2024 Consumer Confidence Report

Water System Name: FEDEX GROUND CA0409194 Report Date: August 2025

We test the drinking water quality for many constituents as required by state and federal regulations. This report shows the results of our monitoring for the period of January 1 to December 31, 2024 and may include earlier monitoring data.

Language in Spanish: Este informe contiene información muy importante sobre su agua para beber. Favor de comunicarse FEDEX GROUND a (530) 332-2000 para asistirlo en español.

Type of water source(s) in use: Groundwater

Name & general location of source(s): Well 01

101 Book Farm Road, Durham, CA 95938

Drinking Water Source Assessment information: A Drinking Water Source Assessment for WELL 01 of the FEDEX GROUND water system was conducted in May 2018. A complete assessment summary is not available for the water source.

Discussion of Vulnerability:

Well 01 is considered vulnerable to Possible Contaminating Activities (PCA). There are NPDES/WDR permitted discharges occurring nearby. Septic systems - low density (<1/acre). Other PCAs are associated with fleet/truck parking terminals, machine shops, vehicle body/repair shops and diesel storage tanks. There are also PCAs associated with cattle grazing and non-irrigated or drip-irrigated crop farming that may impact this well. There are transportation corridors (Highway 99 and Durham Pentz-Road). There are also ephemeral streams nearby.

Acquiring Information:

For more info you may visit https://www.waterboards.ca.gov/drinking_water/certlic/drinkingwater/DWSAP.html or contact: Butte County Environmental Health Program Manager 530-552-3880, 530-538-2165 (fax)

Time and place of regularly scheduled board meetings for public participation: Regularly-scheduled Butte County

meetings are held. The State Water Resources Control Board may offer other opportunities.

For more information, contact: Jesse Arensdorf <u>jesse.arensdorf@fedex.com</u> Phone: (530) 332-2000 ext. 2002

TERMS USED IN THIS REPORT

Level 1 Assessment: A Level 1 assessment is a study of the water system to identify potential problems and determine (if possible) why total coliform bacteria have been found in our water system.

Level 2 Assessment: A Level 2 assessment is a very detailed study of the water system to identify potential problems and determine (if possible) why an *E. coli* MCL violation has occurred and/or why total coliform bacteria have been found in our water system on multiple occasions.

Maximum Contaminant Level (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

Maximum Contaminant Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (U.S. EPA).

Maximum Residual Disinfectant Level (MRDL): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

Primary Drinking Water Standards (PDWS): MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

Public Health Goal (PHG): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

Regulatory Action Level (AL): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

Secondary Drinking Water Standards (SDWS): MCLs for contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.

Treatment Technique (TT): A required process intended to reduce the level of a contaminant in drinking water.

Variances and Exemptions: Permissions from the State Water Resources Control Board (State Board) to exceed an MCL or not comply with a treatment technique under certain conditions.

ND: not detectable at testing limit

ppm: parts per million or milligrams per liter (mg/L)

ppb: parts per billion or micrograms per liter (µg/L)

ppt: parts per trillion or nanograms per liter (ng/L)

ppq: parts per quadrillion or picogram per liter (pg/L)

pCi/L: picocuries per liter (a measure of radiation)

Sources of Drinking Water and Contaminants that May Be Present in Source Water: The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include:

- *Microbial contaminants*, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- *Inorganic contaminants*, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- *Pesticides and herbicides*, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- Organic chemical contaminants, including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems.
- Radioactive contaminants, that can be naturally-occurring or be the result of oil and gas production and mining activities.

Regulation of Drinking Water and Bottled Water Quality: In order to ensure that tap water is safe to drink, the U.S. EPA and the State Board prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. The U.S. Food and Drug Administration regulations and California law also establish limits for contaminants in bottled water that provide the same protection for public health.

About Your Drinking Water Quality - Drinking Water Contaminants Detected: Tables 1, 2, 3, 4, 5, 6 and A list all of the drinking water contaminants that were detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The State Board allows us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of the data, though representative of the water quality, are more than one year old. Any violation of an AL, MCL, MRDL, or TT is asterisked. Additional information regarding the violation is provided later in this report.

TABLE 1 – SAMPLING RESULTS SHOWING THE DETECTION OF COLIFORM BACTERIA								
Microbiological Contaminants	Highest Number of Detections	No. of months in violation	MCL	MCLG	Typical Source of Bacteria			
E. Coli	(In the year 2024)	0	(a)	0	Human and animal fecal waste			

(a) routine and repeat samples are total coliform-positive and either is e. Coli-positive or system fails to take repeat samples following E. Coli-positive routine sample or system fails to analyze total coliform-positive repeat sample for E. Coli.

	TABLE 2 – SAMPLING RESULTS SHOWING THE DETECTION OF LEAD AND COPPER									
Lead and Copper	Sample Date	No. of Samples Collected	90 th Percentile Level Detected	No. Sites Exceeding AL	Range of Detections	AL	PHG	Typical Source of Contaminant		
Lead (ppb)	(Sept 15, 2023)	5	Not Detectable at testing limit	0	ND at all 5 sites sampled	15	0.2	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits		
Copper (ppm)	(Sept 15, 2023)	5	100	0	60 - 100	1.3	0.3	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives		

	TABLE 3 –	SAMPLING	RESULTS FO	R SOD	IUM ANI) HARDN	IESS
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MC		PHG MCLG)	Typical Source of Contaminant
Sodium (ppm)	(2015)	9	n/a	Non	е	None	Salt present in the water and is generally naturally occurring
Hardness (ppm)	(2015)	119	n/a	Non		None	Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring
TABLE 4 – D	ETECTION OF	CONTAMIN	NANTS WITH	A PRIN	<u>MARY</u> DR	INKING	WATER STANDARD
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MC [MRI		PHG ACLG) IRDLG]	Typical Source of Contaminant
Gross Alpha (pCi/L)	(2019)	0.480	0.130 – 1.16	15		(0)	Erosion of natural deposits
Hexavalent Chromium (ug/L)	Sampling was performed in February 2025	1.2	n/a	The MC: was reinsta Octob 1, 20	e L s atted	0.02	Erosion of natural deposits; transformation of naturally occurring trivalent chromium to hexavalent chromium by natural processes and human activities such as discharges from electroplating factories, leather tanneries, wood preservation, chemical synthesis, refractory production, and textile manufacturing facilities.
Nitrate (mg/L)	(2024)	0.7	n/a	10		10	Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits.
Uranium (pCi/L)	(2019)	0.48	ND-1.92	20		0.43	erosion of natural deposits.
TABLE 5 – DE	TECTION OF (CONTAMINA	NTS WITH A	SECO	NDARY D	RINKIN	G WATER STANDARD
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical	Sources of Contaminant
Chloride (mg/L)	(2015)	3	n/a	500	n/a		eaching from natural deposits; r influence
Iron (ug/L)	(2015)	110	n/a	300	n/a		g from natural deposits; al wastes
Specific Conductance	(2015)	252	n/a	1600	n/a		ces that form ions when in

Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Sources of Contaminant
Chloride (mg/L)	(2015)	3	n/a	500	n/a	Runoff/leaching from natural deposits; seawater influence
Iron (ug/L)	(2015)	110	n/a	300	n/a	Leaching from natural deposits; Industrial wastes
Specific Conductance (umhos/cm)	(2015)	252	n/a	1600	n/a	Substances that form ions when in water; seawater influence
Sulfate (mg/L)	(2015)	3	n/a	500	n/a	Runoff/leaching from natural deposits; industrial wastes
Total Dissolved Solids (mg/L)	(2015)	190	n/a	1000	n/a	Runoff/leaching from natural deposits
Zinc (mg/L)	(2015)	0.16	n/a	5	n/a	Runoff/leaching from natural deposits

TABLE 6 – DETECTION OF UNREGULATED CONTAMINANTS							
Chemical or Constituent (and reporting units) Sample Date Level Detected			Range of Detections	Notification Level	Health Effects Language		
Chloromethane (ug/L)	(2021)	0.66	ND - 2	At or above Detection Limit for Reporting 0.5 ug/L	UCMR 3 List 1 Assessment Monitoring		
Vanadium (ug/L)	(2020)	27	n/a	50	Vanadium exposures resulted in developmental and reproductive effects in rats.		

TABLE 7 - VIOLATION OF A MCL, MRDL, AL, TT OR MONITORING REPORTING REQUIREMENT

NONE IN 2024

TABLE 8 – SAMPLING RESULTS SHOWING FECAL INDICATOR-POSITIVE GROUNDWATER SOURCE SAMPLES

NONE IN 2024

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Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	Notification Level	Typical Sources of Contaminant
Calcium (mg/L)	(2015)	23	n/a	n/a	n/a
Magnesium (mg/L)	(2015)	15	n/a	n/a	n/a
pH (units)	(2015)	6.7	n/a	n/a	n/a
Alkalinity (mg/L)	(2015)	110	n/a	n/a	n/a
Aggressiveness Index	(2015)	10.5	n/a	n/a	n/a
Langelier Index	(2015)	-1.3	n/a	n/a	n/a

Additional General Information on Drinking Water

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained at Safe Drinking Water Information | US EPA https://www.epa.gov/ground-water-and-drinking-water/safe-drinking-water-information.

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. U.S. Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available at https://www.cdc.gov/cryptosporidium/prevention/index.html and https://www.cdc.gov/drinking-water/preventing-waterborne-germs-at-home.html.

Lead-Specific Language: If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. The *FEDEX GROUND WATER SYSTEM* is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you do so, you may wish to collect the flushed water and reuse it for another beneficial purpose, such as watering plants. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available at Basic Information about Lead in Drinking Water | US EPA http://www.epa.gov/safewater/lead.

Examples of Source Water Protection Tips for Consumers

Protection of drinking water is everyone's responsibility. You can help protect your community's drinking water source:

- Eliminate excess use of lawn and garden fertilizers and pesticides they contain hazardous chemicals that can reach your drinking water source.
- Dispose of chemicals properly; take used motor oil to a recycling center.
- Pick up after your pets.
- Learn about Healthy Water Shed Protection https://www.epa.gov/hwp. Find a watershed or wellhead protection organization in your community and volunteer to help. If there are no active groups, consider starting one. Use U.S.
 EPA's
 Adopt
 Your
 Watershed https://nepis.epa.gov/Exe/ZyPDF.cgi/20004I2M.PDF?Dockey=20004I2M.PDF.

Water Conservation Tips for Consumers

Did you know that the average U.S. household uses approximately 400 gallons of water per day or 100 gallons per person per day? Luckily, there are many low-cost and no-cost ways to conserve water. Small changes can make a big difference – try one today and soon it will become second nature.

- Take short showers a 5 minutes shower uses 4 to 5 gallons of water compared to up to 50 gallons for a bath.
- Shut off water while brushing your teeth, washing your hair, and shaving and save up to 500 gallons a month.
- Use a water-efficient showerhead. They are inexpensive, easy to install, and can save you up to 750 gallons a month.
- Run your clothes washer and dishwasher only when they are full. You can save up to 1,000 gallons a month.
- Water plants only when necessary.
- Fix leaking toilets and faucets. Faucet washers are inexpensive and take only a few minutes to replace. To check your toilet for a leak, place a few drops of food coloring in the tank and wait. If it seeps into the toilet bowl without flushing, you have a leak. Fixing it or replacing it with a new, more efficient model can save up to 1,000 gallons a month.
- Adjust sprinklers so only your lawn is watered. Apply water only as fast as the soil can absorb it and during the cooler parts of the day to reduce evaporation.
- Teach your kids about water conservation to ensure a future generation that uses water wisely. Make it a family effort to reduce next month's water bill!
- Visit https://www.epa.gov/watersense for more information.