levels in the hot water system.

Water Filtration/Treatment Devices

A smell of rotten eggs can be a sign of bacteria on the filters or in the treatment system. The system can also become clogged over time, so regular filter replacement is important. (Remember to replace your refrigerator filter!)

Lead in Home Plumbing

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. We are responsible for providing high-quality drinking water, but we cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to two minutes before using water for drinking or cooking. (If you do so, you may wish to collect the flushed water and reuse it for another beneficial purpose, such as watering plants.) If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline at (800) 426-4791 or www.epa. gov/safewater/lead.

Water Main Flushing

Distribution mains (pipes) convey water to homes, businesses, and hydrants in your neighborhood. The water entering distribution mains is of very high quality; however, water quality can deteriorate in areas of the distribution mains over time. Water main flushing is the process of cleaning the interior of water distribution mains by sending a rapid flow of water through the mains.

Flushing maintains water quality in several ways. For example, flushing removes sediments like iron and manganese. Although iron and manganese do not pose health concerns, they can affect the taste, clarity, and color of the water. Additionally, sediments can shield microorganisms from the disinfecting power of chlorine, contributing to the growth of microorganisms within distribution mains. Flushing helps remove

stale water and ensures the presence of fresh water with sufficient dissolved oxygen and disinfectant levels and an acceptable taste and smell.

During flushing operations in your neighborhood, some short-term deterioration of water quality, though uncommon, is possible. You should avoid tap water for household uses at that time. If you do use the tap, allow your cold water to run for a few minutes at full velocity before use, and avoid using hot water to prevent sediment accumulation in your hot water tank. Please contact us if you have any questions or if you would like more information on our water main flushing schedule.

FOG (Fats, Oils, and Grease)

You may not be aware of it, but every time you pour fat, oil, or grease (FOG) down your sink (e.g., bacon grease), you are contributing to a costly problem in the sewer collection system. FOG coats the inner walls of the plumbing in your house as well as the walls of underground piping throughout the community. Over time, these greasy materials build up and form blockages in pipes, which can lead to wastewater backing up into parks, yards, streets, and storm drains. These backups allow FOG to contaminate local waters, including drinking water. Exposure to untreated wastewater is a public health hazard. FOG discharged into septic systems and drain fields can also cause malfunctions, resulting in more frequent tank pump-outs and other expenses.

Communities spend billions of dollars every year to unplug or replace grease-blocked pipes, repair pump stations, and clean up costly and illegal wastewater spills. Here are some tips that you and your family can follow to help maintain a well-run system now and in the future:

NEVER:

- Pour fats, oil, or grease down the house or storm drains.
- Dispose of food scraps by flushing them.
- Use the toilet as a wastebasket.

ALWAYS:

- Scrape and collect fat, oil, and grease into a waste container, such as an empty coffee can, and dispose of it with your garbage.
- Place food scraps in waste containers or garbage bags for disposal with solid wastes.
- Place a wastebasket in each bathroom for solid wastes like disposable diapers, creams and lotions, and personal hygiene products, including nonbiodegradable wipes.

What Are PFAS?

Per- and polyfluoroalkyl substances (PFAS) are a group of manufactured chemicals used worldwide since the 1950s to make fluoropolymer coatings and products that resist heat, oil, stains, grease, and water. During production and use, PFAS can migrate into the soil, water, and air. Most PFAS do not break down; they remain in the environment, ultimately finding their way into drinking water. Because of their widespread use and their persistence in the environment, PFAS are found all over the world at low levels. Some PFAS can build up in people and animals with repeated exposure over time.

The most commonly studied PFAS are perfluorooctanoic acid (PFOA) and perfluorooctane sulfonic acid (PFOS). PFOA and PFOS have been phased out of production and use in the United States, but other countries may still manufacture and use them.

Some products that may contain PFAS include:

- Some grease-resistant paper, fast food containers/wrappers, microwave popcorn bags, pizza boxes
- Nonstick cookware
- Stain-resistant coatings used on carpets, upholstery, and other fabrics
- Water-resistant clothing
- Personal care products (shampoo, dental floss) and cosmetics (nail polish, eye makeup)
- Cleaning products
- Paints, varnishes, and sealants

Even though recent efforts to remove PFAS have reduced the likelihood of exposure, some products may still contain them. If you have questions or concerns about products you use in your home, contact the Consumer Product Safety Commission at (800) 638-2772. For a more detailed discussion on PFAS, please visit http://bit.ly/3Z5AMm8.

Fixtures with Green Stains

A green or blue-green stain on kitchen or bathroom fixtures is caused by tiny amounts of copper that dissolve in your home's copper plumbing system when the water sits unused overnight. Copper staining may be the result of a leaky faucet or a faulty toilet flush valve, so be sure your plumbing is in good working order.

Copper stains may also be caused by overly hot tap water. Generally speaking, you should maintain your water temperature at a maximum of 120 degrees Fahrenheit. You should consult the owner's manual for your heater or check with your plumber to determine your current heat setting. Lowering your water temperature will reduce the staining problem and save you money on your energy bill.

Also keep in mind that a tap that is used often throughout the day usually will not produce copper stains, so if you flush the tap for a minute or so before using the water for cooking or drinking, copper levels will be reduced.

Water Conservation Tips

You can play a role in conserving water and save yourself money in the process by becoming conscious of the amount of water your household is using and looking for ways to use less whenever you can. It is not hard to conserve water. Here are a few tips:

- Automatic dishwashers use 15 gallons for every cycle, regardless of how many dishes are loaded. So get a run for your money and load it to capacity.
- Turn off the tap when brushing your teeth.
- Check every faucet in your home for leaks. Just a slow drip can waste 15 to 20 gallons a day. Fix it and you can save almost 6,000 gallons per year.
- Check your toilets for leaks by putting a few drops of food coloring in the tank. Watch for a few minutes to see if the color shows up in the bowl. It is not uncommon to lose up to 100 gallons a day from an invisible toilet leak. Fix it and you save more than 30,000 gallons a year.
- Use your water meter to detect hidden leaks. Simply turn off all taps and water-using appliances. Then check the meter after 15 minutes. If it moved, you have a leak.

What Causes the Pink Stain on Bathroom Fixtures?

The reddish-pink color frequently noted in bathrooms on shower stalls, tubs, tile, toilets, sinks, and toothbrush holders and on pets' water bowls is caused by the growth of the bacterium Serratia marcescens. Serratia is commonly isolated from soil, water, plants, insects, and vertebrates (including humans). The bacteria can be introduced into the house through any of the abovementioned sources. The bathroom provides a perfect environment (moist and warm) for bacteria to thrive.

The best solution to this problem is to clean and dry these surfaces to keep them free from bacteria. Chlorine-based compounds work best, but keep in mind that abrasive cleaners may scratch fixtures, making them more susceptible to bacterial growth. Chlorine bleach can be used periodically to disinfect the toilet and help eliminate the occurrence of the pink residue. Keeping bathtubs and sinks wiped down using a solution that contains chlorine will also help to minimize its occurrence. Serratia will not survive in chlorinated drinking water.

Naturally Occurring Bacteria

The simple fact is bacteria and other microorganisms inhabit our world. They can be found all around us: in our food, on our skin, in our bodies, and in the air, soil, and water. Some are harmful to us, and some are not. Coliform bacteria are common in the environment and generally not harmful themselves. The presence of this bacterial form in drinking water is a concern because it indicates that the water may be contaminated with other organisms that can cause disease. Throughout the year, we tested many water samples for coliform bacteria. In that time, none of the samples came back positive for the bacteria.

Federal regulations require that public water that tests positive for coliform bacteria must be further analyzed for fecal coliform bacteria. Fecal coliform are present only in human and animal waste. Because these bacteria can cause illness, it is unacceptable for fecal coliform to be present in water at any concentration. Our tests indicate no fecal coliform is present in our water.

Water Stress

Water stress occurs when the demand for water exceeds the amount available during a certain period or when poor water quality restricts its use. Water stress causes deterioration of freshwater resources in terms of quantity (aquifer overexploitation, dry rivers, etc.) and quality (eutrophication, organic matter pollution, saline intrusion, etc.).

According to the World Resources Institute (WRI; www. wri.org), the Middle East and North Africa remain the most water-stressed regions on Earth. However, several states in the western half of the U.S. are similarly experiencing extremely high levels of water stress from overuse. It is clear that even in countries with low overall water stress, individual communities may still be experiencing extremely stressed conditions. For example, South Africa and the United States rank #48 and #71 on WRI's list, respectively, yet the Western Cape (the state home to Cape Town) and New Mexico experience extremely high stress levels.

There are undeniably worrying trends in water quality. But by taking action now and investing in better management, we can solve water issues before it's too late.

What Are PPCPs?

When cleaning out your medicine cabinet, what do you do with your expired pills? Many people flush them down the toilet or toss them into the trash. Although this seems convenient, these actions could threaten our water supply.

Recent studies are generating a growing concern over pharmaceuticals and personal care products (PPCPs) entering water supplies. PPCPs include human and veterinary drugs (prescription or over-the-counter) and consumer products, such as cosmetics, fragrances, lotions, sunscreens, and housecleaning products. From 2006 to 2010, the number of U.S. prescriptions increased 12 percent to a record 3.7 billion, while nonprescription drug purchases held steady at around 3.3 billion. Many of these drugs and personal care products do not biodegrade and may persist in the environment for years.

The best and most cost-effective way to ensure safe water at the tap is to keep our source waters clean. Never flush unused medications down the toilet or sink. Instead, check to see if the pharmacy where you made your purchase accepts medications for disposal, or contact your local health department for information on proper disposal methods and drop-off locations. You can also go online (https://goo.gl/aZPgeB) to find more information about disposal locations in your area.

Failure in Flint

The national news coverage of water conditions in Flint, Michigan, has created a great deal of confusion and consternation. The water there has been described as being corrosive; images of corroded batteries and warning labels on bottles of acids come to mind. But is corrosive water bad?

Corrosive water can be defined as a condition of water quality that will dissolve metals (iron, lead, copper, etc.) from metallic plumbing at an excessive rate. There are a few contributing factors, but generally speaking, corrosive water has a pH of less than 7; the lower the pH, the more acidic, or corrosive, the water becomes. (By this definition, many natural waterways throughout the country can be described as corrosive.) While all plumbing will be somewhat affected over time by the water it carries, corrosive water will damage plumbing much more rapidly than water with low corrosivity.

By itself, corrosive water is not a health concern; your morning glass of orange juice is considerably more corrosive than the typical lake or river. What is of concern is that drinking water with elevated levels of the dissolved metals increases adverse health risks. And therein lies the problem.

Public water systems are required to maintain their water at optimal conditions to prevent it from reaching corrosive levels. Rest assured that we routinely monitor our water to make sure that what happened in Flint never happens here.

Safeguard Your Drinking Water

Protection of drinking water is everyone's responsibility. You can help protect your community's drinking water source in several ways:

- Eliminate excess use of lawn and garden fertilizers and pesticides they contain hazardous chemicals that can reach your drinking water source.
- Pick up after your pets.
- If you have your own septic system, properly maintain it to reduce leaching to water sources, or consider connecting to a public water system.
- Dispose of chemicals properly; take used motor oil to a recycling center.
- Volunteer in your community. Find a watershed or wellhead protection organization in your community and volunteer to help. If there are no active groups, consider starting one. Use U.S. EPA's Adopt Your Watershed to locate groups in your community.
- Organize a storm drain stenciling project with others in your neighborhood. Stencil a message next to the street drain reminding people: "Dump No Waste – Drains to River" or "Protect Your Water." Produce and distribute a flyer for households to remind residents that storm drains dump directly into your local water body.

Count on Us

Delivering high-quality drinking water to our customers involves far more than just pushing water through pipes. Water treatment is a complex, time-consuming process. Because tap water is highly regulated by state and federal laws, water treatment plant and system operators must be licensed and are required to commit to long-term, on-the-job training before becoming fully qualified. Our licensed water professionals have a basic understanding of a wide range of subjects, including mathematics, biology, chemistry, and physics. Some of the tasks they complete on a regular basis include:

- Operating and maintaining equipment to purify and clarify water.
- Monitoring and inspecting machinery, meters, gauges, and operating conditions.
- Conducting tests and inspections of water and evaluating the results.
- Maintaining optimal water chemistry.
- Applying data to formulas that determine treatment requirements, flow levels, and concentration levels.
- Documenting and reporting test results and system operations to regulatory agencies.
- Serving our community through customer support, education, and outreach.

So the next time you turn on your faucet, think of the skilled professionals who stand behind each drop.

To the Last Drop

The National Oceanic and Atmospheric Administration (NOAA) defines drought as a deficiency in precipitation over an extended period of time, usually a season or more, resulting in a water shortage causing adverse impacts on vegetation, animals, and people. Drought strikes in virtually all climate zones, from very wet to very dry.

There are primarily three types of drought: Meteorological Drought refers to the lack of precipitation, or the degree of dryness and the duration of the dry period; Agricultural Drought refers to the agricultural impact of drought, focusing on precipitation shortages, soil water deficits, and reduced groundwater or reservoir levels needed for irrigation; and Hydrological Drought pertains to the periods following extended precipitation shortfalls that can impact water supply (i.e., stream flow, reservoir and lake levels, groundwater).

Drought is a temporary aberration from normal climatic conditions; thus, it can vary significantly from one region to another. Although normally occurring, human factors such as water demand can exacerbate the duration and impact that drought has on a region. By following simple water conservation measures, you can help significantly reduce the lasting effects of extended drought.

How Is My Water Treated and Purified?

The typical water treatment process requires several steps. These steps are required to ensure that your water is safe, wholesome, and free of contaminants.

Intake from source water: This is where the water entering the treatment process is screened to remove large debris.

Coagulation: Small particles are brought together to form a large floc, which allows for the majority of sediments to settle out of the water.

Filtration: The remaining finer particles are filtered from the water using specially designed filter media.

Disinfection: A disinfectant is applied to kill any bacteria that may be present in the water.

Storage: The finished water product is stored in sealed tanks, from which it is delivered to the consumer.

Quality Monitoring: Water quality is monitored at the treatment process and throughout the distribution system to ensure that the water is in compliance with federal and state standards at all times.

Table Talk

Cet the most out of the Testing Results data table with this simple suggestion. In less than a minute, you will know all there is to know about your water:

For each substance listed, compare the value in the Amount Detected column against the value in the MCL (or AL, SMCL) column. If the Amount Detected value is smaller, your water meets the health and safety standards set for the substance.

Other Table Information Worth Noting

Verify that there were no violations of the state and/or federal standards in the Violation column. If there was a violation, you will see a detailed description of the event in this report.

If there is an ND or a less-than symbol (<), that means that the substance was not detected (i.e., below the detectable limits of the testing equipment).

The Range column displays the lowest and highest sample readings. If there is an NA showing, that means only a single sample was taken to test for the substance (assuming there is a reported value in the Amount Detected column).

If there is sufficient evidence to indicate from where the substance originates, it will be listed under Typical Source.

Benefits of Chlorination

Disinfection, a chemical process used to control disease-causing microorganisms by killing or inactivating them, is unquestionably the most important step in drinking water treatment. By far, the most common method of disinfection in North America is chlorination.

Before communities began routinely treating drinking water with chlorine (starting with Chicago and Jersey City in 1908), cholera, typhoid fever, dysentery, and hepatitis A killed thousands of U.S. residents annually. Drinking water chlorination and filtration have helped to virtually eliminate these diseases in the U.S. Significant strides in public health are directly linked to the adoption of drinking water chlorination. In fact, the filtration of drinking water and the use of chlorine are probably the most significant public health advancements in human history.

How chlorination works:

Potent Germicide Reduction of many diseasecausing microorganisms in drinking water to almost immeasurable levels.

Taste and Odor Reduction of many disagreeable tastes and odors from foul-smelling algae secretions, sulfides, and decaying vegetation.

Biological Growth Elimination of slime bacteria, molds, and algae that commonly grow in water supply reservoirs, on the walls of water mains, and in storage tanks.

Chemical Removal of hydrogen sulfide (which has a rotten egg odor), ammonia, and other nitrogenous compounds that have unpleasant tastes and hinder disinfection. It also helps to remove iron and manganese from raw water.

What's Your Water Footprint?

Your water footprint? The water footprint of an individual, community, or business is defined as the total volume of freshwater that is used to produce the goods and services that are consumed by the individual or community or produced by the business. For example, 11 gallons of water is needed to irrigate and wash the fruit used to produce one half-gallon container of orange juice; 37 gallons of water is used to grow, produce, package, and ship the beans in that morning cup of coffee; 264 gallons of water is required to produce one quart of milk; and 4,200 gallons of water is required to produce two pounds of beef.

According to the U.S. EPA, the average American uses over 180 gallons of water daily. In fact, in the developed world, one flush of a toilet uses as much water as the average person in the developing world allocates for an entire day's cooking, washing, cleaning, and drinking. The annual American per capita water footprint is about 8,000 cubic feet, twice the global per capita average. With water use increasing sixfold in the past century, our demands for freshwater are rapidly outstripping what the planet can replenish. To check out your own water footprint, visit www. watercalculator.org.

Source Water Assessment

An assessment of the drinking water sources for all TUD water systems was completed in 2013. A vulnerability summary for each system is included. A copy of the complete assessment of each system may be viewed at the Department of Health Services Water Field Operations Branch, Merced District Office, 265 West Bullard Avenue, Suite 101, Fresno.

VULNERABILITY	APPLE VALLEY	PEACEFUL PINES	PHOENIX LAKE	SONORA	PONDEROSA	TUOLUMNE	UPPER BASIN	COLUMBIA	CEDAR RIDGE	SCENIC VIEW	WARDS FERRY
Sewer Collection	X			X		X	X	X			
Septic System Low Density				X		X		X			X
Septic System High Density		X	X	X	X		X		X	X	
Grazing	X						X				X
Other Animal Operations	X						X				
Lumber Processing/ Manufacturing	X			X							
Wood/Pulp/Mills								X			
Recreational/Surface water source				X	X	X	X	X	X	X	
Historic waste dumps/ landfills				X			X				
Auto/Machine Shop				X							
Car Washing				X							
Dry Cleaners				X							
Highways/Transportation Corridor				X							



Test Results

Our water is monitored for many different kinds of substances on a very strict sampling schedule, and the water we deliver must meet specific health standards. Here, we only show those substances that were detected in our water (a complete list of all our analytical results is available upon request). Remember that detecting a substance does not mean the water is unsafe to drink; our goal is to keep all detects below their respective maximum allowed levels.

The state recommends monitoring for certain substances less than once per year because the concentrations of these substances do not change frequently. In these cases, the most recent sample data are included, along with the year in which the sample was taken.

REGULATED SUBSTANCES	UBSTANCES		·											
					Apple Valley	Valley	Cedar	Cedar Ridge	Columbia/Big Hill	/Big Hill	Peaceful Pines	Pines		
SUBSTANCE (UNIT OF MEASURE)		YEAR SAMPLED	MCL D [MRDL]	PHG (MCLG) [MRDLG]	AMOUNT DETECTED	RANGE LOW-HIGH	AMOUNT DETECTED	RANGE LOW-HIGH	AMOUNT DETECTED	RANGE LOW-HIGH	AMOUNT DETECTED	RANGE LOW-HIGH	VIOLATION	TYPICAL SOURCE
Arsenic (ppb)		2022	10	0.004	ND	NA	N ON	NA	ND	NA	ND	NA	No	Erosion of natural deposits; runoff from orchards; glass and electronics production wastes
Asbestos (MFL)		2019		7	ND	NA	QN	NA	ND^2	NA	ND³	NA	N _o	Internal corrosion of asbestos cement water mains; erosion of natural deposits
Chlorine (ppm)		2022	[4.0 (as Cl2)]	[4 (as Cl2)]	1.09	0.73-1.51	1.58	1.4–1.8	1.65	1.6–1.7	0.93	0.4-1.5	No	Drinking water disinfectant added for treatment
Control of DBP precursors [TOC] (ppm)	precursors	2022	L	NA	NA	NA	1.3	1.1–1.8	NA	NA	NA	NA	No	Various natural and human- made sources
Fluoride (ppm)		2021	2.0	П	N O N	Z	0.09²	ND-0.18 ²	ND^2	NA	0.2	NA V	°Z	Erosion of natural deposits; water additive that promotes strong teeth; discharge from fertilizer and aluminum factories
Gross Alpha Particle Activity (pCi/L)	rticle Activity	2020	15	(0)	2.45	NA	NDŞ	NA	NDŠ	NA	ND®	NA	No	Erosion of natural deposits
HAA5 [sum of 5 haloacetic acids]—Stage 2 (ppb)	5 haloacetic (ppb)	2020	09	NA	2	NA	42.5²	35-47²	49.5²	37–59²	2.7	NA	oN.	By-product of drinking water disinfection
Magnesium (ppm)	·m)	2022	NA	NA	19	NA	2.2	0.81–3.6	0.82	0.81-0.83	101	NA	No	NA
Nitrate [as nitrogen] (ppm)	ogen] (ppm)	2022	10	10	0.367	ND-0.63	ΩN	NA	ND	NA	ND	NA	S	Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits
TTHMs [total trihalomethanes]-Stage 2 (ppb)	.]-Stage 2 (ppl	2020	80	NA	0.5	N A	35.25²	$32-40^{2}$	505	42–58²	4.6	NA	oN o	By-product of drinking water disinfection
Turbidity (NTU)	J)	2021	H	NA	ND	NA	0.08^{2}	0.04-0.082	0.072	0.04-0.072	0.12 0	0.12-0.12	No	Soil runoff
Tap water samples were collected for lead and copper analyses from sample sites throughout the community	were collected for	or lead and c	opper analys	es from sampl	e sites throug	hout the comm	1							
			Ap	Apple Valley		Cedar Ridge	වි	Columbia/ Big Hill	Pe	Peaceful Pines				
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED AL	PHG L (MCLG)	AMOUNT DETECTED (90TH %ILE)	SITES ABOVE AL/ TOTAL SITES	AMOUNT / DETECTED ES (90TH %ILE)	IT SITES ED ABOVE AL/ ILE) TOTAL SITES	AMOUNT L/ DETECTED TES (90TH %ILE)	NT SITES ED ABOVE AL/ ILE) TOTAL SITES	AMOUNT L/ DETECTED ES (90TH %ILE)	IT SITES ED ABOVE AL/ LE) TOTAL SITES	S VIOLATION		TYPICAL SOURCE	
Copper (ppm)	2020 1.3	.3 0.3	0.255	0/5	0.150	06/0	0.120	0 0/20	0.033	3 0/5	ν ^ο	Internal corr erosion of na preservatives	al corrosion o 1 of natural d atives	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives
Lead (ppb)	2020 15	5 0.2	ND	0/5	7.5	3/30	ON	0/20	ND	0/5	No	Interna systema erosion	Internal corrosion of house systems; discharges from in erosion of natural deposits	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits

REGULATED SUBSTANCES	STANCES													
					Phoenix Lake	Lake	Ponderosa	erosa	Scen	Scenic View	So	Sonora		
SUBSTANCE (UNIT OF MEASURE)		YEAR SAMPLED	MCL [MRDL]	PHG (MCLG) [MRDLG]	AMOUNT DETECTED	RANGE LOW-HIGH	AMOUNT DETECTED	RANGE LOW-HIGH	AMOUNT	RANGE LOW-HIGH	AMOUNT	RANGE LOW-HIGH	VIOLATION	TYPICAL SOURCE
Arsenic (ppb)		2022	10	0.004	ND	NA	Q Z	NA	ND	NA	ND	NA	°Z	Erosion of natural deposits; runoff from orchards; glass and electronics production wastes
Asbestos (MFL)		2019	7		ND	NA	ND	NA	ND⁵	Z	ND^2	NA	°Z	Internal corrosion of asbestos cement water mains; erosion of natural deposits
Chlorine (ppm)		2022	[4.0 (as Cl2)]	[4 (as Cl2)]	1.06	0.6–1.53	1.67	1.6–1.7	1.62	1.5–1.7	1.82	1.7–2.0	No	Drinking water disinfectant added for treatment
Control of DBP precursors [TOC] (ppm)	ecursors	2022	LL	NA	NA	NA	1.3	1.1–1.6	1.3	0.9–1.6	1.4	0.8–2.8	N _o	Various natural and human- made sources
Fluoride (ppm)		2021	2.0	-	0.13	NA A	ND^2	NA	ND²	NA	ND^2	NA	ž	Erosion of natural deposits; water additive that promotes strong teeth; discharge from fertilizer and aluminum factories
Gross Alpha Particle Activity (pCi/L)	e Activity	2020	15	(0)	2.891	NA	ND ⁵	NA	3.531	NA	ND	NA	%	Erosion of natural deposits
HAA5 [sum of 5 haloacetic acids]—Stage 2 (ppb)	loacetic	2020	09	NA	12^2	NA	51²	46–57²	282	15–38²	40.06²	28–56²	%	By-product of drinking water disinfection
Magnesium (ppm)		2022	NA	NA	221	NA	0.82	NA	13.05	1.1–25	8.5	1.0–16	No	NA
Nitrate [as nitrogen] (ppm)	(mdd) [1	2022	10	10	ND	NA	ND	NA A	1.0	ND-2.0	ND	NA A	°Z	Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits
TTHMs [total trihalomethanes]-Stage 2 (ppb)	tage 2	2020	80	NA	41²	NA	50.75²	39–59²	36.75²	23–43²	60.56²	45–72²	, N	By-product of drinking water disinfection
Turbidity (NTU)		2021	II	NA	0.1	0.1-0.1	0.072	0.05-0.072	0.062	0.05-0.062	0.06^{2}	$0.04-0.06^{2}$	No	Soil runoff
Tap water samples were collected for lead and copper analyses from sample sites throughout the community Phoenix Lake Ponderosa	collected for l	aad and cop	per analyse.	alyses from sample Phoenix Lake	sites through	hout the comm Ponderosa	ı	Scenic View	L	Sonora	ı			
SUBSTANCE Y (UNIT OF Y MEASURE) SAN	YEAR SAMPLED AL	PHG (MCLG)	AMOUNT DETECTED (90TH %ILE)	SITES ABOVE AL/ TOTAL SITES	AMOUNT / DETECTED SS (90TH %ILE)	SITES D ABOVE AL/ E) TOTAL SITES	AMOUNT L/ DETECTED ES (90TH %ILE)	IT SITES ED ABOVE AL/ LE) TOTAL SITES	AMOUNT / DETECTED ES (90TH %ILE)	NT SITES TED ABOVE AL/ ILE) TOTAL SITES	-/ ES VIOLATION	ON TYPICAL SOURCE	SOURCE	
Copper (ppm) 2	2020 1.3	0.3	0.255	9/2	0.100	0/111	0.092	0/10	0.0882	8 ² 0/31 ²	Š		l corrosion of of natural de atives	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives
Lead (ppb) 2	2020 15	0.2	8.2	5/0	ND	0/111	ND	0/101	ND	² 0/31 ²	Š.		Internal corrosion of house systems; discharges from in erosion of natural deposits	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits

REGULATED SUBSTANCES	BSTANCE	S											
						Tuolu	Tuolumne	Пррег	Upper Basin	Wards Ferry	Ferry		
SUBSTANCE (UNIT OF MEASURE)		Ŋ	YEAR	MCL [MRDL]	PHG (MCLG) [MRDLG] [AMOUNT DETECTED	RANGE LOW-HIGH	AMOUNT DETECTED	RANGE LOW-HIGH	AMOUNT DETECTED	RANGE LOW-HIGH	VIOLATION	TYPICAL SOURCE
Arsenic (ppb)			2022	10	0.004	ND	NA	ND	NA	ND	NA	N _o	Erosion of natural deposits; runoff from orchards; glass and electronics production wastes
Asbestos (MFL)			2019	7	7	ND	NA	ND^2	NA	ND^4	NA	No	Internal corrosion of asbestos cement water mains; erosion of natural deposits
Chlorine (ppm)			2022	[4.0 (as Cl2)]	[4 (as Cl2)]	1.45	1.2–1.7	1.72	1.57–1.83	0.76	0.56–1.07	No	Drinking water disinfectant added for treatment
Control of DBP precursors [TOC] (ppm)	precursors		2022	TT	NA	1.5	1.1–1.8	1.2	0.95–1.65	NA	NA	%	Various natural and human-made sources
Fluoride (ppm)			2021	2.0	П	ND^2	NA	0.058²	ND-0.15 ²	N	NA	No	Erosion of natural deposits; water additive that promotes strong teeth; discharge from fertilizer and aluminum factories
Gross Alpha Particle Activity (pCi/L)	ticle Activit		2020	15	(0)	ND	NA	ND°	NA	1.227	NA	%	Erosion of natural deposits
HAA5 [sum of 5 haloacetic acids]—Stage 2 (ppb)	haloacetic pb)		2020	09	NA	452	40-54²	40.632	29–50²	2	NA	%	By-product of drinking water disinfection
Magnesium (ppm)	a)		2022	NA	NA	0.7	NA	3.17	0.82–7.5	161	NA	No	NA
Nitrate [as nitrogen] (ppm)	gen] (ppm)		2022	10	10	ND	NA	N	NA	3.8	NA	%	Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits
TTHMs [total trihalomethanes]-Stage 2 (ppb)	-Stage 2		2020	80	NA	46.5²	43–49²	31.63²	26–37²	8.9	NA	No	By-product of drinking water disinfection
Turbidity (NTU)			2021	TT	NA	0.08^{2}	$0.03-0.08^{2}$	0.07^{2}	$0.03-0.07^{2}$	ND	NA	No	Soil runoff
Tap water samples were collected for lead and copper analyses from sample sites throughout the community	rere collected	for lead	and copp	er analyses	from sample	sites through	out the commu	nity					
				Tuol	Tuolumne	5	Upper Basin	8	Wards Ferry				
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED	AL (§ _	PHG [AMOUNT DETECTED (90TH %ILE)	SITES ABOVE AL/ TOTAL SITES	AMOUNT DETECTED S (90TH %ILE)	T SITES D ABOVE AL/ E) TOTAL SITES	AMOUNT / DETECTED SS (90TH %ILE)	T SITES ED ABOVE AL/ E) TOTAL SITES	S VIOLATION	N TYPICAL SOURCE	OURCE	
Copper (ppm)	2020	1.3	0.3	0.170^{1}	0/101	0.200	0/22	0.625	1 0/51	Š	Internal deposits	corrosion of teaching fro	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives
Lead (ppb)	2020	15	0.2	ND	0/101	ND	2/22	ND	0/51	N _o	Internal industria	corrosion of al manufactu	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits

SECONDARY SUBSTANCES	3STANCES															
				Apple Valley	/alley	peg	Cedar Ridge	Colun	Columbia/Big Hill	Peg	Peaceful Pines	Ph	Phoenix Lake			
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED	SMCL	PHG (MCLG)	AMOUNT DETECTED	RANGE LOW- HIGH	AMOUNT DETECTED	RANGE LOW-HIGH	AMOUNT DETECTED	F RANGE D LOW-HIGH	E AMOUNT H DETECTED	RANGE INT LOW- TED HIGH	E AMOUNT DETECTED	RANGE NT LOW- TED HIGH	SE VIOLATION		TYPICAL SOURCE
Chloride (ppm)	2021	200	NS	8.1	NA	3.65²	3.6–3.7²	3.8²	3.1–4.5²	5² ND	NA	9.5	NA NA	No		Runoff/leaching from natural deposits; seawater influence
Iron (ppb)	2021	300	NS	ND	NA A	39²	ND-78 ²	ND	NA	ND ²)² NA	ND ²	NA NA	No		Leaching from natural deposits; industrial wastes
Manganese (ppb)	2021	90	NS	111	NA	92	$ND-18^2$	ND^2	NA	ND²	D ² NA	ND ²	2 NA	No		Leaching from natural deposits
Sulfate (ppm)	2021	200	NS	13	NA	2.65²	ND-5.3 ²	ND ₂	NA	3.2	NA	3.3	NA	No		Runoff/leaching from natural deposits; industrial wastes
Zinc (ppm)	2021	5.0	NS	ND	NA	ND^2	NA	ND^2	NA	ON	NA	QN	NA	No		Runoff/leaching from natural deposits; industrial wastes
				Ponderosa	rosa	Sce	Scenic View		Sonora		Tuolumne	в				
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED	SMCL	PHG (MCLG)	AMOUNT DETECTED	RANGE LOW- HIGH	AMOUNT	RANGE LOW-HIGH	AMOUNT		RANGE /	AMOUNT DETECTED	RANGE LOW- HIGH V	VIOLATION	TYPICAL SOURCE	JRCE	
Chloride (ppm)	2021	500	NS	3.62	NA	15.32	5.6–25²	147.32	·	4.6–290²	4.42	NA	No	Runoff/le	ching from 1	Runoff/leaching from natural deposits; seawater influence
Iron (ppb)	2021	300	NS	ND^2	NA	ND^2	NA	ND^2		NA	ND^2	NA	Š	Leaching	rom natural	Leaching from natural deposits; industrial wastes
Manganese (ppb)	2021	90	NS	ND^2	NA	11.52	ND-23 ²	² 14 ²		13–15²	ND^2	NA	Š	Leaching	Leaching from natural deposits	deposits
Sulfate (ppm)	2021	500	NS	ND_2	NA	11.1752	$0.37-22^{2}$	4.05		$3.7-4.4^{2}$	ND^2	NA	No	Runoff/le	iching from 1	Runoff/leaching from natural deposits; industrial wastes
Zinc (ppm)	2021	5.0	NS	ND^2	NA	ND_2	NA	0.175^{2}		ND-0.35 ²	ND^2	NA	No	Runoff/le	ching from 1	Runoff/leaching from natural deposits; industrial wastes
				j	Upper Basin		Wards Ferry	_								1 Sampled in 2021.
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED	SMCL	PHG (MCLG)	AMOUNT DETECTED	RANGE LOW-HIGH		AMOUNT DETECTED	RANGE LOW- HIGH VIC	VIOLATION	TYPICAL SOURCE	RCE					- Sampled in 2022. 3 Sampled in 2016. 4 Sampled in 2018.
Chloride (ppm)	2021	200	NS	2.88²	1.6-4.4²			NA	%	Runoff/lea	ching fron	natural de	posits; sea	Runoff/leaching from natural deposits; seawater influence	nce	5 Sampled in 2014.
Iron (ppb)	2021	300	SN	34^{2}	ND	ND-170 ²	ND	NA	°N	Leaching f	Leaching from natural deposits; industrial wastes	1 deposits;	industrial	wastes		Sampled in 2019.
Manganese (ppb)	2021	50	NS	19.8^{2}	Ŋ	ND-80 ²	ND	NA	oN	Leaching f	Leaching from natural deposits	1 deposits				8 Unregulated contaminant monitoring
Sulfate (ppm)	2021	900	SN	0.58^{2}	-dN	ND-2.9 ²	4.1	NA	oN.	Runoff/lea	ching fron	natural de	posits; inc	Runoff/leaching from natural deposits; industrial wastes	ses	Resources Control Board determine
Zinc (ppm)	2021	5.0	NS	0.036^{2}	ND-(ND-0.180 ²	ND	NA	No	Runoff/Iea	ching fron	natural de	posits; inc	Runoff/leaching from natural deposits; industrial wastes	es	where certain contaminants occur
UNREGULATED SUBSTANCES 8	SUBSTANC	CES 8														to be regulated.
				Apple Valley	ley	Ced	Cedar Ridge	Colum	Columbia/Big Hill	_	Peaceful Pines		Phoenix Lake	ake		
SUBSTANCE (UNIT OF MEASURE)		YEAR		AMOUNT DETECTED	RANGE LOW-HIGH	AMOUNT	RANGE LOW-HIGH	AMOUNT	RANGE D LOW-HIGH	E AMOUNT		RANGE AN LOW-HIGH DE	AMOUNT	RANGE T	TYPICAL SOURCE	
Hardness, Total [as CaCO3] (ppm)	, CaCO3]	2021	21	200	NA	65 ²	10–120²	10.5^{2}	$10-111^{2}$	1^2 79		NA	270	NA	NA	
Sodium (ppm)		2021	21	14	NA	7.15²	$4.4-9.9^{2}$	6.9^{2}	6.8–7²	72 18		NA	21	NA 1	NA	
			Ponderosa		Scenic View	We	Sonora		Tuolumne	nne	Прре	Upper Basin	Wa	Wards Ferry		
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED	AMOUNT		RANGE AMI	AMOUNT DETECTED L	RANGE LOW-HIGH	AMOUNT DETECTED 1	RANGE LOW-HIGH	AMOUNT	RANGE LOW-HIGH	AMOUNT DETECTED	RANGE LOW-HIGH	AMOUNT	RANGE D LOW-HIGH	TYPICAL	
Hardness, Total [as CaCO3] (ppm)	2021	11^{2}		NA 13	131.5 ² 1	13–250²	101 ² 1	12–190²	10^2	NA	34.78^{2}	9.9–78²	140	ZA	NA	
Sodium (ppm)	2021	5.8 ²		NA 1	13.1 ² 7	7.2–19²	13^2	11–15²	8.2^{2}	NA	6.82^{2}	3.4–9.7²	111	NA	NA	

BY THE NUMBERS

The number of Olympic-sized swimming pools it would take to fill up all of Earth's water.

800 TRILLION

The average cost in cents for about 5 gallons of water supplied to a home in the U.S.

The percent of Earth's water that is salty or otherwise undrinkable, or locked away and unavailable in ice caps and glaciers.

The average daily number of gallons of total home water use for each person in the U.S.

The percent of Earth's surface that is covered by water.

330 willion

The amount of water on Earth in cubic miles.

The percent of the human brain that contains water.

75

Definitions

90th %ile: The levels reported for lead and copper represent the 90th percentile of the total number of sites tested. The 90th percentile is equal to or greater than 90% of our lead and copper detections.

AL (Regulatory Action Level): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

MCL (Maximum Contaminant Level): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs (SMCLs) are set to protect the odot, taste, and appearance of drinking water.

MCLG (Maximum Contaminant Level Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. EPA.

MFL (million fibers per liter): A measure of the presence of asbestos fibers that are longer than 10 micrometers.

MRDL (Maximum Residual Disinfectant Level): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

MRDLG (Maximum Residual Disinfectant Level Goal): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

NA: Not applicable.

ND (**Not detected**): Indicates that the substance was not found by laboratory analysis.

NS: No standard.

NTU (Nephelometric Turbidity Units): Measurement of the clarity, or turbidity, of water. Turbidity in excess of 5 NTU is just noticeable to the average person.

pCi/L (picocuries per liter): A measure of radioactivity.

PDWS (Primary Drinking Water Standard): MCLs and MRDLs for contaminants that affect health, along with their monitoring and reporting requirements and water treatment requirements.

PHG (Public Health Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California EPA.

ppb (parts per billion): One part substance per billion parts water (or micrograms per liter).

ppm (parts per million): One part substance per million parts water (or milligrams per liter).

TT (Treatment Technique): A required process intended to reduce the level of a contaminant in drinking water.

