

# 2020 Annual Consumer Confidence Report

Ontario's annual water quality report takes you inside the world of your high-quality drinking water. This report is designed to provide important information about the sources of your water, what it contains, and how it compares to standards set by regulatory agencies.

The Ontario Municipal Utilities Company is pleased to report that during the past year, water delivered to your home or business meets or surpassed all federal drinking water requirements.

Safe and reliable drinking water supplies are necessary for public health, fire protection, economic development, and the overall quality of life. Businesses and residents are encouraged to use the drinking water supplies as efficiently as possible.

Este informe contiene información muy importante sobre su agua para beber. Favor de comunicarse Ontario Municipal Utilities Company a 1425 South Bon View Avenue, Ontario, CA 91761 para asistirlo en español.



### **2020 CONSUMER CONFIDENCE REPORT**

To ensure safe drinking water, public water systems must comply with federal and state drinking water standards. The Ontario Municipal Utilities Company and its trained, certified water quality professionals collect thousands of water samples that are delivered to a State certified laboratory for analysis. We are pleased to report there were no federal water quality violations during 2020.

The public is encouraged to participate on issues concerning the City's water. Meetings of the Ontario City Council are scheduled on the first and third Tuesday of each month beginning at 6:30pm at Ontario City Hall, 303 East "B" Street, Ontario, CA 91761. Check the City's website at https://www.ontarioca.gov/calendar or call (909) 395-2000 for more information.

Para garantizar agua potable segura, los sistemas públicos de agua deben cumplir con las normas federales y estatales de agua potable. El Municipal Utilities Company Ontario y sus capacitados, certificados profesionales de la calidad del agua recogen miles de muestras de agua que se entregan a un laboratorio certificado por el estado para su análisis. Nos complace informar que no había violaciónes federales de calidad del agua durante el año 2020.

El público es alentado a participar en asuntos con respecto al agua de la Ciudad. Las reuniones del establecimiento de Ontario se programa el primer y tercer martes de cada mes a las 6:30 P.M., por la calle 303 "B" Street, Ontario. Para más información, vaya al Web site de la Ciudad https://www.ontarioca.gov/calendar o llame (909) 395-2000.

### **Regulatory Information**

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA's Safe Drinking Water Hotline (1-800-426-4791).

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

### CONTAMINANTS THAT MAY BE PRESENT IN SOURCE WATER INCLUDE:

- Microbial contaminants, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- Inorganic contaminants, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- Pesticides and herbicides, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential
- Organic chemical contaminants, including synthetic and volatile organic chemicals, that are by-products of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems.
- Radioactive contaminants, that can be naturally-occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, the U.S. Environmental Protection Agency (U.S. EPA) and the State Water Resources Control Board (State Water Board) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. State Water Board regulations also establish limits for contaminants in bottled water that provide the same protection for public health.

### Nitrate

Nitrate in drinking water at levels above 10 mg/L is a health risk for infants of less than six months of age. Such nitrate levels in drinking water can interfere with the capacity of the infant's blood to carry oxygen, resulting in a serious illness; symptoms include shortness of breath and blueness of the skin. Nitrate levels above 10 mg/L may also affect the ability of the blood to carry oxygen in other individuals, such as pregnant women and those with certain specific enzyme deficiencies. If you are caring for an infant, or you are pregnant, you should ask advice from your health care provider.

### Lead

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. Ontario Municipal Utilities Company is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at <a href="http://www.epa.gov/safewater/lead">http://www.epa.gov/safewater/lead</a>.

### Total Coliform Rule

This Consumer Confidence Report (CCR) reflects changes in drinking water regulatory requirements during 2016. All water systems are required to comply with the state Total Coliform Rule. Effective April 1, 2016, all water systems are also required to comply with the federal Revised Total Coliform Rule. The new federal rule maintains the purpose to protect public health by ensuring the integrity of the drinking water distribution system and monitoring for the presence of microbials (i.e., total coliform and *E. coli* bacteria). The U.S. EPA anticipates greater

public health protection as the new rule requires water systems that are vulnerable to microbial contamination to identify and fix problems. Water systems that exceed a specified frequency of total coliform occurrences are required to conduct an assessment to determine if any sanitary defects exist. If found, these must be corrected by the water system.

Coliforms are bacteria that are naturally present in the environment and are used as an indicator that other, potentially harmful, waterborne pathogens may be present or that a potential pathway exists through which contamination may enter the drinking water distribution system. We found coliforms indicating the need to look for potential problems in water treatment or distribution. When this occurs, we are required to conduct an assessment to identify problems and to correct any problems that were found during these assessments. During the past year we were required to conduct one Level 1 Assessment. One Level 1 Assessment was completed in September 2020. In addition, we were required to take two corrective actions and we completed one of these actions on October 31, 2020. As the City continues to improve its water pipelines, the second corrective action is expected to be completed in December 2021.

### **Important Health Information**

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. U.S. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline (1-800-426-4791).

### Kidney Dialysis/Aquariums

Customers who have unique water-quality needs and who use specialized home treatments, such as kidney dialysis machines, should make the necessary adjustments to remove chloramines. Customers who have fish tanks in their homes or businesses should also take precautions to remove chloramines prior to adding water to tanks.

### **Source Water Assessment**

An assessment of the drinking water sources for the Ontario Municipal Utilities Company was completed in May 2002 and reviewed by the State Board in 2019. The sources are considered most vulnerable to the following activities associated with contaminants detected in the water supply: high density housing; sewer collection systems; parks; golf courses; the application of fertilizers, pesticides, herbicides; metal plating, finishing and fabricating; wood pulp processing and paper mills; and recreational use of surface water sources.

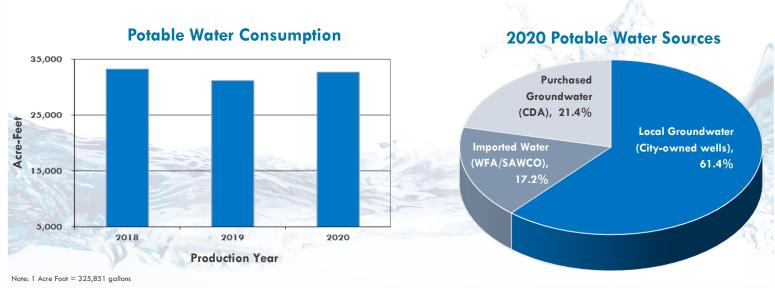
A copy of the complete assessment is available at State Water Board San Bernardino District Office at 464 West 4<sup>th</sup> Street, Suite 437, San Bernardino, CA 92401. You may request a summary of the assessment be sent to you by contacting the State Water Board's San Bernardino District Office or Ontario Municipal Utilities Company at (909) 395-2678.

### **Ontario's Drinking Water Sources**

Ontario's water supplies are comprised of surface water and groundwater. Surface water originates from the State Water Project (supplied by the Metropolitan Water District of Southern California) and is treated locally by Water Facilities Authority (WFA) using conventional water treatment methods. Groundwater supplies consist of City-owned wells (local groundwater), San Antonio Water Company (SAWCO), and Chino Desalter Authority (CDA) wells. To view particular water sources based on your location, see the map on page 9.

### Water Quality Monitoring

In 2020, the Ontario Municipal Utilities Company collected over 20,500 potable water samples to test for more than 150 possible constituents. Samples were collected at water sources and throughout the distribution system. The following tables (pages 5-8) have been compiled for your information showing which constituents were detected in the City's drinking water during 2020 as compared to state and federal water quality standards.



### **Abbreviations & Definitions**

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AL CELL/mil	Action Level	MRL	Minimum Reporting Level set by EPA for unregulated contaminant monitoring	ppq	parts per quadrillion or picograms per liter (pg/L)
CFU/mL DLR	Colony-forming units per milliliter  Detection limits for the purpose of reporting: State determined level	NA	Not Applicable: no State or Federal standards established	ppt	parts per trillion or nanograms per liter (ng/L)
	that a test can detect the constituent	ND	Not Detected: sample was taken and	RAA	Running Annual Average
HPC	Heterotrophic Plate Count: a		constituent was not detected	TON	Threshold Odor Number
	bacteriological test that counts the	NL	Notification Level	TT	Treatment Technique
	number of bacteria per milliliter of sample	NR	No Range: all results were the same value	μS/cm	microSiemen per centimeter; or micromho per centimeter (µmho/cm)
LRAA	Location Running Annual Average	NTU	Nephelometric Turbidity Units	" = "	Equal
MCL	Maximum Contaminant Level	pCi/L	picoCuries per Liter	" > "	Greater than
MCLG	Maximum Contaminant Level Goal	PHG	Public Health Goal	" < "	Less than
MRDL	Maximum Residual Disinfectant Level	ppb	parts per billion or micrograms per liter	" ≤ "	Less than or equal to
MRDLG	Maximum Residual Disinfectant Level		(µg/L)	"#"	Number
	Goal	ppm	parts per million or milligrams per liter $(mg/L)$	" % "	Percent

One part per **million (ppm)**IS LIKE

1 second in 11.5 days

1 drop in 13.6 gallons

One part per **billion (ppb)**IS LIKE

1 second in 32 years

1 drop in 13,563 gallons

One part per **trillion (ppt)**IS LIKE

1 second in 32,000 years

1 drop in 13.6 million gallor

One part per **quadrillion (ppq)**IS LIKE

1 second in 32 million years

1 drop in 13.6 billion gallons

### **Definitions**

**90th Percentile:** The value in a data set in which 90 percent of the set is less than or equal to this value.

**Disinfection By-Product:** Compounds which are formed from mixing of organic or mineral precursors in the water with ozone, chlorine or chloramine. Total Trihalomethanes and Haloacetic Acids are disinfection by-products.

**Level 1 Assessment:** A Level 1 Assessment is a study of the water system to identify potential problems and determine (if possible) why total coliform bacteria have been found in our water system.

**Locational Running Annual Average (LRAA):** The Running Annual Average (RAA) at one sample location.

Maximum Contaminant Level (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

Maximum Contaminant Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. EPA.

Maximum Residual Disinfectant Level (MRDL): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

**Notification Level (NL):** Notification levels are health –based advisory levels established by the State Board for chemicals in drinking water that lack maximum contaminant levels (MCLs).

Primary Drinking Water Standard (Primary Standard): MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

**Public Health Goals (PHG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

**Regulatory Action Level (AL):** The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**Running Annual Average (RAA):** The yearly average which is calculated every 3 months using the previous 12 months' data.

Secondary Drinking Water Standard (Secondary Standard): MCLs for contaminants that do not affect health but are used to monitor the aesthetics of the water.

**Treatment Technique (TT):** A required process intended to reduce the level of a contaminant in drinking water.

Ontario Municipal Utilities Company - 2020 Distribution System Table											
CONSTITUENT	UNITS	MCL or [AL] or (MRDL)	PHG or [MCLG] or (MRDLG)	CA DLR [MRL]	Range Average	OMUC's Entire Distribution System	Major Sources in Drinking Water				
MICROBIOLOGICAL											
Total Coliform (State Total Coliform Rule)	%	5	[0]	NA	Highest Total Coliform Monthly % Positive # of Months in Violation	5.47% 1	Naturally present in the environment.				
Heterotrophic Plate Count (HPC)	CFU/mL	TT	NA	[1]	Range <b>Average</b>	ND to 3,900 <b>3.73</b>	Naturally present in the environment.				
				PHYSICA	L PARAMETEI	RS					
рН	pH Unit	6.5 - 8.5	NA	[1]	Range <b>Average</b>	7.3 to 8.9 <b>8.0</b>	Measurement of hydrogen ion activity.				
Turbidity	NTU	5	NA	0.1	Range <b>Average</b>	ND to 1.1 <b>0.13</b>	Soil runoff.				
		DISIN	IFECTION BY-	PRODUC	TS AND DISIN	FECTANT RESID	UALS				
Haloacetic Acids (HAA₅)	ppb	LRAA = 60	NA	2.0^	Range Highest LRAA	ND to 17	Byproduct of drinking water disinfection.				
Total Trihalomethanes (TTHMs)	ppb	LRAA = 80	NA	1	Range <b>Highest LRAA</b>	0.5 to 56 <b>52</b>	Byproduct of drinking water disinfection.				
Total Chlorine Residual (chloramines & chlorine)	ppm	(4)	(4)	NA	Range <b>Average</b>	ND to 1.9	Drinking water disinfectant added for treatment.				
METALS AT CONSUMER'S PLUMBING (2018)											
Copper	ppb	[1300]	300	50	NA	90th percentile: 160 ppb (0 exceeded AL/ 57 samples)	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives.				
Lead	ppb	[15]	0.2	5	NA	90th percentile: ND (0 exceeded AL/ 57 samples)	Internal corrosion of household plumbing systems; discharges from industrial manufacturers; erosion of natural deposits.				
		UNREG	ULATED CON	AMINA	NT MONITORII	NG RULE 4 (201	8-2020)				
Anatoxin-a	ppb	NA	NA	[0.03]	Range <b>Average</b>	NR <b>ND</b>	Cyanobacteria, formerly referred to as blue-green algae, are found naturally in lakes, rivers, ponds and other surface waters.				
Cylindrospermopsin	ppb	NA	NA	[0.09]	Range <b>Average</b>	NR <b>ND</b>	Cyanobacteria, formerly referred to as blue-green algae, are found naturally in lakes, rivers, ponds and other surface waters.				
Total Microcystins	ppb	NA	NA	[0.3]	Range <b>Average</b>	NR <b>ND</b>	Cyanobacteria, formerly referred to as blue-green algae, are found naturally in lakes, rivers, ponds and other surface waters.				
Manganese	ppb	NA	NA	[0.3]	Range <b>Averag</b> e	ND to 6.2 <b>0.30</b>	Naturally-occurring element; commercially available in combination with other elements and minerals; used in steel production, fertilizer, batteries and fireworks; drinking water and wastewater treatment chemical; essential nutrient.				
1-butanol	ppb	NA	NA	[0.3]	Range <b>Average</b>	ND to 2.6 <b>0.10</b>	Used as a solvent, food additive and n production of other chemicals.				
Bromochloroacetic acid (BCAA)	ppb	NA	NA	[0.3]	Range <b>Average</b>	ND to 5.7 <b>2.4</b>	Byproduct of drinking water disinfection.				
Bromodichloroacetic acid (BDCAA)	ppb	NA	NA	[0.5]	Range <b>Average</b>	ND to 6.6 <b>2.3</b>	Byproduct of drinking water disinfection.				
Chlorodibromoacetic acid (CDBAA)	ppb	NA	NA	[0.3]	Range <b>Average</b>	ND to 4.3 <b>2.0</b>	Byproduct of drinking water disinfection.				
Tribromoacetic acid (TBAA)	ppb	NA	NA	[2.0]	Range <b>Average</b>	ND to 6.0 1.6	Byproduct of drinking water disinfection.				

<sup>^</sup>DLR =1.0 ppb for each HAA5 analyte except for monochloroacetic acid which has a DLR = 2.0ppb.

Unregulated contaminant monitoring helps USEPA and the State Water Resources Control Board to determine where certain contaminants occur and whether the contaminants need to be regulated.

The State allows us to monitor for some contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of our data, though representative, are more than one year old.

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	Ontai	rio Munici	pal Uti	lities Co	mpany	- 2020	Source	Water (	Quality Table		
	Local Ground Water WFA w/ SAWCO										
CONSTITUENT	UNITS	MCL [NL]	PHG or [MCLG]	CA DLR [MRL]	Average	Range	Average	Range	Major Sources in Drinking Water		
PRIMARY STANDARDS - Mandatory Health-Related Standards											
CLARITY											
Combined Filter	NTU	TT = 1 NTU	NA	NA	NA	NA	0.15 H	lighest	Soil Runoff.		
Effluent Turbidity	%	TT					% ≤ 0.3	100%			
ORGANIC CHEMICALS											
Dibromochloropro- pane (DBCP)	ppt	200	1.7	10	1.9	ND to 31	NA	NA	Banned nematicide that may still be present in soils due to runoff or leaching from former use on soybeans, cotton, vineyards, tomatoes, and tree fruit.		
RADIOACTIVITY											
Gross Alpha Particle Activity	pCi/L	15	[0]	3	1.6	0.5 to 4.2	0.8	ND to 4.0	Erosion of natural deposits.		
Uranium	pCi/L	20	0.43	1	NA	NA	4.1	NR	Erosion of natural deposits.		
INORGANIC CHEMIC	CALS					I	I				
Aluminum	ppb	1000	600	50	ND	NR	53	ND to 110	Erosion of natural deposits; residue from some surface water treatment processes.		
Arsenic	ppb	10	0.004	2.0	0.4	ND to 3.5	0.6	ND to 2.8	Erosion of natural deposits; runoff from orchards; glass and electronics production wastes.		
Barium	ppm	1	2	0.1	0.1	ND to 0.1	NA	NA	Discharges of oil drilling wastes and from metal.		
Chromium, Total	ppb	50	[100]1	10	3.5	1.8 to 5.5	NA	NA	Discharge from steel and pulp mills and chrome plating; erosion of natural deposits.		
Fluoride (Naturally- occurring)	ppm	2	1	0.1	0.18	0.10 to 0.34	0.13	ND to 0.23	Erosion of natural deposits; water additive that promotes strong teeth; discharge from fertilizer and aluminum factories.		
Nitrate (as Nitrogen)	ppm	10	10	0.4	2.5	0.7 to 4.9	1.4	ND to 4.2	Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits.		
Nitrate & Nitrite (as Nitrogen)	ppm	10	10	[0.2]	2.5	1.1 to 4.9	1.4	ND to 4.2	Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits.		
Perchlorate	ppb	6	1	4	1.6	0.4 to 4.1	NA	NA	Perchlorate is an inorganic chemical used in solid rocket propellant, fireworks, explosives, flares, matches, and a variety of industries. It usually gets into drinking water as a result of environmental contamination from historic aerospace or other industrial operations that used or use, store, or dispose of perchlorate and its salts.		
			SE	CONDARY	STANDARDS	S - Aestheti	c Standards		1		
Aluminum	ppb	200	600	50	ND	NR	53	ND to 110	Erosion of natural deposits; residue from some surface water treatment processes.		
Chloride	ppm	500	NA	[1]	7.9	4.6 to 15	49	26 to 62	Runoff or leaching from natural deposits; seawater influence.		
Iron	ppb	300	NA	100	ND	NR	65	ND to 390	Leaching from natural deposits; industrial wastes.		

	Ontar	io Munic	ipal Uti	ilities Co	ompany	- 2020	Source	Water	Quality Table
					Local Gro	und Water	Imported Water, WFA w/ SAWCO		
CONSTITUENT	UNITS	MCL [NL]	PHG or [MCLG]	CA DLR [MRL]	Average	Range	Average	Range	Major Sources in Drinking Water
Manganese	ppb	50	NA	20	ND	NR	6.0	ND to 30	Leaching from natural deposits.
Odor Threshold	TON	3	NA	1	ND	NR	2.0	1.0 to 3.0	Naturally-occurring organic materials.
Specific Conductance (EC)	µS/cm	1600	NA	[1]	331	290 to 440	452	360 to 610	Substances that form ions when in water; seawater influence.
Sulfate	ppm	500	NA	0.5	15	6.0 to 33	41	34 to 51	Runoff or leaching from natural deposits; industrial wastes.
Total Dissolved Solids	ppm	1000	NA	NA	211	180 to 290	262	190 to 390	Runoff or leaching from natural deposits.
Turbidity	NTU	5	NA	[0.10]	0.18	0.12 to 0.29	0.50	ND to 2.1	Soil runoff.
					OTHER PAR	RAMETERS			
Alkalinity (Total)	ppm	NA	NA	[3]	141	120 to 160	111	73 to 240	Naturally-occurring carbonate; measures the water's ability to neutralize acid.
Bicarbonate	ppm	NA	NA	[3]	141	120 to 160	137	89 to 300	
Boron	ppb	[1000]	NA	100	NA	NA	104	120 to 160	Naturally-occurring element; runoff or leaching from natural deposits and fertilizer use; industrial wastes.
Calcium	ppm	NA	NA	[1]	42	27 to 54	37	23 to 92	Naturally-occurring mineral.
Corrosivity (Aggressiveness Index)	Al	NA	NA	NA	NA	NA	12.1	11.8 to 12.6	Elemental balance in water; affected by temperature, other factors.
Corrosivity (Saturation Index)	SI	NA	NA	NA	NA	NA	0.62	0.39 to 1.37	Elemental balance in water; affected by temperature, other factors.
Hardness as CaCO <sub>3</sub> (Total)	ppm	NA	NA	[3]	134	83 to 190	138	80 to 300	Naturally-occurring mineral; the sum of calcium and magnesium present in water.
Hexavalent Chromium	ppb	2 (see footnote)	0.02	[1]	3.6	1.8 to 5.9	NA	NA	Discharge from electroplating factories, leather tanneries, wood preservation, chemical synthesis, refractory production, and textile manufacturing facilities; erosion of natural deposits.
Magnesium	ppm	NA	NA	[1]	7.3	3.5 to 13	11.1	8.4 to 18	Naturally-occurring mineral.
рН	pH units	NA	NA	[1]	8.1	7.9 to 8.2	8.0	7.8 to 8.1	Measurement of hydrogen ion activity.
Potassium	ppm	NA	NA	[1]	1.8	1.6 to 2.4	2.3	2.0 to 2.5	Naturally-occurring mineral.
Sodium	ppm	NA	NA	[1]	21	15 to 37	37.6	9.8 to 50	Naturally-occurring mineral; seawater influence.
Total Organic Carbon (TOC)	ppm	TT	NA	0.3	NA	NA	2.2	1.8 to 2.6	Various natural and man-made sources.
Vanadium	ppb	[50]	NA	3.0	NA	NA	2.6	ND to 5.0	Various natural and man-made sources.

<sup>&</sup>lt;sup>1</sup> California withdrew the PHG for total chromium in November 2011.

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<sup>&</sup>lt;sup>2</sup> There is currently no MCL for hexavalent chromium. The previous MCL of 10ppb was withdrawn on September 11, 2017. The Ontario Municipal Utilities Company (OMUC) will continue to monitor this constituent.

The State allows us to monitor for some contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of our data, though representative, are more than one year old.

Provide   Naturally		Ont	ario M	lunicipo	al Utilit	ies Cor	npany	- 2020	Sourc	e Wate	r Qual	ity Table	
Constituent   Units   MicCo   MicCo   CAD RI   Average   Range					_				_				
Note   Part	CONSTITUENT	UNITS				Average	Range	Average	Range	Average	Range	Major Sources in Drinking Water	
Chromium, Total   ppb   50   [100]*   10   ND   NR   0.11   ND to 3.9   2.5   2.2 to 5.6				PRIM	ARY STA	NDARDS	- Manda	tory Heal	th-Relate	d Standar	ds		
Chromium, Total   ppb   50   [100]*   10   ND   NR   0.11   ND   0.39   2.5   2.5   5.6   chrome plating; erosion of natural deposits, were recurring)   ppm   2	INORGANIC CHEM	ICALS											
Mindel   Part	Chromium, Total	ppb	50	[100]*	10	ND	NR	0.11		2.5		1 0,	
Second Normal   Second Norma	(Naturally-	ppm	2	1	0.1	ND	NR	ND		0.11	NR	additive that promotes strong teeth; discharge from fertilizer and aluminum	
Chloride		ppm	10	10	0.4	4.2		5.1		7.2		Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits.	
Chloride					SECON	DARY ST	ANDARD	S - Aestho	etic Stanc	lards			
Specific Conductance (EC)   US/cm   1600   NA   [1]   580   580 to 600   491   520   737   430 to 600   77	Chloride	ppm	500	NA	[1]	105		79		138		ũ	
Sulface   ppm   SOU   NA   D.5   1.1   1.2   8.6   13   22   NR   deposits; industrial westes.	l .	μS/cm	1600	NA	[1]	580	580 to	491	350 to	737	430 to	Substances that form ions when in	
Solida (TDS)	Sulfate	ppm	500	NA	0.5	1.1		8.6		22	NR	_	
Turbidity		ppm	1000	NA	NA	450		308		587		_	
Alkalinity (Total)   ppm   NA   NA   [3]   97   74 to   106   100 to   130 to   130 to   150 measures the water's ability to neutralize acid.    Calcium   ppm   NA   NA   [1]   53   47 to   58   53   44 to   54   96   59 to   100 Naturally-occurring mineral.    Hardness as   CaCO <sub>3</sub> (Total)   ppm   NA   NA   [3]   180   160 to   200   164   130 to   287   170 to   300		NTU	5	NA	[0.10]	0.1		0.15		0.15			
Alkalinity (Total)         ppm         NA         NA         [3]         97         A to 120         106         100 to 150         132 to 150         133 to neutralize acid.         measures the water's ability to neutralize acid.           Calcium         ppm         NA         NA         [1]         53         47 to 58         53         44 to 54         96         59 to 100         Naturally-occurring mineral.           Hardness as CaCO <sub>3</sub> (Total)         ppm         NA         NA         [3]         180         160 to 200         164         130 to 170         287         170 to 300         Naturally-occurring mineral, the sum-calcium and magnesium present in water.           Hexavalent Chromium         ppb         **         0.02         [1]         ND         NR         ND         ND to 2.3         2.0 to 300         Naturally-occurring mineral, the sum-calcium and magnesium present in water.           Magnesium         ppm         NA         NA         [1]         12         11 to 3.7         7.1 to 4.0         2.0         2.0 to 15.4         Discharge from electroplating factor in water.           Magnesium         ppm         NA         NA         [1]         12         11 to 3.7         7.1 to 5.2 to 7.1 to 11         Naturally-occurring mineral.           Perfluoroctanoic						ОТ	HER PAR	AMETERS					
Marting   Part   NA   NA   NA   NA   NA   NA   NA   N	Alkalinity (Total)	ppm	NA	NA	[3]	97		106		132		measures the water's ability to	
Part	Calcium	ppm	NA	NA	[1]	53		53		96		Naturally-occurring mineral.	
Hexavalent Chromium		ррт	NA	NA	[3]	180		164		287		Ŭ i	
Magnesium		ppb	**	0.02	[1]	ND	NR	ND		2.3		Discharge from electroplating factories, leather tanneries, wood preservation, chemical synthesis, refractory production, and textile manufacturing facilities; erosion of natural deposits.	
Perfluorooctanoic Acid (PFOA)  Perfluorooctane-sulfonic Acid (PFOS)  Potassium  Popm  NA  NA  [1]  NA  NA  NA  [1]  NA  NA  NA  NA  NA  NA  NA  NA  NA  N	Magnesium	ppm	NA	NA	[1]	12		7.2		10		Naturally-occurring mineral.	
Perfluoroctanels Acid (PFOA)  Perfluoroctanels ppt [5.1] *** [2] ND NR ND NR 2.6 ND to 3.6 leaching from water resistant consumproducts; fire-retarding foam release products; fire-reta	рН	pH units	NA	NA	[1]	7.9		8.1		8.0		Measurement of hydrogen ion activity.	
sulfonic Acid (PFOS)         ppt         [6.5]         ***         [2]         ND         NR         ND         NR         3.9         ND to 7.7         leaching from water resistant consumproducts; fire-retarding foam release           Potassium         ppm         NA         NA         [1]         1.2         1.0 to 1.3         1.8         1.4 to 2.1         2.4         2.0 to 2.4         Naturally-occurring mineral.           Sodium         ppm         NA         NA         [1]         30         29 to 31         30         24 to 31         35         23 to 36 influence.           Total Silica         ppm         NA		ppt	[5.1]	***	[2]	ND	NR	ND	NR	2.6		Discharge from industrial factories; leaching from water resistant consumer products; fire-retarding foam releases.	
Potassium         ppm         NA         NA         [1]         1.2         1.3         1.8         2.1         2.4         2.4         Naturally-occurring mineral.           Sodium         ppm         NA         NA         III         30         29 to 31         30         24 to 31         35         23 to 36 influence.           Total Silica         ppm         NA         NA         NA         NA         NR         19         13 to 25         NR           14 Diagram         ppm         NA         NA         NA         NA         NA         NA         NA         NA	sulfonic Acid	ppt	[6.5]	***	[2]	ND	NR	ND	NR	3.9		Discharge from industrial factories; leaching from water resistant consumer products; fire-retarding foam releases.	
Sodium         ppm         NA         NA         [1]         30         31         30         31         35         36 influence.           Total Silica         ppm         NA         NA         NA         NA         NA         19         13 to 25         NR           14 Diagram         ppm         NA	Potassium	ppm	NA	NA	[1]	1.2		1.8		2.4		Naturally-occurring mineral.	
Total Silica	Sodium	ppm	NA	NA	[1]	30		30		35		Naturally-occurring mineral; seawater influence.	
	Total Silica	ppm	NA	NA	NA	8.0	NR	19		25	NR		
	1,4-Dioxane	ppb	[1]	NA	[0.07]	NA	NA	NA	NA	0.26			

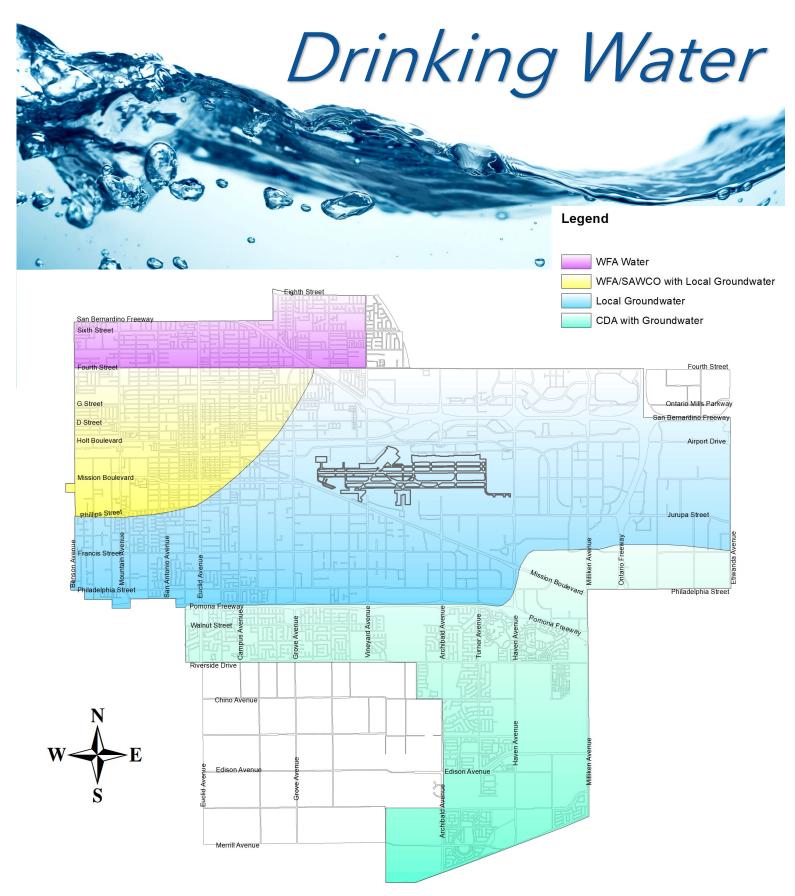
<sup>\*</sup> California withdrew the PHG for total chromium in November 2011.

<sup>\*\*</sup> There is currently no MCL for hexavalent chromium. The previous MCL of 10ppb was withdrawn on September 11th, 2017. The Ontario Municipal Utilities Company (OMUC) will continue to monitor this constituent.

<sup>\*\*\*</sup> The California Office of Environmental Health Hazard Assessment is developing a PHG for PFOA & PFOS. There is no established PHG at this time.

All water quality data reported for CDA and JCSD were taken from treated water sample locations with the exemption of PFAS data from Ion Exchange Plant (870 Zone). JCSD has removed all sources above the response level (10 ppt) and will not be used until treatment is in place. Any source above the notification level (est. August 22, 2019) has been put on emergency standby only.

# The Source of Your



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## **City Officials**

Mayor Paul S. Leon

Mayor pro Tem Alan D. Wapner

Jim W. Bowman
Debra Dorst-Porada
Ruben Valencia

City Manager
Scott Ochoa

Utilities General Manager
Scott Burton



