2022 Consumer Confidence Report
Water System Name: SUSAN RIVER PARK WATER CO CA1800503 Report Date: June 26, 2023
We test the drinking water quality for many constituents as required by state and federal regulations. This report shows the results of our monitoring for the period of January 1 to December 31, 2022 and may include earlier monitoring data.
Language in Spanish: Este informe contiene información muy importante sobre su agua para beber. Favor de comunicarse <u>SUSAN_RIVER_PARK_WATER_COMPANY</u> a (530) 249-7253 para asistirlo en español.
Type of water source(s) in use: Groundwater
Name & general location of source(s): Well 01 (Daily use), Well 02 (Standby for fire suppression, not used in 2022) Bangham Lane, Johnstonville CA 96130
Drinking Water Source Assessment information:
These Assessments were done using the Default Groundwater System Method.
An assessment for WELL 01 was done in April 2003. Well 01 - is considered most vulnerable to the following activities not associated with any detected contaminants:
Septic systems - high density [>1/acre]
An assessment for WELL 02 (Standby for Fire Suppression) was conducted in April 2003. Well 02 - is considered most vulnerable to the following activities not associated with any detected contaminants:
Septic systems - high density [>1/acre]
Discussion of Vulnerability:
Due to the detection of Arsenic, Well 01 is considered vulnerable to activities that may have contributed to or caused the release of Arsenic. In particular, Arsenic is believed to be associated with runoff from orchards, glass and electronics production wastes and the erosion of natural deposits. Arsenic was detected for Well 01 before July 2000 with results reaching up to 4.0 ug/L compared to the MCL of 50.0 ug/L. Arsenic was detected for Well 01 in 2011 with results of 4.0 ug/L and in 2020 with results of 4.0 ug/L compared to the current MCL of 10.0 ug/L
Due to the detection of Arsenic, Well 02 is considered vulnerable to activities that may have contributed to or caused the release of Arsenic. In particular, Arsenic is believed to be associated with runoff from orchards, glass and electronics production wastes and the erosion of natural deposits. Arsenic was detected for Well 02 before December 1999 with results reaching up to 3.0 ug/L compared to the MCL of 50.0 ug/L. Arsenic was detected for Well 02 in 2007 with results of 2.0 ug/L and in 2022 with results of 3.0 ug/L compared to the current MCL of 10.0 ug/L
Acquiring Information:
A copy of the assessment may be viewed at Susan River Park Water Co., 30 Roop St., Susanville, CA 96130.
You may request a summary of the assessment by contacting: Steve Watson, District Engineer, Lassen District, Division of Drinking Water 364 Knollcrest Drive, Suite 101 Redding CA 96002 530.224.4800 (phone); 530.224.4844 (fax)
Time and place of regularly scheduled board meetings for public participation: Regularly-scheduled Susan River Park Water Company water user meetings are not held. The State Water Resources Control Board and the California Public Utilities Commission may offer other opportunities

NSWT 2022 CCR Form April 2023

Phone:

(530) 249-7253

For more information, contact: _ Mike Herman

TERMS USED IN THIS REPORT

Level 1 Assessment: A Level 1 assessment is a study of the water system to identify potential problems and determine (if possible) why total coliform bacteria have been found in our water system.

Level 2 Assessment: A Level 2 assessment is a very detailed study of the water system to identify potential problems and determine (if possible) why an *E. coli* MCL violation has occurred and/or why total coliform bacteria have been found in our water system on multiple occasions.

Maximum Contaminant Level (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

Maximum Contaminant Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (U.S. EPA).

Maximum Residual Disinfectant Level (MRDL): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

Primary Drinking Water Standards (PDWS): MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

Public Health Goal (PHG): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

Regulatory Action Level (AL): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

Secondary Drinking Water Standards (SDWS): MCLs for contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.

Treatment Technique (TT): A required process intended to reduce the level of a contaminant in drinking water.

Variances and Exemptions: Permissions from the State Water Resources Control Board (State Board) to exceed an MCL or not comply with a treatment technique under certain conditions.

ND: not detectable at testing limit

ppm: parts per million or milligrams per liter (mg/L)

ppb: parts per billion or micrograms per liter (μg/L)

ppt: parts per trillion or nanograms per liter (ng/L)

ppq: parts per quadrillion or picogram per liter (pg/L)

pCi/L: picocuries per liter (a measure of radiation)

Sources of Drinking Water and Contaminants that May Be Present in Source Water: The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include:

- *Microbial contaminants*, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- *Inorganic contaminants*, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- Pesticides and herbicides, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- *Organic chemical contaminants*, including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems.
- Radioactive contaminants, that can be naturally-occurring or be the result of oil and gas production and mining activities.

Regulation of Drinking Water and Bottled Water Quality: In order to ensure that tap water is safe to drink, the U.S. EPA and the State Board prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. The U.S. Food and Drug Administration regulations and California law also establish limits for contaminants in bottled water that provide the same protection for public health.

About Your Drinking Water Quality - Drinking Water Contaminants Detected: Tables 1, 2, 3, 4, 5 and 6 list all of the drinking water contaminants that were detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The State Board allows us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of the data, though representative of the water quality, are more than one year old. Any violation of an AL, MCL, MRDL, or TT is asterisked. Additional information regarding the violation is provided later in this report.

TABLE 1 – SAMPLING RESULTS SHOWING THE DETECTION OF COLIFORM BACTERIA								
Microbiological Contaminants	Highest Number of Detections	No. of months in violation	MCL	MCLG	Typical Source of Bacteria			
E. Coli	(in the year) 0	0	(a)	0	Human and animal fecal waste			

(a) Routine and repeat samples are total coliform-positive and either is E. coli-positive or system fails to take repeat samples following E. coli-positive routine sample or system fails to analyze total coliform-positive repeat sample for E. coli.

	TABLE 2 – SAMPLING RESULTS SHOWING THE DETECTION OF LEAD AND COPPER									
Lead and Copper	Sample Date	No. of Samples Collected	90 th Percentile Level Detected	No. Sites Exceeding AL	AL	PHG	No. of Schools Requesting Lead Sampling	Typical Source of Contaminant		
Lead (ppb)	(2021)	5	ND	0	15	0.2	0	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits		
Copper (ppm)	(2021)	5	ND	0	1.3	0.3	Not applicable	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives		

	TABLE 3	– SAMPLING I	RESULTS FO	R SODIUN	I AND HAI	RDNESS
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Source of Contaminant
Sodium (ppm)	(2020)	21	n/a	None	None	Salt present in the water and is generally naturally occurring
Hardness (ppm)	(2020)	101	n/a	None	None	Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring
TABLE 4 – DET	ECTION C	F CONTAMIN	ANTS WITH	A <u>PRIMAI</u>	<u>ry</u> drinki	NG WATER STANDARD
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL [MRDL]	PHG (MCLG) [MRDLG]	Typical Source of Contaminant
Arsenic (ug/L)	(2020)	4	n/a	10	0.004	Erosion of natural deposits; runoff from orchards; glass and electronics production wastes
Fluoride (mg/L)	(2020)	0.1	n/a	2.0	1	Erosion of natural deposits; water additive which promotes strong teeth; discharge from fertilizer and aluminum factories
Gross Alpha (pCi/L)	(2020)	2.21	n/a	15	(0)	Erosion of natural deposits
TABLE 5 – DETE	CTION OF	CONTAMINA	NTS WITH A	SECONDA	ARY DRINI	KING WATER STANDARD
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	SMCL	PHG (MCLG)	Typical Source of Contaminant
Chloride (mg/L)	(2020)	4	n/a	500	n/a	Runoff/leaching from natural deposits; seawater influence
<mark>Manganese (</mark> μg/L)	(2020)	150*	<mark>n/a</mark>	50	<mark>n/a</mark>	Leaching from natural deposits
Specific Conductance (µS/cm)	(2020)	294	n/a	1,600	n/a	Substances that form ions when in water; seawater influence
Sulfate (mg/L)	(2020)	15	n/a	500	n/a	Runoff/leaching from natural deposits; industrial wastes
Total Dissolved Solids (mg/L)	(2020)	170	n/a	1000	n/a	Runoff/leaching from natural deposits

TABLE 6 – DETECTION OF UNREGULATED CONTAMINANTS with NOTIFICATION LEVELS

NONE DETECTED

ADDITIONAL DETECTIONS								
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	Notification Level	Typical Source of Contaminant			
Calcium (mg/L)	(2020)	29	n/a	n/a	n/a			
Magnesium (mg/L)	(2020)	7	n/a	n/a	n/a			
pH (units)	(2020)	7.8	n/a	n/a	n/a			
Alkalinity (Total) mg/L	(2020)	120	n/a	n/a	n/a			
Aggressiveness Index	(2020)	11.7	n/a	n/a	n/a			
Langelier Index	(2020)	-0.07	n/a	n/a	n/a			

Additional General Information on Drinking Water

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline (1-800-426-4791).

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. U.S. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline (1-800-426-4791).

Lead-Specific Language: If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. <u>SUSAN RIVER PARK WATER COMPANY</u> is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you do so, you may wish to collect the flushed water and reuse it for another beneficial purpose, such as watering plants. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline (1-800-426-4791) or at http://www.epa.gov/lead.

About our Manganese: Manganese was found at levels that exceed the secondary MCL of 50 μg/L. The manganese MCL was set to protect you against unpleasant aesthetic effects (e.g., color, taste, and odor) and the staining of plumbing fixtures (e.g., tubs and sinks) and clothing while washing. The high manganese levels are due to leaching of natural deposits.

Source Water Protection Tips for Consumers

Protection of drinking water is everyone's responsibility. You can help protect your community's drinking water source in several ways:

- Eliminate excess use of lawn and garden fertilizers and pesticides they contain hazardous chemicals that can reach your drinking water source.
- Pick up after your pets.
- If you have your own septic system, properly maintain your system to reduce leaching to water sources or consider connecting to a public water system.
- Dispose of chemicals properly; take used motor oil to a recycling center.

Water Conservation Tips for Consumers

Did you know that the average U.S. household uses approximately 400 gallons of water per day or 100 gallons per person per day? Luckily, there are many low-cost and no-cost ways to conserve water. Small changes can make a big difference – try one today and soon it will become second nature.

- Take short showers a 5 minutes shower uses 4 to 5 gallons of water compared to up to 50 gallons for a bath.
- Shut off water while brushing your teeth, washing your hair, and shaving and save up to 500 gallons a month.
- Water-efficient showerheads are inexpensive, easy to install, and can save you up to 750 gallons a month.
- Run your clothes washer and dishwasher only when they are full. You can save up to 1,000 gallons a month.
- Fix leaking toilets and faucets. Faucet washers are inexpensive and take only a few minutes to replace. To check your toilet for a leak, place a few drops of food coloring in the tank and wait. If it seeps into the toilet bowl without flushing, you have a leak. Fixing it or replacing it with a new, more efficient model can save up to 1,000 gallons a month.
- Adjust sprinklers so only your lawn is watered. Apply water only as fast as the soil can absorb it and during the cooler parts of the day to reduce evaporation.