2020 Consumer Confidence Report

Water System Name: Olivet University 3301863 Report Date: 3/23/2021

We test the drinking water quality for many constituents as required by state and federal regulations. This report shows the results of our monitoring for the period of January 1 to December 31, 2020 and may include earlier monitoring data.

Este informe contiene información muy importante sobre su agua para beber. Favor de comunicarse <u>OLIVET</u> <u>UNIVERSITY</u> a 36401 Tripp Flats Rd- Anza, CA 92539 para asistirlo en español.

Type of water source(s) in use: Groundwater / Wells

Name & general location of source(s): 36401 Tripp Flats Road, Anza CA 92539

Drinking Water Source Assessment information:

A source water assessment was conducted for Olivet university (Anza Trinity) in December 2002. The sources were considered most vulnerable to the following activities not associated with any detected contaminants: agricultural irrigation. A detailed copy of the assessment is available at Riverside County Department of Environmental Health.

Time and place of regularly scheduled board meetings for public participation: N/A

For more information, contact:

Paul Gorino – Water Operator

Phone: (951) 663-0951

Ezra Bartovic - Facility Director

(646) 807-5259

TERMS USED IN THIS REPORT

Maximum Contaminant Level (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

Maximum Contaminant Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (U.S. EPA).

Public Health Goal (PHG): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

Maximum Residual Disinfectant Level (MRDL): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

Primary Drinking Water Standards (PDWS): MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

Secondary Drinking Water Standards (SDWS): MCLs for contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.

Treatment Technique (TT): A required process intended to reduce the level of a contaminant in drinking water.

Regulatory Action Level (AL): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

Variances and Exemptions: Permissions from the State Water Resources Control Board (State Board) to exceed an MCL or not comply with a treatment technique under certain conditions.

Level 1 Assessment: A Level 1 assessment is a study of the water system to identify potential problems and determine (if possible) why total coliform bacteria have been found in our water system.

Level 2 Assessment: A Level 2 assessment is a very detailed study of the water system to identify potential problems and determine (if possible) why an *E. coli* MCL violation has occurred and/or why total coliform bacteria have been found in our water system on multiple occasions.

ND: not detectable at testing limit

ppm: parts per million or milligrams per liter (mg/L)

ppb: parts per billion or micrograms per liter (µg/L)

ppt: parts per trillion or nanograms per liter (ng/L)

ppq: parts per quadrillion or picogram per liter (pg/L)

pCi/L: picocuries per liter (a measure of radiation)

Olivet University Form Revised March 2021

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include:

- Microbial contaminants, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- Inorganic contaminants, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- Pesticides and herbicides, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- Organic chemical contaminants, including synthetic and volatile organic chemicals, that are byproducts of
 industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff,
 agricultural application, and septic systems.
- Radioactive contaminants, that can be naturally-occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, the U.S. EPA and the State Board prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. The U.S. Food and Drug Administration regulations and California law also establish limits for contaminants in bottled water that provide the same protection for public health.

Tables 1, 2, 3, 4, 5, and 6 list all of the drinking water contaminants that were detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The State Board allows us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of the data, though representative of the water quality, are more than one year old. Any violation of an AL, MCL, MRDL, or TT is asterisked. Additional information regarding the violation is provided later in this report.

Microbiological Contaminants (complete if bacteria detected)	Highest No. of Detections	No. of Months in Violation	MCL	MCLG	Typical Source of Bacteria
Total Coliform Bacteria (state Total Coliform Rule)	(In a month)	0	1 positive monthly sample ^(a)	. 0	Naturally present in the environment
Fecal Coliform or E. coli (state Total Coliform Rule)	(In the year) 0	0	A routine sample and a repeat sample are total coliform positive, and one of these is also fecal coliform or <i>E. coli</i> positive		Human and animal fecal waste
E. coli (federal Revised Total Coliform Rule)	(In the year) 0	0	(b)	0	Human and animal fecal waste

(a) Two or more positive monthly samples is a violation of the MCL

(b) Routine and repeat samples are total coliform-positive and either is E. coli-positive or system fails to take repeat samples following E. coli-positive routine sample or system fails to analyze total coliform-positive repeat sample for E. coli.

TABLE 2 – SAMPLING RESULTS SHOWING THE DETECTION OF LEAD AND COPPER									
Lead and Copper (complete if lead or copper detected in the last sample set)	Sample Date	No. of Samples Collected	90 th Percentile Level Detected	No. Sites Exceeding AL	AL	PHG	No. of Schools Requesting Lead Sampling	Typical Source of Contaminant	
Lead (ppb)	7/08/ 2019	5	.004	0	15	0.2		Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits	
Copper (ppm)	7/0 8 / 2019	5	.027	O.	1.3	0.3	Not applicable	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives	

	TABLE 3	- SAMPLING I	RESULTS FOR	SODIUM	AND HARD	NESS
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Source of Contaminant
Sodium (ppm) Well 5	2/1/3/18	68		None	None	Salt present in the water and is generally naturally occurring
Sodium (ppm) Well 6	3/12/20	55	•	None	None	Salt present in the water and is generally naturally occurring
Hardness (ppm) Well 5	2/13/18	260	-	None	None	Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring
Hardness (ppm) Well 6	3/12/20	260	-	None	None	Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring
TABLE 4 – DET	ECTION O	F CONTAMIN	ANTS WITH A	<u>PRIMARY</u>	DRINKING	G WATER STANDARD
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL [MRDL]	PHG (MCLG) [MRDLG]	Typical Source of Contaminant
Gross Alpha (pCi/L) Well 5	11/8/17	1.375	ND-5.5	15	(0)	Erosion of natural deposits.
Gross Alpha (pCi/L) Well 6	2/2018 to 11/2018	5.7	4 -6.75	15	(0)	Erosion of natural deposits.
Total Radium (pCi/L) Well 6	2/2018 to 11/2018	0.855	0.647-1.14	3	0.05	Erosion of natural deposits
Fluoride (ppm) Well 5	2/13/18	0.21	-	2.0	(1)	Erosion of natural deposits.
Fluoride (ppm) Well 6	3/12/20	0.21	-	2.0	(1)	Erosion of natural deposits.
Aluminum Well 5 (ppm)	2/13/2018	0,690	•	1	0.6	Erosion of natural deposits; residue from some surface water treatment processes
Nickel Well #5 (ug/L)	2/13/2018	25	25	100	10	Erosion of natural deposits; discharge from metal factories
HAA5 (Halo Acetic Acids) Distribution	7/10/18	5.1	4.4-5.8	60	N/A	By-product of drinking water disinfection
TTHMs (Total Trihalomethanes) (ppb) Distribution	7/10/18	24.85	22.3-27.4	80	N/A	By-product of drinking water disinfection
TABLE 5 - DETY	CTION OF	CONTAMINA	<u>2</u> a fitiw 2ta	ECONDAR	Y DRINKI	IG WATER STANDARD
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	SMCL	PHG (MCLG)	Typical Source of Contaminant
Iron Well 6 (ppb)	2020 Quarterly	3.0	2.1-3.5	300	NA	Leaching from natural deposits
Manganese Well 6 (ppb)	2020 Quarterly	0.29	0.27-0.35	50	NA	Leaching from natural deposits
Iron (After Treatment) Well 6 (ppb)	Quarterly	0	0	300	NA	Leaching from natural deposits
Manganese (After Treatment) Well 6 (ppb)	2020 Quarterly	0	0	50	NA	Leaching from natural deposits
Chloride Well 6 (ppm)	3/12/20	26	-	500	NA	Runoff/leaching from natural deposits
Odor Well 6 (TON)	3/12/2020	1		3		Naturally-occurring organic materials
Color Well6 (UNITS)	3/12/2020	25		15		Naturally-occurring organic materials

Well 6 Potassium		3/12/20	6.7		1.0		None	
Magnesium		3/12/20	18	<u> </u>	1	.0	None	
otassium Vell 5		2/13/18	8.2	•	1.0		None	
Magnesium Vell 5			2/13/18 18		1.0		None	
Chemical or Constituent (and reporting units)		Sample Date	Level Detected	Range of Detections	p.	tion Level	Health Effects Language	
			6 – DETECTION	OF UNREGU	LATED CO	NTAMINA	NTS	
							experience liver and heart effects	
Well 5	(ppb)	,			1		containing nickel in excess of the MCL over many years may	
Nickel		2/13/18	25	-	100	12	Some people who drink water	
Aluminum Well 5	(ppb)	2/13/18	690	-	200	NA	Erosion of natural deposits; residua from some surface water treatment	
Odor Well 5 (1		3/12/2020	2		3		Naturally-occurring organic materials	
Well 5	(umhos/cm)			-		NA	water, seawater influence	
Well 5	(ppm) onductance	2/13/18	790		1600	NA	deposits Substances that form ions when in	
Well 5 Total Diss	(NTU) olved Solids	2/13/18	570		1000	NA	Runoff/leaching from natural	
Well 5 Turbidity	(ppm)	2/13/18	25		5	ŇA	deposits Soil runoff	
Sulfate		2/13/18	270	-	500	NA	Runoff/leaching from natural	
Chloride Well 5	(ppm)	2/13/18	38	-	500	NA	Runoff/leaching from natural deposits	
Manganes Well 5	se (ppb)	March 2019	239	-	50	NA	Leaching from natural deposits	
Iron Well 5	(ppb)	March 2019	7200	•	300	NA	Leaching from natural deposits	
Well 6	(ppb)	3/12/20	96	-	5000	NA .	Runoff/leaching from natural deposits;industrial wastes	
Well 6	(umhos/cm)			-			water, seawater influence	
Well 6	(ppm) onductance	3/12/20	750		1600	NA	deposits Substances that form ions when in	
Well 6 Total Diss	(NTU) olved Solids	3/12/20	530		1000	NA	Runoff/leaching from natural	
Turbidity	(ppm)	3/12/20	12	-	5	NΛ	deposits Soil runoff	
Sulfate Well 6	(nom)	3/12/20	250	-	500	NA	Runoff/leaching from natural	

Additional General Information on Drinking Water

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline (1-800-426-4791).

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. U.S. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline (1-800-426-4791).

Lead-Specific Language: If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service

lines and home plumbing. OLIVET UNIVERSITY is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. [OPTIONAL: If you do so, you may wish to collect the flushed water and reuse it for another beneficial purpose, such as watering plants.] If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline (1-800-426-4791) or at http://www.epa.gov/lead.

APPENDIX F: Certification Form (Suggested Format)

Consumer Confidence Report Certification Form

(to be submitted with a copy of the CCR)

(To certify electronic delivery of the CCR, use the certification form on the State Water Board's website at

http://www.swrcb.ca.gov/drinking_water/certlic/drinkingwater/CCR.shtml)

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Water Sy	stem	Name:	OLIVET UNIVERSITY					
Water Sy	/stem	Number:	3301863					
was distri availability contained	ibuted hav in th	d on3/2 ve been gi e report is	above hereby certifice 3/21 (date) to cover). Further, the correct and consisted State Water Resou	customers (and system certifer ont with the con	d app ies ti mplia	propriate notices of hat the information nce monitoring data		
Certified	by:	Name:	PAUL GORIN	0 0	•			
		Signature:	Tank k	Son				
	,	Title:	WATER OPE	RATOR				
	i	Phone Number:	(951) 663-09)51 [Date	3/23/21		
below by o	check was	ring all items	ery used and good-fails that apply and fill-in by mail or other direct	where appropr	iate:	,		
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This form is provided as a convenience for use to meet the certification requirement of the California Code of Regulations, section 64483(c).