Where Does Your Water Come From?

Sources of the City of Tracy's water supply include the Stanislaus River, the Delta-Mendota Canal, and groundwater pumped from wells. In 2019, 49% of the water supply, or 2.8 billion gallons, came from the Stanislaus River. Water from the Delta-Mendota Canal comprised 47% of the total water supply, or 2.6 billion gallons. The groundwater supply comprised 4%, or 0.21 billion gallons.

During 2020, the City anticipates having an adequate water supply for the community.



Water Quality Control

Before the water reaches your property, samples are collected and tested in State-certified laboratories. The City of Tracy has a water quality monitoring program and inspection system that ensures safe drinking water is delivered to you and your family.

As required by the Federal Safe Drinking Water Act, the City's water supplies must meet stringent water quality standards set by the California Department of Public Health and the United States Environmental Protection Agency. The City of Tracy completed a watershed sanitary survey of its drinking water sources in 2016. This survey can be obtained by contacting the Water Treatment Superintendent at the number provided below.

Water customers who are landlords receiving this report are asked to share this information with any tenant or user on the premises. The City of Tracy staff is available to answer your questions and provide further information: (209) 83 I-6302.







INSTALL DRIP-IRRIGATION
saves
15 GALLONS
each time you water

ADJUST SPRINKLER TO WATER
PLANTS, NOT DRIVEWAY
saves
12-15 GALLONS
each time you water

Learn more ways to save water inside and outside of your home at www.saveourH2O.ora!





2019 Consumer Confidence Report



Think Inside the Triangle

CITY OF TRACY

The City of Tracy is pleased to report that from January I - December 31, 2019 the water delivered to your home or business complied with, or exceeded, all state and federal drinking water requirements! Within this brochure is a table that lists detectable and non-detectable substances found in the City's drinking water, and the maximum allowable substance levels set by United States Environmental Protection Agency (USEPA).



In California, drinking water standards, also called Maximum Contaminant Levels (MCLs), are set in two categories: Primary Standards related to public health, and Secondary Standards which relate to the aesthetic qualities such as taste, odor, and color. Within you will find a complete listing of both types of standards along with the results of the analysis of your water supply.

Este informe contiene información muy importante sobre su agua potable. Tradúzcalo ó hable con alguien que lo entienda bien.

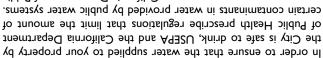
laws that regulate drinking water. and to verify that states are enforcing the consumers are receiving drinkable water uses this data to ensure that the detected in the drinking water. USEPA Department of Public Health if they were reports directly to the California monitors for these substances and substances. Each system continually systems must use to remove these specifies various treatments that water substances in drinking water and also prindreds national TOT simil (SDWA), USEPA is responsible for setting Under the Safe Drinking Water Act Safe Drinking Water Act



Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised persons such as those with cancer undergoing chemotherapy, persons who have undergone organ transsome elderly and infants, can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. USEPA/CDC (Center for Disease Control) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline (800) 426-4791.

Special Health Information

.1974-624 (008) Safe Drinking Water Hotline obtained by calling the USEPA's potential health effects can be mation about contaminants and poses a health risk. More inforsarily indicate that the water contaminants does not necescontaminants. The presence of least small amounts of some bly be expected to contain at ing bottled water, may reasonahealth. Drinking water, includthe same protection for public tled water that must provide limits for contaminants in bot-Health regulations also establish California Department of Public



tivities.

gas stations, urban runoff and septic systems; Radio Active Contaminants, which can be naturally occurring or be the result of oil and gas production and mining ac-

residential uses;

Organic Chemical Contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and can, also come from

gas production, mining, or farming;

Pesticides and Herbicides, which may come from a variety of sources such as agriculture, urban storm water runoff, and

tems, agricultural livestock operations, and wildlife;

Inorganic Contaminants, such as salts and metals, which
can be naturally occurring or result from urban storm water
runoff, industrial or domestic wastewater discharges, oil and

• Microbial Contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic sys-

The sources of drinking water (both tap and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or human activity. Contaminants that may be present in source water include:

Substances Expected to be in the Drinking Water



tion goals! tinued efforts will assist the City in attaining its water conserva-3 o r o n l i n e a t www.thinkinsidethetriangle.com. You're conmay report any water waste by calling(209)831-633 http://www.water.ca.gov/ waterconditions/drought/. Also, you For more information on drought conditions vis it windy day; and do not water for longer than 8 minutes per cycle. tion timers in the winter months; never water landscaping on a features. Some simple voluntary measures are: turning off irrigaturning off non-recirculating fountains and ornamental water psugpeld sprayer and bucket when washing your own car; and outdoor water conservation measures include: using a triggered homeowners and businesses is out-door activities. Mandatory toilets to more efficient models. The biggest use of water by also consider changing out high water consuming appliances and always turning off water while brushing teeth. Businesses might trash receptacle, repairing drips and leaking faucets quickly, and pooing, washing full loads of laundry, never using the toilet as a include: taking shorter showers, turning water off while sham-What are you able to do to help? Some simple indoor measures

Water You Doing to Conserve?

Mew laws require that all water suppliers, like the City of Tracy, must reduce their daily water consumption for all users. This includes indoor and outdoor water uses as well as commercial, industrial and institutional water uses. In order to meet the ever-tincreasing mandates by the State, the City of Tracy enacts Water Stages for conservation measures. Currently, the City is in Stage I of its Water Conservation Drdinance. This limits outdoor water use, such as irrigating your landscape, to the hours of N water use, such as irrigating your landscape, to the hours of N hose to san, each day. It also prohibits using water from your

MAKE CONSERVATION A WAY OF LIFE.

does this mean for you?

pack means the rivers, lakes and streams will not receive the water they need to remain at or above ideal conditions. What

On average, the Sierra snowpack supplies about 50 percent of California's needs as it melts in the spring and early summer to meet higher water demands in the summer and fall. The City of Tracy depends on this snowmelt to fill the reservoir that supplies its water. Unfortunately, a very dry winter with a less than average snowwinter with a less than average snowwinter with a less than average snow-



CONSERVATION: Everyone's Responsibility!

What's in My Water?

	TREATED SUF	TREATED SURFACE WATER WELL WATER				REGULATORY LIMIT		TYPICAL SOURCES
-								
ANALYTICAL PARAMETER	SOUTH SAN JOAQUIN IRRIGATION DISTRICT	JOHN JONES WATER TREATMENT PLANT	AVERAGE	MINIMUM	MAXIMUM	MCLG OR PHG	MAXIMUM CONTAMINANT LEVEL (MCL)	
PRIMARY STANDARDS								
INORGANIC (ug/L)								
Arsenic	ND	ND	1	ND	3	0	10 ug/L	Erosion of natural deposits
Barium	ND	17	28	1	47	2000	2000 ug/L	Erosion of natural deposits
Chromium	ND	ND	6	ND	8	100	100 ug/L	Erosion of natural deposits
Copper	ND	ND	2	ND	10	1300	1300 ug/L	Erosion of natural deposits
Iron	ND	ND	0.1	ND	0.2	NA	300 ug/L	Erosion of natural deposits
Manganese	ND	11	5	ND	29	NA	50 ug/L	Erosion of natural deposits
FLUORIDE (mg/L)								
Fluoride	ND	ND	0.1	ND	0.2	1	2 mg/L	Erosion of natural deposits
NITRATE/NITRITE								
Nitrate (as NO3)	ND	1	8	ND	16	45	45 mg/L	Runoff from fertilizer use; Erosion of
Nitrate + Nitrite (sum as N)		ND	2	ND	4	10	10 mg/L	natural deposits
Nitrite (as N)	ND	ND	ND	ND	ND	1	1 mg/L	
REGULATED ORGANICS (ug/L)								
TRIHALOMETHANE		ND	ND	ND	ND	NA	/1	
Bromodichloromethane Bromoform		ND ND	ND ND	ND ND	ND 1	NA NA	ug/L ug/L	
Chloroform		ND	1	ND	10	NA NA	ug/L ug/L	By-products of drinking water chlorina- tion
Dibromochloromethane		ND	ND	ND	1	NA NA	ug/L	
Total Trihalomethane		ND	2	ND	10	NA NA	80 ug/L	
SECONDARY STANDARDS							0,	
Aesthetic - Related								
Apparent Color (Units)	ND	ND	ND	ND	ND	NA	15 units	Naturally occurring organic materials
Foaming Agents (MBAS) (mg/L)	ND	ND	ND	ND	ND	NA	0.5 mg/L	Municipal and industrial waste discharge
				.,,,	ND		0.56/ -	Municipal and industrial waste discharge
Odor (TON)	1	2	1	ND	2	NA	3 TON	Naturally occurring organic materials
Potassium (K) (mg/L)	1 ND	2 1.40	4	ND 2	2 5	NA NA	3 TON NS	Naturally occurring organic materials Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2	1 ND 0.2	2 1.40 ND	4 1	ND 2 ND	2 5 2	NA	3 TON	Naturally occurring organic materials
Potassium (K) (mg/L)	1 ND	2 1.40	4	ND 2	2 5	NA NA	3 TON NS	Naturally occurring organic materials Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2	1 ND 0.2	2 1.40 ND	4 1	ND 2 ND	2 5 2	NA NA NA	3 TON NS	Naturally occurring organic materials Erosion of natural deposits Soil runoff
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L)	1 ND 0.2 52	2 1.40 ND 17	4 1 170	ND 2 ND 110	2 5 2 230	NA NA NA	3 TON NS	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L)	1 ND 0.2 52 ND	2 1.40 ND 17 ND	4 1 170 ND	ND 2 ND 110 ND	2 5 2 230 ND	NA NA NA NA	3 TON NS	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L)	1 ND 0.2 52 ND	2 1.40 ND 17 ND ND	4 1 170 ND ND	ND 2 ND 110 ND ND	2 5 2 230 ND	NA NA NA NA NA	3 TON NS 5 NTU	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L)	1 ND 0.2 52 ND ND 43	2 1.40 ND 17 ND ND 42	4 1 170 ND ND 139	ND 2 ND 110 ND ND 93	2 5 2 230 ND ND	NA NA NA NA NA NA NA NA	3 TON NS 5 NTU	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L)	1 ND 0.2 52 ND ND 43 ND 14 1.8	2 1.40 ND 17 ND ND 42 ND 8.8 3.7	4 1 170 ND ND 139 2 70 26	ND 2 ND 110 ND ND 93 ND 29 10	2 5 2 230 ND ND 190 3 92 34	NA	3 TON NS 5 NTU NS NS NS NS NS	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L)	1 ND 0.2 52 ND ND 43 ND 14	2 1.40 ND 17 ND ND 42 ND 8.8	4 1 170 ND ND 139 2 70	ND 2 ND 110 ND ND 93 ND 29	2 5 2 230 ND ND 190 3	NA	3 TON NS 5 NTU NS NS NS NS	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L) Total Hardness (CaCO3) (mg/L)	1 ND 0.2 52 ND ND 43 ND 14 1.8 4.1	2 1.40 ND 17 ND ND 42 ND 8.8 3.7 11	4 1 170 ND ND 139 2 70 26 136 283	ND 2 ND 110 ND ND 93 ND 29 10 65 110	2 5 2 230 ND ND 190 3 92 34 170 350	NA N	NS NS NS NS NS NS NS	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L)	1 ND 0.2 52 ND ND 43 ND 14 1.8 4.1	2 1.40 ND 17 ND ND 42 ND 8.8 3.7	4 1 170 ND ND 139 2 70 26 136	ND 2 ND 110 ND ND 93 ND 29 10 65	2 5 2 230 ND ND 190 3 92 34 170	NA N	NS N	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L) Total Hardness (CaCO3) (mg/L) TDS (mg/L) Specific Conductance (umhos/cm)	1 ND 0.2 52 ND ND 43 ND 14 1.8 4.1 41 64	2 1.40 ND 17 ND ND 42 ND 8.8 3.7 11 37 82	4 1 170 ND ND 139 2 70 26 136 283 752	ND 2 ND 110 ND ND 93 ND 29 10 65 110 386 683	2 5 2 230 ND ND 190 3 92 34 170 350 876	NA N	NS N	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L) Total Hardness (CaCO3) (mg/L) TDS (mg/L) Specific Conductance (umhos/cm) Chloride (mg/L)	1 ND 0.2 52 ND ND 43 ND 14 1.8 4.1 41 64 96	2 1.40 ND 17 ND ND 42 ND 8.8 3.7 11 37 82 140	4 1 170 ND ND 139 2 70 26 136 283 752 1190	ND 2 ND 110 ND ND 93 ND 29 10 65 110 386 683 48	2 5 2 230 ND ND 190 3 92 34 170 350 876 1344	NA N	3 TON	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits Substances that form ions when in water Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L) Total Hardness (CaCO3) (mg/L) TDS (mg/L) Specific Conductance (umhos/cm)	1 ND 0.2 52 ND ND 43 ND 14 1.8 4.1 41 64	2 1.40 ND 17 ND ND 42 ND 8.8 3.7 11 37 82	4 1 170 ND ND 139 2 70 26 136 283 752	ND 2 ND 110 ND ND 93 ND 29 10 65 110 386 683	2 5 2 230 ND ND 190 3 92 34 170 350 876	NA N	3 TON	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L) Total Hardness (CaCO3) (mg/L) TDS (mg/L) Specific Conductance (umhos/cm) Chloride (mg/L) Sulfate (mg/L) pH	1 ND 0.2 52 ND ND 43 ND 14 1.8 4.1 41 64 96	2 1.40 ND 17 ND ND 42 ND 8.8 3.7 11 37 82 140	4 1 170 ND ND 139 2 70 26 136 283 752 1190	ND 2 ND 110 ND ND 93 ND 29 10 65 110 386 683 48	2 5 2 230 ND ND 190 3 92 34 170 350 876 1344	NA N	3 TON	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits Substances that form ions when in water Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L) Total Hardness (CaCO3) (mg/L) TDS (mg/L) Specific Conductance (umhos/cm) Chloride (mg/L) Sulfate (mg/L) pH WATER DISTRIBUTION DATA SHEET	1 ND 0.2 52 ND ND 43 ND 14 1.8 4.1 41 64 96 2.8 1.4	2 1.40 ND 17 ND ND 42 ND 8.8 3.7 11 37 82 140	4 1 170 ND ND 139 2 70 26 136 283 752 1190 226	ND 2 ND 110 ND ND 93 ND 29 10 65 110 386 683 48 88	2 5 2 230 ND ND 190 3 92 34 170 350 876 1344 180 300	NA N	NS N	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L) Total Hardness (CaCO3) (mg/L) TDS (mg/L) Specific Conductance (umhos/cm) Chloride (mg/L) Sulfate (mg/L) pH	1 ND 0.2 52 ND ND 43 ND 14 1.8 4.1 41 64 96 2.8 1.4	2 1.40 ND 17 ND ND 42 ND 8.8 3.7 11 37 82 140	4 1 170 ND ND 139 2 70 26 136 283 752 1190 226	ND 2 ND 110 ND ND 93 ND 29 10 65 110 386 683 48 88	2 5 2 230 ND ND 190 3 92 34 170 350 876 1344 180 300	NA N	3 TON NS NS 5 NTU NS NS NS NS NS 1000 mg/L 1600 umhos/cm 250 mg/L 500 mg/L 500 mg/L 500 mg/L 500 mg/L	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L) Total Hardness (CaCO3) (mg/L) TDS (mg/L) Specific Conductance (umhos/cm) Chloride (mg/L) PH WATER DISTRIBUTION DATA SHEET BACTERIOLOGICAL (% Present)	1 ND 0.2 52 ND ND 43 ND 14 1.8 4.1 41 64 96 2.8 1.4 7.9	2 1.40 ND 17 ND ND 42 ND 8.8 3.7 11 37 82 140 12 25 9.3	4 1 170 ND ND 139 2 70 26 136 283 752 1190 236 7.4	ND 2 ND 110 ND 110 ND 93 ND 29 10 65 110 386 683 48 88 7.1	2 5 2 230 ND ND 190 3 92 34 170 350 876 1344 180 300 7.9	NA N	NS N	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits Substances that form ions when in water Erosion of natural deposits Erosion of natural deposits
Potassium (K) (mg/L) Turbidity (NTU)2 Bicarbonate (HCO3) (mg/L) Carbonate (CO3) (mg/L) Hydroxide Alkalinity (OH) (mg/L) Total Alkalinity (CaCO3)(mg/L) Boron (B) (mg/L) Calcium (Ca) (mg/L) Magnesium (Mg) (mg/L) Sodium (Na) (mg/L) Total Hardness (CaCO3) (mg/L) TDS (mg/L) Specific Conductance (umhos/cm) Chloride (mg/L) Sulfate (mg/L) pH WATER DISTRIBUTION DATA SHEET BACTERIOLOGICAL (% Present) Coliform Density	1 ND 0.2 52 ND ND 43 ND 14 1.8 4.1 41 64 96 2.8 1.4 7.9	2 1.40 ND 17 ND ND 42 ND 8.8 3.7 11 37 82 140 12 25 9.3	4 1 170 ND ND 139 2 70 26 136 283 752 1190 236 7.4	ND 2 ND 110 ND 110 ND 93 ND 29 10 65 110 386 683 48 88 7.1	2 5 2 230 ND ND 190 3 92 34 170 350 876 1344 180 300 7.9	NA N	3 TON NS NS 5 NTU NS NS NS NS NS 1000 mg/L 1600 umhos/cm 250 mg/L 500 mg/L 500 mg/L 500 mg/L 500 mg/L	Naturally occurring organic materials Erosion of natural deposits Soil runoff Erosion of natural deposits Substances that form ions when in water Erosion of natural deposits Erosion of natural deposits

DEFINITIONS

AL (Action Level): The concentration of a contaminant, which, if exceeded, triggers treatment or other requirements, which a water system must follow.

MCL (Maximum Contaminant Level): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible.

Secondary MCLs (SMCL): Are set to protect the odor, taste, and appearance of drinking water.

MCLG (Maximum Contaminant Level Goal): The level of a contaminant in drinking water below, which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency. PHG (Public Health Goal): The level of a contaminant in drinking water below which there is no known or

expected risk to health. PHGs are set by the California Environmental Protection Agency.

PDWS (Primary Drinking Water Standard): MCLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

TT (Treatment Technique): A required process intended to reduce the level of a contaminant in drinking water.

NA: Not applicable.ND: Not detected.NS: No standard.

NTU (Nephelometric Turbidity Units): Measurement of the clarity, or turbidity, of water.

ppb (Parts Per Billion): One part per billion (or micrograms per liter).**ppm (Parts Per Million):** One part per million (or milligrams per liter).

pCi/L (**Picocuries Per Liter**): A measure of the natural rate of radioactive disintegration. **umhos/cm** (**Micromhos Per Centimeter**): A measure of electrical conductance.

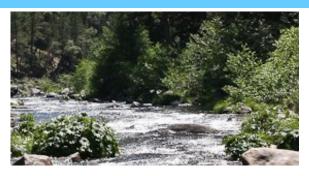
DISINFECTION PRACTICES

The City effectively disinfects the drinking water using three processes: ultraviolet (UV) light, chlorine, and chloramines.

CHLORAMINES: Chloramines are used late spring, summer, and early fall depending on the temperature of the water. When the water temperature rises the chlorine residual can degrade quickly and form Trihalomethanes, as seen in the above chart. Chloramine is a compound formed by adding ammonia to chlorine. This compound is a more stable disinfectant that does not degrade as quickly as just chlorine. Also chloramines do not react as quickly with the organics in the water and form Trihalomethanes. However, chloramines must be removed for kidney dialysis treatment and may require recalibration of dialysis equipment. If you are receiving kidney dialysis treatment, please contact your doctor or dialysis technician.

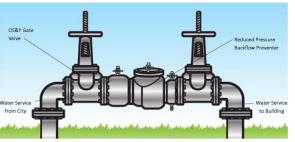
SAMPLING RESULTS SHOWING THE DETECTION OF LEAD AND COPPER										
Lead and Copper (To be completed only if there was a detection of lead or copper in the last sample set)	# Of Samples Collected	90TH Percentile Level Detected	# Sites Exceeding AL	AL	MCLG	Typical Source of Contaminant				
Lead (ppb)	48	0.001	0	0.015	0	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits.				
Copper (ppm)	48	0.43	0	1.3	1.3	Internal corrosion of household water plumbing systems; erosion of natural deposits; leaching from wood preservatives.				

Note: The City's water is in complete compliance with regulations related to lead and copper.



Stanislaus River Water

The City of Tracy is committed to providing a safe, reliable and affordable water supply to meet the needs of the community today and in the future. The City has participated with the cities of Manteca, Lathrop, Escalon, and the South San Joaquin Irrigation District to bring high quality Sierra water from the Stanislaus River. This water source has increased the reliability of City water supplies by having a third source of supply and redundancy in treatment facilities. Delivery of this water comprises the majority of water consumed in the City and is the only supply source used during the winter months. The Stanislaus River water supply is very soft water and has significantly reduced the minerals in the City's water supply. You may no longer need to use a water softener.



Backflow prevention assemblies protect our drinking water supply.

Cross Connection Protection

Backflow prevention assemblies are designed to allow water to flow into your home or office from the public water system but not allow water to flow in the reverse direction, creating effective cross connection protection. Reverse flow can carry untreated pollutants and contaminants back to the public water system, compromising the water quality for all customers. Backflow prevention assemblies are required to be tested annually to ensure they are effectively protecting the public water system. If your residence has an active well on the premises or your business has fire sprinklers and/or landscaping, you should have a backflow prevention assembly. For questions regarding annual testing requirements, please call Erich Delmas, Laboratory Supervisor at (209) 831-6325.

Water Source Assessment

An assessment of the drinking water sources for the City of Tracy's water system was completed in June 2001. The sources are considered most vulnerable to the following activities: airports (maintenance and fueling areas), gas stations (historic and current), mining activities (historic and current), septic and waste landfill dumps (historic and current). You may request a copy of the assessment by contacting the Water Treatment Superintendent, Lea Emmons, at (209) 831-6302.

The native groundwater under Tracy contains boron. Boron is a naturally occurring, non-carcinogenic, unregulated contaminant. Six of the City's wells contain elevated levels of boron. Although well water comprises only a small portion of the City's total water supply, well water does contain boron. Some pregnant women who drink water containing boron may have an increased risk of developmental effects in their baby, based on studies in laboratory animals.

SAMPLING RESULTS SHOWING TREATMENT OF SURFACE WATER SOURCES

<u>Treatment Technique:</u> A required process intended to reduce the level of a contaminant in drinking water (type of approved filtration technology used).

Turbidity of the filtered water must:

- Be less than or equal to 0.3 NTU in 95% of measurements in a month.
- Not exceed I NTU for more than eight consecutive hours.
- 3. Not exceed 3 NTU at any time.

Turbidity Performance Standards: Turbidity (measured in NTU) is a measurement of the cloudiness of water and is a good indicator of water quality and filtration performance. Turbidity results, which meet performance standards, are considered to be in compliance with filtration requirements (that must be met through the water treatment process).

Lowest monthly percentage of samples that met Turbidity Performance Standard No.1: 100%. Highest single turbidity measurement during 2018 was .18 NTU.