

# **La Cumbre Mutual Water Company**

# 695 Via Tranquila Santa Barbara 967-2376 2018 CONSUMER CONFIDENCE REPORT DATA

Este informe contiene información muy importante sobre su agua beber. Tradúzcalo ó hable con alguien que lo entienda bien.

Please see last page for key to abbreviations.

Parameter	Units	State MCL	PHG (MCLG)	Range Average	GROUND WATER	SURFACE WATER	Major Sources in Drinking Water
PRIMARY STANDA			_ ' _ '	1	WAILK	WAILK	Major Sources in Brinking Water
CLARITY (a)							-
Combined Filter Effluent Turbidity	NTU	TT=1 NTU		ghest Single leasurement	NA	0.10	Soil runoff
Lindent Tarbiaity		TT=95% of samp		leasurement	NA NA	100% Sample	gs <= 0.3 NTU
MICROBIOLOGICAL (b)							
Fotal Coliform Bacteria	Samples	5% of monthly	(0)	Reporting			Naturally present in the environment
(Distribution System)	Campies	samples (b)	(0)	Value	0 Positives	0.0%	Indically present in the environment
Fecal Coliform and				Range	0 Positives	0 Positives	
E. coli (Distribution System)	Samples	(b)	(0)	Average Highest	0 Positives 0 Positives	0 Positives 0 Positives	Human and animal fecal waste
Disinfectant Byproduct	s - Disinfe	ectant Residuals	Disinfection Byr	•	-	01 03111763	
Fotal Trihalomethanes	<u>J Disinite</u>	otant Residuais	Districction By	Range	26.9 - 48.6	0.56 - 62	By-product of drinking water
Distribution System)( c) Haloacetic Acids (c)	ppb	80	NA	Average Range	38 5.5 - 13.3	53 ND - 26	disinfection By-product of drinking water
Distribution System)	ppb	60	NA	Average	9.4	24	disinfection
Disinfectant - Free		MRDL as Cl2	MRDLG as Cl2	Range	0.20 - 5.10 (d)	ND - 1.72	Measurement of the disinfectant
Chlorine Residual Control of DBP pre-	ppm	4.0	4.0	Average Range	0.95 (d) 0.62 - 1.2 (d)	0.69 ND - 3.21	used in the production of drinking water TOC has no health effects. However, it provide
cursors - TOC	ppm	TT	NA	Average	0.89 (d)	2.72	a medium for the formation of disinfection by-
NODOANIO OUEMOA							products. Various natural and manmade source
NORGANIC CHEMICA	LS			Range	ND	ND - 60	Residue from water treatment process:
Aluminum	ppb	1000	600	Average	NA	30	Erosion of natural deposits
Arsenic	nnh	10	0.004	Range Average	ND NA	ND NA	Erosion of natural deposits
Arsenic	ppb	IU	0.004	Range	NA ND	NA ND	Erosion of natural deposits
Barium	ppm	1	2	Average	NA	NA	·
Fluoride	ppm	2	1	Range Average	0.32 - 0.63 0.50	ND - 0.45 0.33	Erosion of natural deposits; water additive for tooth health
	ррпп		'	Range	ND - 1.9	ND	Runoff & leaching from fertilizer
Nitrate (as NO <sub>3</sub> )	ppm	45	45	Average	0.6	NA NA	use; sewage; natural erosion
Hexavalent chromium, Cr VI	ppb	10	0.02	Range Average	ND NA	NA 0.022	Discharge from electroplating factories, leather tanneries, wood preservation, chemical synthes
							refractory production, and textile manufacturing
							facilities; erosion of natural deposits
EAD & COPPER RULE	S - Moni	tored at the custo	omers tap. Numb	er of sites e	xceeded Action	Level = 0	
				Value	2.50		Internal corrosion of household plumbing
Copper	ppm	1.3	0.3	90th %	0.50	NA	systems; erosion of natural deposits; leaching from wood preservatives.
				Value			Internal corrosion of household plumbing
₋ead	ppb	15	0.2	90th %	ND	NA	systems; discharges from industrial
							manufactures; erosion of natural deposits.
RADIOCHEMISTRY - Ra	dioactive	Contaminants		_			
Gross Alpha				Range	ND - 5.1	ND	Erosion of natural deposits
2 " 000	pCi/L	15	MCLG, 0	Average	3.0	NA	
Radium 228	pCi/L	N/A	N/A	Range Average	0 - 0.283 0.0566	ND NA	Erosion of natural deposits
				rivorago	0.0000	10.0	
SECONDARY STA	NDARD	SAesthetic S	Standards	_			
Chloride	nnm	500	NA	Range Average	46 - 130 88.3	33 - 140 60	Runoff/leaching from natural deposits; seawater influence
Sillonde	ppm	500	INA	Range	ND	ND	Seawater irinderice
Color (ACU)	Units	15	NA	Average	NA	NA	Naturally occurring organic materials
Conner	nnh	1000	NIA	Range Average	ND NA	ND NA	Corrosion of plumbing systems; erosion of natur
Copper	ppb	1000	NA	Range	ND - 540	ND - 21	deposits; leaching from wood preservatives Leaching from natural deposits;
ron	ppb	300	NA	Average	87.8	14	industrial wastes
Manganese	nnh	50	NA	Range Average	ND NA	ND - 1.2 0.53	Leaching from natural deposits
viarigariese	ppb	30	INA	Range	1 - 4	2 - 3	Naturally occurring organic materials
Odor Threshold	Units	3	NA	Average	2.3	3	
Specific Conductance	µmho/ cm	1600	NA	Range Average	1100 - 1400 1300	550 - 1050 867	Substances that form ions when in water; seawater influence.
	JIII	1000	14/1	Range	250 - 320	2.3 - 335	Runoff/leaching from natural deposits;
Sulfate	ppm	500	NA	Average	268.7	210	industrial wastes
Fotal Dissolved Solids	ppm	1000	NA	Range Average	650 - 950 836.7	270 - 804 595	Runoff/leaching from natural deposits; seawater influence
				Range	0.7 - 2.34	0.05 - 0.16	Soil runoff
Furbidity (Monthly)	NTU	5	NA	Average Range	1.7 ND - 0.33	0.09 ND	Naturally occurring in trace amounts, but can
		5.0	NA	Average	0.11	NA NA	be detected in soft, acidic water systems
Zinc	ppm						
		roguloted					
Additional Parame		regulated):		Pango	2/10 . 220	51.205	Runoff/leaching from natural deposits:
Additional Parame		regulated):	NA	Range Average	240 - 330 300	51 - 205 170	Runoff/leaching from natural deposits; seawater influence
Additional Parame Alkalinity (Total) as CaCO <sub>3</sub> equivalents	ters (Un	NA		Average Range	300 110 - 140	170 21 - 106	seawater influence Runoff/leaching from natural deposits;
Additional Parame Alkalinity (Total) as CaCO <sub>3</sub> equivalents Calcium as Ca	ters (Un		NA NA	Average Range Average	300 110 - 140 126.7	170 21 - 106 79	seawater influence Runoff/leaching from natural deposits; seawater influence
Additional Parame Alkalinity (Total) as CaCO <sub>3</sub> equivalents Calcium as Ca Hardness (Total) as	ppm	NA		Average Range Average Range	300 110 - 140 126.7 420 - 550	170 21 - 106	seawater influence Runoff/leaching from natural deposits;
Additional Parame Alkalinity (Total) as CaCO <sub>3</sub> equivalents  Calcium as Ca Hardness (Total) as CaCO <sub>3</sub>	ppm ppm	NA NA NA	NA NA	Average Range Average Range Average Average Range	300 110 - 140 126.7 420 - 550 493.3 37 - 47	170 21 - 106 79 58 - 476 330 1.4 - 45	seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits;
Additional Parame Alkalinity (Total) as CaCO <sub>3</sub> equivalents Calcium as Ca Hardness (Total) as CaCO <sub>3</sub>	ppm ppm ppm	NA NA	NA	Average Range Average Range Average Average Range Average Average	300 110 - 140 126.7 420 - 550 493.3 37 - 47 42.3	170 21 - 106 79 58 - 476 330 1.4 - 45	seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence
Additional Parame Alkalinity (Total) as CaCO <sub>3</sub> equivalents  Calcium as Ca Hardness (Total) as CaCO <sub>3</sub>	ppm ppm	NA NA NA	NA NA	Average Range Average Range Average Average Range	300 110 - 140 126.7 420 - 550 493.3 37 - 47	170 21 - 106 79 58 - 476 330 1.4 - 45	seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits;
Additional Parame Alkalinity (Total) as CaCO <sub>3</sub> equivalents  Calcium as Ca Hardness (Total) as CaCO <sub>3</sub> Magnesium	ppm ppm ppm ppm ppm pH Units	NA NA NA NA	NA NA NA	Average Range Average Range Average Range Average Range Average Range Average Range	300 110 - 140 126.7 420 - 550 493.3 37 - 47 42.3 7.3 - 7.6 7.5 2.3 - 2.9	170 21 - 106 79 58 - 476 330 1.4 - 45 30 7.50 - 9.03 7.76 3.8 - 5.1	seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits;
Additional Parame Alkalinity (Total) as CaCO <sub>3</sub> equivalents Calcium as Ca Hardness (Total) as CaCO <sub>3</sub> Magnesium OH	ppm ppm ppm ppm ppm	NA NA NA	NA NA NA	Average Range Average Range Average Range Average Range Average Range Average Average Average Average	300 110 - 140 126.7 420 - 550 493.3 37 - 47 42.3 7.3 - 7.6 7.5 2.3 - 2.9 2.6	170 21 - 106 79 58 - 476 330 1.4 - 45 30 7.50 -9.03 7.76 3.8 - 5.1	seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits  Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence
Additional Parame Alkalinity (Total) as CaCO <sub>3</sub> equivalents  Calcium as Ca Hardness (Total) as CaCO <sub>3</sub> Magnesium	ppm ppm ppm ppm ppm pH Units	NA NA NA NA	NA NA NA	Average Range Average Range Average Range Average Range Average Range Average Range	300 110 - 140 126.7 420 - 550 493.3 37 - 47 42.3 7.3 - 7.6 7.5 2.3 - 2.9	170 21 - 106 79 58 - 476 330 1.4 - 45 30 7.50 - 9.03 7.76 3.8 - 5.1	seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits;
Additional Parame Alkalinity (Total) as CaCO <sub>3</sub> equivalents Calcium as Ca Hardness (Total) as CaCO <sub>3</sub> Magnesium OH	ppm ppm ppm ppm ppm pH Units	NA NA NA NA NA NA	NA NA NA NA	Average Range Average Range Average Range Average Range Average Range Average Average Range Average Range	300 110 - 140 126.7 420 - 550 493.3 37 - 47 42.3 7.3 - 7.6 7.5 2.3 - 2.9 2.6 85 - 120	170 21 - 106 79 58 - 476 330 1.4 - 45 30 7.50 -9.03 7.76 3.8 - 5.1 4.4 51 - 80	seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits;
Additional Parame  Alkalinity (Total) as  CaCO <sub>3</sub> equivalents  Calcium as Ca  Hardness (Total) as  CaCO <sub>3</sub> Magnesium  OH	ppm ppm ppm ppm pH Units ppm	NA NA NA NA NA NA	NA NA NA NA	Average Range Average Range Average Range Average Range Average Range Average Average Range Average Range	300 110 - 140 126.7 420 - 550 493.3 37 - 47 42.3 7.3 - 7.6 7.5 2.3 - 2.9 2.6 85 - 120	170 21 - 106 79 58 - 476 330 1.4 - 45 30 7.50 -9.03 7.76 3.8 - 5.1 4.4 51 - 80	seawater influence Runoff/leaching from natural deposits; seawater influence Leaching from natural deposits Runoff/leaching from natural deposits; seawater influence Runoff/leaching from natural deposits;

Range

Average

1000 (AL)

ppb

140 - 270

206.7

NA

850

Runoff/leaching from rocks and soil, wastewater,

and fertilizers/pesticides.

Water System Name: La Cumbre Mutual Water Company

Report Date: May 2019

In 2018, 59% of our water was from our wells, the remaining 41% was State Project Water after flowing into Lake Cachuma and being treated by the Santa Barbara City Cater Surface Water Treatment Plant. Therefore, the surface water quality portion of this report comes from the city of Santa Barbara. Sections of our service area along State Street and Modoc Road receive water that was treated entirely by the city of Santa Barbara.

Time and place of regularly scheduled board meetings for public participation: Once a month at 695 Via Tranquila, please call for exact date and times 967-2376.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include:

- Contaminants that may be present in source water include:

   Microbial contaminants, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.

   Inorganic contaminants, such as salts and metals, that can be naturally-occurring or result from urban storm water runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.

   Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban storm water runoff, and residential uses.

   Organic chemical contaminants, including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban storm water runoff, and septic systems.

   Padigostive contaminants, which can be paturally occurring or be the result of oil and gas production and mining activities.
- · Radioactive contaminants, which can be naturally-occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, USEPA and the State Water Resources Control Board (SWRCB) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. SWRCB regulations also establish limits for contaminants in bottled water that must provide the same protection for public health.

This report lists all of the drinking water contaminants that were detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The SWRCB requires us to monitor for certain contaminants less than once per year because the concentrations of these contaminants are not expected to vary significantly from year to year. Some of the data, though representative of the water quality, are more than one year old.

Additional General Information On Drinking Water:
All drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA's Safe Drinking Water Hotline at 1-800-426-4791 or www.epa.gov/safewater/.

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice from their health care providers about drinking water. USEPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline 1-800-426-4791 or www.epa.gov/safewater/.

### ABBREVIATIONS AND NOTES

Type of water sources in use: Five ground water wells and State Project surface water from Lake Cachuma through Santa Barbara City Cater Treatment Plant.

Name of Sources: Well #16, Well #17, Well #18, Well #19 & Well #21 and seven metered connections to Santa Barbara City Water. Note: Depending on where you live, our water is a mixture of groundwater and surface water.

Water Quality Report: Listed are substances detected in the drinking water. Not listed are more than 135 regulated and unregulated substances that were below the laboratory detection level.

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. La Cumbre Water Co. is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.

### Definitions:

Public Health Goal (PHG): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the Environmental Protection Agency

Maximum Contaminate Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency

Maximum Contaminate Level (MCLs): The highest level of a contaminant

that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a disinfectant (chlorine) added for water treatment below which there is no known or expected risk to health. MRDLGs are set by the U. S. **Environmental Protection Agency** 

Maximum Residual Disinfectant Level (MRDL): The level of a disinfectant (chlorine) added for water treatment that may not be exceeded at the consumer's tap.

**DBP**: Disinfection Byproducts

- (a) Turbidity (NTU) is a measure of the cloudiness of the water and it is a good indicator of the effectiveness of our filtration system. Monthly turbidity values for ground water are listed in the Secondary
- Standards section.

  (b) Total coliform MCLs: The State MCL for coliforms is no more than 1 per month for water systems which collect less than 40 samples per month (La Cumbre Water). Systems which collect over 40 routine samples may not have more than 5% positive per month
- (c) Compliance based on the quarterly annual average distribution
- system samples.

  Although reported under ground water these readings were taken from the distribution system and are a combination of ground and surface water.

Regulatory Action Level (AL): The concentration of a contaminant which if exceeded, triggers a treatment or other requirement which a water system must follow.

**Treatment Technique (TT)**: A required process intended to reduce the level of a contaminant in drinking water.

Primary Drinking Water Standards (PDWS): MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements and water treatment requirements.

Secondary Drinking Water Standards (SDWS): MCL's for contaminants that effect taste, odor or appearance of drinking water. Contaminants with SWDS do not affect the health at MCL levels.

Unregulated Contaminant Monitoring Regulations (UCMR): Data generated by the new UCMR will be used to evaluate and prioritize contaminants on the Drinking Water Contaminant Candidate List, a list of contaminants EPA is considering for possible new drinking water standards. Also known as "State Regulated Contaminants with No MCLs".

NA: Not Applicable ND: Not Detected

## Abbreviations

<" = Less Than

AL = Regulatory Action Level

ACU = Apparent Color Units

MCL = Maximum Contaminant Level
MCLG = Maximum Contaminant Level Goal

MRDL = Maximum Residual Disinfectant Level
MRDLG = Maximum Residual Disinfectant Goal

NA = not applicable
NC = Not Collected
ND = None Detected
NTU = Nephelometric Turbidity Units

pCi/L = PicoCuries per liter PHG = Public Health Goal

ppb = parts per billion, or micrograms per liter (µg/L)

ppm = parts per million, or milligrams per liter (mg/L)
TOC = Total Organic Carbon
TT = Treatment Technique

umho/cm = micromhos per centimeter

(unit of specific conductance of water)