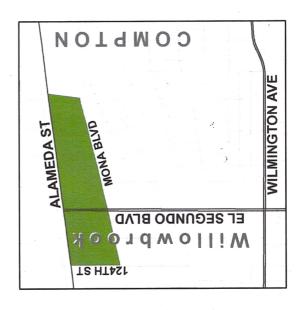
S018 CONSUMER CONFIDENCE REPORT



Este informe contiene información muy importante sobre su agua potable. Tradúzcalo ó hable con alguien que lo enteinda bien. Para obtener una copia en Español, llame a (310) 631-2625.

LYNWOOD PARK MUTUAL WATER COMPANY 2644 EAST 124TH STREET COMPTON, CALIFORNIA 90222

> LYNWOOD PARK MUTUAL WATER COMPANY 2018 CONSUMER CONFIDENCE REPORT

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Since 1991, California water utilities have been providing information on water served to its consumers. This report is a snapshot of the tap water quality that we provided last year. Included are details about where your water comes from, how it is tested, what is in it, and how it compares with state and federal limits. We strive to keep you informed about the quality of your water, and to provide a reliable and economic supply that meets all regulatory requirements.



Where Does My Tap Water Come From?

Your tap water comes from local, deep groundwater wells located within our service area. These wells supply the

area shown on the adjacent map. The quality of groundwater delivered to your home is presented in this report.

How is My Drinking Water Tested?

Your drinking water is tested regularly for unsafe levels of chemicals, radioactivity and bacteria at the source and in the distribution system. We test weekly, monthly, quarterly, annually or less often depending on the substance. State and federal laws allow us to test some substances less than once per year because their levels do not change frequently. All water quality tests are conducted by specially trained technicians in state-certified laboratories.

What Are Drinking Water Standards?

The U.S Environmental Protection Agency (USEPA) limits the amount of certain substances allowed in tap water. In California, the State Water Resources Control Board (State Board) regulates tap water quality by enforcing limits that are at least as stringent as the USEPA's. Historically, California limits are more stringent than the Federal ones.

There are two types of these limits, known as standards. Primary standards protect you from substances that could potentially affect your health. Secondary standards regulate substances that affect the aesthetic qualities of

water. Regulations set a Maximum Contaminant Level (MCL) for each of the primary and secondary standards. The MCL is the highest level of a substance that is allowed in your drinking water.

Public Health Goals (PHGs) are set by the California Environmental Protection Agency. PHGs provide more information on the quality of drinking water to customers, and are similar to their federal counterparts, Maximum Contaminant Level Goals (MCLGs). PHGs and MCLGs are advisory levels that are nonenforceable. Both PHGs and MCLGs are concentrations of a substance below which there are no known or expected health risks.

How Do I Read the Water Quality Table?

Although we test for over 100 substances, regulations require us to report only those found in your water. The first column of the water quality table lists substances detected in your water. The next columns list the average concentration and range of concentrations found in your drinking water. Following are columns that list the MCL and PHG or MCLG, if appropriate. The last column describes the likely sources of these substances in drinking water.

To review the quality of your drinking water, compare the highest concentration and the MCL. Check for substances greater than the MCL. Exceedence of a primary MCL does not usually constitute an immediate health threat. Rather, it requires testing the source water more frequently for a short duration. If test results show that the water continues to exceed the MCL, the water must be treated to remove the substance, or the source must be removed from service.

Why Do I See So Much Coverage in the News About the Quality Of Tap Water?

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include:

 Microbial contaminants, including viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife;

- Inorganic contaminants, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining or farming;
- Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses;
- Organic chemical contaminants, including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems;
- Radioactive contaminants, which can be naturally occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, the USEPA and the State Water Resources Control Board (State Board) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. The State Board regulations also establish limits for contaminants in bottled water that must provide the same protection for public health.

All drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA's Safe Drinking Water Hotline (1-800-426-4791). You can also get more information on tap water by logging on to these helpful web sites:

- http://www.epa.gov/dwstandardsregulations/2018drinking-water-standards-and-advisory-tables (USEPA's web site)
- https://www.waterboards.ca.gov/drinking_water/ certlic/drinkingwater/Chemicalcontaminants.html (State Board web site)

Lead: If present, elevated levels of lead can cause serious health problem, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with services lines and home plumbing. Lynwood Park Mutual Water Company is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe

Drinking Water Hotline or at http://www.epa.gov/lead.

Should I Take Additional Precautions?

Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. The USEPA/Centers for Disease Control guidelines on appropriate means to lessen the risk of infection of Cryptosporidium and other microbial contaminants are available from the USEPA's Safe Drinking Water Hotline (1-800-426-4791).

Source Water Assessment

The Lynwood Park Mutual Water Company conducted an assessment of its groundwater supplies in 2002. Groundwater supplies are considered most vulnerable to automobile body shops, automobile repair shops, junk/scrap/salvage yards, schools, leaking underground storage tanks, and high density septic systems. A copy of the approved assessment may be obtained by contacting Felix Contreras at (310) 631-2625.

How Can I Participate in Decisions On Water Issues That Affect Me?

The public is welcome to attend monthly meetings on the first Tuesday of each month at 2:00 p.m. and the annual meeting at 2644 East 124th Street in Compton, CA 90222.

How Do I Contact My Water Agency If I Have Any Questions About Water Quality?

If you have specific questions about your tap water quality, please contact Felix Contreras at (310) 631-2625.

Some Helpful Water Conservation Tips

- Fix leaky faucets in your home save up to 20 gallons every day for every leak stopped
- Save between 15 and 50 gallons each time by only washing full loads of laundry
- Adjust your sprinklers so that water lands on your lawn/garden, not the sidewalk/driveway – save 500 gallons per month
- Use organic mulch around plants to reduce evaporation

 save hundreds of gallons a year
- Visit http://www.epa.gov/watersense for more information.

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Results are from the most recent testing performed in accordance with state and federal drinking water regulations

The State allows monitoring for some contaminants less than once per year because the concentrations of these contaminants do not change frequently.

Some of the data, though representative, are more than one year old

PRIMARY STANDARDS MONITORED AT THE SOL		IRCE-MANDATED FOR PUBLIC HEALTH	JBLIC HEAL	H	
ORGANIC		GROUNDWATER	PRIMARY	MCLG	MAJOR SOURCES IN DRINKING WATER
CHEMICALS (ug/l)	AVERAGE	RANGE	MCL	or PHG	
Tetrachloroethylene (PCE) (a)	3.90	2.6 - 5.4	5	90.0	Discharge from metal degreasing sites and other factories. Some people who drink water containing tetrachloroethylene in excess of the MCL over many years may experience liver, kidney or immune system problems.
INORGANICS Sampled from 2016-2018 (b)	(g) (g)				
	1.1	ND - 3.3	10	0.004	Erosion of natural deposits; glass/electronics production wastes; runoff
Barium (mg/l)	0.12	0.12	1	2 (c)	Oil drilling waste and metal refinery discharge; erosion of natural deposits
Fluoride (mg/l)	0.36	0.34 - 0.38	2.0	1 (c)	Erosion of natural deposits, water additive that promotes strong teeth
Nitrate (mg/l as N)	1.5	1.1 - 1.9	10	10 (c)	Runoff and leaching from fertilizer use/septic tanks/sewage, natural erosion
RADIOLOGICAL - (pCi/l) (Results are from 2016-2018) (b)	3-2018) (b) (d)				
Gross Alpha	6.3	5.7 - 6.9	15 (e)	0	Erosion of natural deposits
Radium 226	90.0	ND - 0.12	(P) Y	0.05	Erosion of natural deposits
Radium 228	0.02	ND - 0.06	(n) c	0.019	Erosion of natural deposits
Uranium	4.5	4.2 - 5.0	20 (e)	0.43 (c)	0.43 (c) Erosion of natural deposits
PRIMARY STANDARDS MONITORED IN THE DIS	TORED IN THE DIST	TRIBUTION SYSTEM - MANDATED FOR PUBLIC HEALTH	MANDATED	FOR P	ЛВ LIC НЕАLTH
	DISTRIBU	DISTRIBUTION SYSTEM	PRIMARY	MCLG	•
MICROBIALS	AVERAGE # POSITIVE	RANGE OF # POSITIVE	MCL	or PHG	
					Motivably proposal in the conference Coliforms are broducin that are notinally proposal in the

PRIMARY STANDARDS MONITORED IN THE DIST	TORED IN THE DIST	TRIBUTION SYSTEM - MANDATED FOR PUBLIC HEALTH	MANDATED	FOR PI	BLIC HEALTH
	DISTRIBU	TION SYSTEM	PRIMARY	MCLG	•
MICROBIALS	AVERAGE # POSITIVE	RANGE OF # POSITIVE	MCL	or PHG	
Total Coliform Bacteria	0	0 - 3	> 1 positive	0	Naturally present in the environment. Coliforms are bacteria that are naturally present in the environment and are used as an indicator that other, potentially-harmful, bacteria may be present. Coliforms were found in more samples than allowed and this was a warning of potential problems.
Fecal Coliform and E. Coli Bacteria	0	0 - 1	0	0	Human and animal fecal waste
No. of Acute Violations	0	0	-		
DISINFECTION BY-PRODUCTS (f)	DISTRIBU	ITION SYSTEM	PRIMARY	MCLG	
AND DISINFECTION RESIDUALS	HIGHEST RUNNING ANNUAL AVERAGE	RANGE	MCL or [MRDL]	or PHG	
Total Trihalomethanes-TTHMS (µg/l)	1.3	ND - 5.0	80	-	By-product of drinking water chlorination
Haloacetic Acids (µg/l)	0.3	ND - 1.2	09		By-product of drinking water disinfection
Total Chlorine Residual (mg/l)	1.9	ND - 5.6	4.0 (g)	4.0 (h)	Drinking water disinfectant added for treatment. Some people who use water containing chlorine in excess of the MRDL coulod experience irritating effects to their eyes and nose. Some people who drink water containing chlorine well in excess of the MRDL could experience stomach discomfort.
AT THE TAP	DISTRIBU	ITION SYSTEM			
PHYSICAL CONSTITUENTS 10 sites sampled in 2016	90%ile	# OF SITES ABOVE THE AL	ACTION LEVEL AL	MCLG or PHG	
Copper (mg/l)	0.01 (i)	0	1.3 AL	0.3 (c)	Internal corrosion of household plumbing, erosion of natural deposits
Lead (µg/l)	(j) QN	0	15 AL	0.2 (c)	0.2 (c) Internal corrosion of household plumbing, industrial manufacturer discharges

SECONDARY STANDARDS MO	NITORED AT THE S	OURCE-FOR AESTHETIC PURPOSES	IC PURPOSE	5	
Sampled from 2018,2018 (h)	GROU	JNDWATER	SECONDARY	MCLG	
	AVERAGE	RANGE	MCL	or PHG	
A constitution of the control of the	12.3	12.2 - 12.4	Non-corrosive		Natural/industrially-influenced balance of hydrogen/carbon/oxygen in water
Aggressiveness mack (concisions)	44.7	43.0 - 46.0	200		Runoff/leaching from natural deposits, seawater influence
Chionde (mg/l)	1	_	3		Naturally-occurring organic materials
Odor (threshold odor number)	740	700 - 730	1,600		Substances that form ions when in water, seawater influence
Specific Conductance (uS/cm)	00	97 - 100	200		Runoff/leaching from natural deposits, industrial wastes
Sulfate (mg/l)	25	440 - 460	1,000		Runoff/leaching from natural deposits
Total Dissolved Solids (Ing/l)	CN CN	QN	5		Soil runoff
I urbidity (NI U)					

: PURPOSES			Naturally-occurring organic materials	Naturally-occurring organic materials		Soil runoit
HETIC	MCLG	or PHG	ŀ			
-FOR AEST	SECONDARY	MCL	15	6		
ITORED IN THE DISTRIBUTION SYSTEM-FOR AESTHETIC PURPOSES	DISTRIBUTION SYSTEM	RANGE	<3 - 7.5	10.30	0.5 - 0.1	0.1 - 4.7
	6	AVERAGE	8		_	0.3
SECONDARY STANDARDS MON	CENEDAL	GENERAL CONSTITUENTS	PHYSICAL CONSTITUTION S	Color (color units)	Odor (threshold odor number)	Turbidity (NTU)

FOOTNOTES

ADDITIONAL CHEMICALS OF INTEREST	INTEREST	
Sampled from 2016-2018 (b)		GROUNDWATER
	AVERAGE	RANGE
Alkalinity (Total as CACO3) (mg/l)	180.0	180.0
Bicarbonate Alkalinity (ma/l)	220.0	220.0
Calcium (mg/l)	82.3	78.0 - 86.0
1,4-Dioxane (ug/l) (j)	3.2	2.7 - 3.7
Magnesium (mg/l)	17.3	17.0 - 18.0
pH (standard unit)	7.7	7.6 - 7.8
Potassium (mg/l)	3.4	3.0 - 3.6
Sodium (ma/l)	37.7	37.0 - 38.0
Total Hardness (mg/l)	273.3	270 - 280

(a) California	(a) California Public Health Goal (PHG). Other advisory levels listed in this column are
federal Maxim	federal Maximum Contaminant Level Goals (MCLGs).
(b) Indicates of	(b) Indicates dates sampled for groundwater sources only.
(c) California	(c) California Public Health Goal (PHG). Other advisory levels listed in this column are
federal Maxin	federal Maximum Contaminant Level Goals (MCLGs).
(d) Combined	(d) Combined Radium 226 + Radium 228 has a Maximum Contaminant Level (MCL) of 5 pCi/L. For
individual Rac	individual Rad 226 and Rad 228, the PHG is 0.05 pCi/L for Rad 226 & 0.019 pCi/L for Rad 228.
(e) MCL comp	(e) MCL compliance based on 4 consecutive quarters of sampling.
(f) Running ar	(f) Running annual average used to calculate average, range, and MCL compliance.
(g) Maximum	(g) Maximum Residual Disinfectant Level (MRDL)
(h) Maximum	(h) Maximum Residual Disinfectant Level Goal (MRDLG)
(i) 90th percer	(i) 90th percentile from the most recent sampling at selected customer taps.
(j) The Notific	(j) The Notification Level of 1 ug/l for 1,4-Dioxane was exceeded in three wells in 2018. Some
people who u	people who use water containing 1,4-dioxane in excess of the Notification Level over many years
may experien	may experience liver or kidney problems and may have an increased risk of getting cancer, based
on studies in	on studies in laboratory animals.

	pG// = picoCuries per liter	ND = constituent not detected at the reporting limit NTU = nephelometric turbidity units	uS/cm = microSiemens per centimeter	μg/l = micrograms per liter or parts per billion (equivalent to 1 drop in 42,000 gallons)
	SI = saturation index	ND = constituent not detected at	to 1 drop in 42 gallons)	o 1 drop in 42,000,000 gallons)
ABBREVIATIONS	< = less than	NA = constituent not analyzed	ma/l = milliorams ner liter or parts per million (equivalent to 1	ng/l = nanograms per liter or parts per trillion (equivalent to 1

umum Contaminant Level (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically

Maximum Contaminant Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency.

Maximum Residual Disinfectant Level (MRDL): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water. disinfectant to control microbial contaminants.

ublic Health Goal (PHG): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency. Varification Level: The level at which notification of the public water system governing body is required. A health-based advisory level for an unregulated contaminant. reatment Technique (TT): A required process intended to reduce the level of a contaminant in drinking water.

Secondary Drinking Water Standards (SDWS): MCLs and MRDLs for contaminants that affect the aesthetic qualities (taste, odor, or appearance) of drinking water. Contaminants with SDWSs do not affect the health at the MCL levels. Primary Drinking Water Standards (PDWS): MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements. legulatory Action Level (AL): The concentration of a contaminant which, ir exceeded, triggers treatment or other requirements that a water system must follow. Variances & Exemptions: Department permission to exceed an MCL or not comply with a treatment technique under certain conditions