## **2019** Consumer Confidence Report

Water System Name: **KETTLEMAN CITY CSD** Report Date 2019

We test the drinking water quality for many constituents as required by state and federal regulations. This report shows the results of our monitoring for the period of January 1 to December 31, 2019 and may include earlier monitoring data.

Este informe contiene información muy importante sobre su agua para beber. Favor de comunicarse Kettleman City CSD a 559-386-5866 para asistirlo en español.

Type of water source(s) in use: GROUND WATER 1-10/19, SURFACE WATER TREATMENT 11-12/2019

Name & general location of source(s): Well 2A (Maud) is located in the southwest portion of the residential area adjacent to the District's office. Well 3 (Becky Pease) is located in the southeast portion of the residential area near the

Kings County Fire Station. Both wells became Standby wells once the Surface Water Treatment Plant became active Surface Water Treatment Plant (SWTP) is located on the California Aqueduct. Aqueduct water is recognized as moderately difficult to treat due to the low to moderate hardness, low alkalinity, rapidly variable temperature, and low to moderate turbidity. It has been observed that this water changes noticeably from the normal characteristics, and that these changes happen very rapidly and without warning. Temperature changes and pH variations of 7.5 to 9.5 are fairly commonplace. For a short duration, turbidity has been known to increase by as much as 300 percent. This water also experiences periodic algae growth. Reductions to agricultural allocations have greatly reduced flows through the aqueduct, combined with low reservoir levels, raw water has become increasingly difficult to treat.

Time and place of regularly scheduled board meetings for public participation:

Third Tuesday at 6PM at 110 General Petroleum Ave. Kettleman City, CA 93239

For more information, contact:

Rosa Maldonado

### Phone: (559

#### (559)386-5866

#### TERMS USED IN THIS REPORT

Maximum Contaminant Level (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

Maximum Contaminant Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (U.S. EPA).

**Public Health Goal (PHG)**: The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

**Maximum Residual Disinfectant Level (MRDL)**: The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

**Primary Drinking Water Standards (PDWS)**: MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

**Secondary Drinking Water Standards (SDWS)**: MCLs for contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.

**Treatment Technique (TT)**: A required process intended to reduce the level of a contaminant in drinking water.

**Regulatory Action Level (AL)**: The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**Variances and Exemptions**: Permissions from the State Water Resources Control Board (State Board) to exceed an MCL or not comply with a treatment technique under certain conditions.

**Level 1 Assessment**: A Level 1 assessment is a study of the water system to identify potential problems and determine (if possible) why total coliform bacteria have been found in our water system.

**Level 2 Assessment**: A Level 2 assessment is a very detailed study of the water system to identify potential problems and determine (if possible) why an *E. coli* MCL violation has occurred and/or why total coliform bacteria have been found in our water system on multiple occasions.

**ND**: not detectable at testing limit

ppm: parts per million or milligrams per liter (mg/L)

**ppb**: parts per billion or micrograms per liter (μg/L)

ppt: parts per trillion or nanograms per liter (ng/L)

ppq: parts per quadrillion or picogram per liter (pg/L)

pCi/L: picocuries per liter (a measure of radiation)

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

#### Contaminants that may be present in source water include:

- *Microbial contaminants*, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- *Inorganic contaminants*, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- *Pesticides and herbicides*, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- Organic chemical contaminants, including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems.
- Radioactive contaminants, that can be naturally-occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, the U.S. EPA and the State Board prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. The U.S. Food and Drug Administration regulations and California law also establish limits for contaminants in bottled water that provide the same protection for public health.

Tables 1, 2, 3, 4, 5, and 6 list all of the drinking water contaminants that were detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The State Board allows us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of the data, though representative of the water quality, are more than one year old. Any violation of an AL, MCL, MRDL, or TT is asterisked. Additional information regarding the violation is provided later in this report.

TABLE 1 – SAMPLING RESULTS SHOWING THE DETECTION OF COLIFORM BACTERIA						
Microbiological Contaminants (complete if bacteria detected)	Highest No. of Detections	No. of Months in Violation	MCL	MCLG	Typical Source of Bacteria	
Total Coliform Bacteria (state Total Coliform Rule)	0	0	1 positive monthly sample <sup>(a)</sup>	0	Naturally present in the environment	
Fecal Coliform or <i>E. coli</i> (state Total Coliform Rule)	0	0	A routine sample and a repeat sample are total coliform positive, and one of these is also fecal coliform or <i>E. coli</i> positive		Human and animal fecal waste	
E. coli (federal Revised Total Coliform Rule)	09-25-19	0	<b>(b)</b>	0	Human and animal fecal waste	

<sup>(</sup>a) Two or more positive monthly samples is a violation of the MCL

(b) Routine and repeat samples are total coliform-positive and either is *E. coli*-positive or system fails to take repeat samples following *E. coli*-positive routine sample or system fails to analyze total coliform-positive repeat sample for *E. coli*.

TABLE 2 – SAMPLING RESULTS SHOWING THE DETECTION OF LEAD AND COPPER								
Lead and Copper (complete if lead or copper detected in the last sample set)	Sample Date	No. of Samples Collected	90 <sup>th</sup> Percentile Level Detected	No. Sites Exceeding AL	AL	PHG	No. of Schools Requesting Lead Sampling	Typical Source of Contaminant
Lead (ppb)	9-25-19	10	0	0	15	0.2	0	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits
Copper (ppm)	9-25-19	10	0	0	1.3	0.3	Not applicable	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives

TABLE 3 – SAMPLING RESULTS FOR SODIUM AND HARDNESS						
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Source of Contaminant
Sodium (ppm) well	2017	275	240-310	None	None	Salt present in the water and is generally naturally occurring
Sodium (ppm) SWTP	10-23-19	55	NA	None	None	Salt present in the water and is generally naturally occurring
Hardness (ppm) well	2017	175	170-180	None	None	Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring
Hardness (ppm) SWTP	10-23-19	100	NA	None	None	Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring

		TA	BLE 4 – DETE	CTION		
TABLE 4 – DET	TABLE 4 – DETECTION OF CONTAMINANTS WITH A <u>PRIMARY</u> DRINKING WATER STANDARD					
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL [MRDL]	PHG (MCLG) [MRDLG]	Typical Source of Contaminant
Inorganic Contaminants						
Arsenic (ppb)*well	2018	12.75	4.9-17	10	0.004	Erosions of natural deposits;
						runoff from orchards; glass and electronics production wastes.
Arsenic (ppb) SWTP	10-23-19	2.6	NA	10	0.004	Erosion of natural deposits; runoff form orchards; glass and electronics production wastes.
Fluoride (ppm)well	2017	0.065	ND-0.13	2.0	1	Erosion of natural deposits; water additive which promotes strongt eeth; discharge from fertilizer and aluminum factories
Selenium (ppb) well	2017	3.4	ND-6.8	50	30	Discharge from petroleum, glass and metal refineries; erosion of natural deposits; discharge from mines and chemical manufacturers; runoff from livestock lots (feed additives).
Copper (mg/L) well	2019	6.9	0-69	AL-1.3	0.3	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservative
Gross Alpha (pCi/L) SWTP	10-23-19	2.04	NA	15	0	Erosion of natural deposits
Radium 228(pCi/L) SWTP	10-23-19	0.243	NA	5	NA	Erosion of natural deposits
Volatile Organic Contaminar	nts					
Before Benzene treatment (ppb)*well	2019	15.75	20-80	1	0.15	Discharge from plastics, dyes and nylon factories, leaching from gas storage tanks and landfills
Treated Benzene (ppb) well	2019	0	NA	1	0.15	Discharge from plastics, dyes and nylon factories; leaching from gas storage tanks and landfills.
1.2 Dichloroethane (ng/L)well	2019	0.013	0-0.53	500	400	Discharge from industrial chemical factories
Disinfection Byproducts, Dis	sinfectant resid	uals, and Disinfect	ion Byproduct Pre	cursors		
Haloacetic Acids HAA5( (ug/L)	2019	10.5	0-21	60	N/A	Byproduct of drinking water disinfection
Total Trihalomethanes (ug/L)	2019	40.83	0.66-81	80	N/A	Byproduct of drinking water disinfection
TABLE 5 – DETI	ECTION OF	CONTAMINA	NTS WITH A S	SECONDAR	<u>Y</u> DRINKI	NG WATER STANDARD
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	SMCL	PHG (MCLG)	Typical Source of Contaminant
Total Dissolved Solids (TDS) (ppm) well	2019	1873	1800-2000	1000	N/A	Runoff/leaching from natural deposits.
Total Dissolved Solds (TDS) (ppm) SWTP	10-23-2019	250	NA	1000	NA N/A	Runoff/leaching from natural deposits
Manganese (ppb)*well	2019	115.3	81-130	50	N/A	Leaching from natural deposits.
Manganese (ppb) SWTP	10-23-19	34	NA	50	NA	Leaching from natural deposits
Odor – Threshold (Units) SWTP	10-23-19	1.5	NA	3	N/A	Naturally- occurring organic materials.

Turbidity (NTU)SWTP	10-23-19	1.4	NA	5	N/A	Soil runoff.
Color (units) SWTP	10-23-19	25	NA	15	NA	Naturally-occurring organic materials
Specific Conductance (EC) (uhmos/cm)* well	2019	1873	1800-200	1600	N/A	Substances that form ions when in water; seawater influence.
Specific Conductance (EC) (umhos/cm) SWTP	10-23-19	480	NA	1600	NA	Substances that form ions when in water; seawater influence
Chloride (ppm) SWTP	10-23-19	78	NA130-450	500	N/A	Runoff/leaching from natural deposits; seawater influence.
Sulfate (ppm) SWTP	10-23-19	30	NA	500	N/A	Runoff/leaching from natural deposits; industrial wastes.

#### **Additional General Information on Drinking Water**

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline (1-800-426-4791).

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. U.S. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline (1-800-426-4791).

Lead-Specific Language: If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. KETTLEMAN CITY CSD is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline (1-800-426-4791) or at <a href="http://www.epa.gov/lead">http://www.epa.gov/lead</a>.

# Summary Information for Violation of a MCL, MRDL, AL, TT, or Monitoring and Reporting Requirement

VIOLATION OF A MCL, MRDL, AL, TT, OR MONITORING AND REPORTING REQUIREMENT						
Violation	Explanation	Duration	Actions Taken to Correct the Violation	Health Effects Language		
BENZENE/ MANGANESE/ SPECIFIC CONDUCTANCE	Poor quality ground water	Through 11-2019	A new membrane surface water treatment plant (SWTP) was built using State Water Bond Funds. The public was no longer served	Some people who drink water containing these constituents in excess of the MCL over many years may experience changes in their		

	water from the wells after 11- 2019	overall health and may be at increased risk of getting cancer.
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## For Water Systems Providing Groundwater as a Source of Drinking Water

TABLE 7 – SAMPLING RESULTS SHOWING FECAL INDICATOR-POSITIVE GROUNDWATER SOURCE SAMPLES							
Microbiological Contaminants (complete if fecal-indicator detected)	Total No. of Detections	Sample Dates	MCL [MRDL]	PHG (MCLG) [MRDLG]	Typical Source of Contaminant		
E. coli	0	2019	0	(0)	Human and animal fecal waste		
Enterococci	0	2019	TT	N/A	Human and animal fecal waste		
Coliphage	0	2019	TT	N/A	Human and animal fecal waste		

## For Systems Providing Surface Water as a Source of Drinking Water

TABLE 8 - SAMPLING RESULTS SHOWING TREATMENT OF SURFACE WATER SOURCES				
Treatment Technique (a) (Type of approved filtration technology used)	Membrane			
	Turbidity of the filtered water must:			
Turbidity Performance Standards (b)	1 – Be less than or equal to0.1 NTU in 95% of measurements in a month.			
(that must be met through the water treatment process)	2 – Not exceed1.0 NTU for more than eight consecutive hours.			
	3 – Not exceed _1.0 NTU at any time.			
Lowest monthly percentage of samples that met Turbidity Performance Standard No. 1.	Reporting			
Highest single turbidity measurement during the year				
Number of violations of any surface water treatment requirements	0			

<sup>(</sup>a) A required process intended to reduce the level of a contaminant in drinking water.

<sup>(</sup>b) Turbidity (measured in NTU) is a measurement of the cloudiness of water and is a good indicator of water quality and filtration performance. Turbidity results which meet performance standards are considered to be in compliance with filtration requirements.