# **2021** Consumer Confidence Report

Water System Name: FEDEX GROUND CA0409194 Report Date: July 2022

We test the drinking water quality for many constituents as required by state and federal regulations. This report shows the results of our monitoring for the period of January 1 to December 31, 2020 and may include earlier monitoring data.

Language in Spanish: Este informe contiene información muy importante sobre su agua para beber. Favor de comunicarse FEDEX GROUND a (530) 332-2000 para asistirlo en español.

**Type of water source(s) in use:** Groundwater

Name & general location of source(s): Well 01

101 Book Farm Road, Durham, CA 95938

**Drinking Water Source Assessment information:** A Drinking Water Source Assessment for WELL 01 of the FEDEX

GROUND water system has not been completed. Well 01 - does not have a source assessment on file.

#### **Discussion of Vulnerability:**

An assessment summary is not available for the water source.

#### **Acquiring Information:**

For more info you may visit https://www.waterboards.ca.gov/drinking\_water/certlic/drinkingwater/DWSAP.html or contact: Butte County Environmental Health Program Manager 530-552-3880, 530-538-2165 (fax)

Time and place of regularly scheduled board meetings for public participation: Regularly-scheduled County meetings

are held. The State Water Resources Control Board may offer other opportunities.

For more information, contact: Jesse Arensdorf jesse.arensdorf@fedex.com Phone: (530) 332-2000 ext. 2002

#### TERMS USED IN THIS REPORT

**Level 1 Assessment**: A Level 1 assessment is a study of the water system to identify potential problems and determine (if possible) why total coliform bacteria have been found in our water system.

**Level 2 Assessment**: A Level 2 assessment is a very detailed study of the water system to identify potential problems and determine (if possible) why an *E. coli* MCL violation has occurred and/or why total coliform bacteria have been found in our water system on multiple occasions.

Maximum Contaminant Level (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

**Maximum Contaminant Level Goal (MCLG)**: The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (U.S. EPA).

Maximum Residual Disinfectant Level (MRDL): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

**Primary Drinking Water Standards (PDWS)**: MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

**Public Health Goal (PHG)**: The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

**Regulatory Action Level (AL)**: The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**Secondary Drinking Water Standards (SDWS)**: MCLs for contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.

**Treatment Technique (TT)**: A required process intended to reduce the level of a contaminant in drinking water.

**Variances and Exemptions**: Permissions from the State Water Resources Control Board (State Board) to exceed an MCL or not comply with a treatment technique under certain conditions.

**ND**: not detectable at testing limit

**ppm**: parts per million or milligrams per liter (mg/L) **ppb**: parts per billion or micrograms per liter (µg/L)

**ppt**: parts per trillion or nanograms per liter (ng/L)

**ppq**: parts per quadrillion or picogram per liter (pg/L)

pCi/L: picocuries per liter (a measure of radiation)

Sources of Drinking Water and Contaminants that May Be Present in Source Water: The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

#### Contaminants that may be present in source water include:

- *Microbial contaminants*, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- *Inorganic contaminants*, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- *Pesticides and herbicides*, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- Organic chemical contaminants, including synthetic and volatile organic chemicals, that are byproducts of
  industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff,
  agricultural application, and septic systems.
- Radioactive contaminants, that can be naturally-occurring or be the result of oil and gas production and mining activities.

**Regulation of Drinking Water and Bottled Water Quality:** In order to ensure that tap water is safe to drink, the U.S. EPA and the State Board prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. The U.S. Food and Drug Administration regulations and California law also establish limits for contaminants in bottled water that provide the same protection for public health.

**About Your Drinking Water Quality - Drinking Water Contaminants Detected:** Tables 1, 2, 3, 4, 5, 6 and 8 list all of the drinking water contaminants that were detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The State Board allows us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of the data, though representative of the water quality, are more than one year old. Any violation of an AL, MCL, MRDL, or TT is asterisked. Additional information regarding the violation is provided later in this report.

TABLE 1 – SAMPLING RESULTS SHOWING THE DETECTION OF COLIFORM BACTERIA						
Microbiological Contaminants (complete if bacteria detected)	Highest No. of Detections	No. of Months in Violation	MCL MCLG		Typical Source of Bacteria	
Total Coliform Bacteria (state Total Coliform Rule)	(1 in a month)	0	1 positive monthly sample <sup>(a)</sup>	0	Naturally present in the environment	
Fecal Coliform or <i>E. coli</i> (state Total Coliform Rule)	(0 in the year)	0	A routine sample and a repeat sample are total coliform positive, and one of these is also fecal coliform or <i>E. coli</i> positive		Human and animal fecal waste	
E. coli (federal Revised Total Coliform Rule)	(0 in the year)	0	(b)	0	Human and animal fecal waste	

<sup>(</sup>a) Two or more positive monthly samples is a violation of the MCL

(b) Routine and repeat samples are total coliform-positive and either is *E. coli*-positive or system fails to take repeat samples following *E. coli*-positive routine sample or system fails to analyze total coliform-positive repeat sample for *E. coli*.

TABLE 2	TABLE 2 – SAMPLING RESULTS SHOWING THE DETECTION OF LEAD AND COPPER							
Lead and Copper (complete if lead or copper detected in the last sample set)	Sample Date	No. of Samples Collected	90 <sup>th</sup> Percentile Level Detected	No. Sites Exceeding AL	AL	PHG	No. of Schools Requesting Lead Sampling	Typical Source of Contaminant
Lead (ppb)	(2020)	5	0	0	15	0.2	Not applicable	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits
Copper (ppm)	(2020)	5	.286	0	1.3	0.3	Not applicable	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives

TABLE 3 – SAMPLING RESULTS FOR SODIUM AND HARDNESS						
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Source of Contaminant
Sodium (ppm)	(2015)	9	n/a	None	None	Salt present in the water and is generally naturally occurring
Hardness (ppm)	(2015)	119	n/a	None	None	Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring
TABLE 4 – DET	TECTION O	F CONTAMIN	ANTS WITH	A <u>PRIMAI</u>	<u>RY</u> DRINK	ING WATER STANDARD
Chemical or Constituent	Sample	Level	Range of	MCL	PHG (MCLG)	Total Samuel Contact

Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL [MRDL]	PHG (MCLG) [MRDLG	Typical Source of Contaminant
Gross Alpha (pCi/L)	(2019)	0.480	0.130 - 0.475	15	0	Erosion of natural deposits
Nitrate (mg/L)	(2021)	0.8	n/a	10	10	Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits.
Total Radium (pCi/L) (for NT-NC systems)	(2019)	0.028	0 - 0.057	5	(0)	Erosion of natural deposits
Combined Uranium (pCi/L)	(2019)	0.507	0 – 1.920	5	0.43	Erosion of natural deposits

# TABLE 5 – DETECTION OF CONTAMINANTS WITH A <u>SECONDARY</u> DRINKING WATER STANDARD

Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Sources of Contaminant
Chloride (mg/L)	(2015)	3	n/a	500	n/a	Runoff/leaching from natural deposits; seawater influence
Iron (ug/L)	(2015)	110	n/a	300	n/a	Leaching from natural deposits; Industrial wastes
Specific Conductance (umhos/cm)	(2015)	252	n/a	1600	n/a	Substances that form ions when in water; seawater influence
Sulfate (mg/L)	(2015)	3	n/a	500	n/a	Runoff/leaching from natural deposits; industrial wastes
Total Dissolved Solids (mg/L)	(2015)	190	n/a	1000	n/a	Runoff/leaching from natural deposits
Zinc (mg/L)	(2015)	0.16	n/a	5	n/a	Runoff/leaching from natural deposits

## TABLE 6 – DETECTION OF UNREGULATED CONTAMINANTS

Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	Notification Level	Health Effects Language
Chloromethane (ug/L)	(2021)	0.66	ND - 2	DLR = .5	UCMR 3 List 1 Assessment Monitoring
Hexavalent Chromium (ug/L)	(2020)	1.4	n/a	0.02 +	Some people who drink water containing Hexavalent Chromium in excess of the MCL over many years may have an increased risk of getting cancer.
Vanadium (ug/L)	(2020)	27	n/a	50	Vanadium exposures resulted in developmental and reproductive effects in rats.

+ There is currently no MCL for hexavalent chromium. The previous MCL of 10 ug/L was withdrawn on 9/11/17.

TABLE 7 – VIOLATION OF A MCL, MRDL, AL, TT OR MONITORING REPORTING REQUIREMENT

**NONE IN 2021** 

TABLE 8 – SAMPLING RESULTS SHOWING FECAL INDICATOR-POSITIVE GROUNDWATER SOURCE SAMPLES NONE DETECTED

ADDITIONAL DETECTIONS								
Chemical or Constituent (and reporting units)	Sample Date	ample Date Level Detected Range of Detections Notification 1		Notification Level	Typical Sources of Contaminant			
Calcium (mg/L)	(2015)	23	n/a	n/a	n/a			
Magnesium (mg/L)	(2015)	15	n/a	n/a	n/a			
pH (units)	(2015)	6.7	n/a	n/a	n/a			
Alkalinity (mg/L)	(2015)	110	n/a	n/a	n/a			
Aggressiveness Index	(2015)	10.5	n/a	n/a	n/a			
Langelier Index	(2015)	-1.3	n/a	n/a	n/a			

# **Additional General Information on Drinking Water**

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline (1-800-426-4791).

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. U.S. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline (1-800-426-4791).

Lead-Specific Language: If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. The <u>EFEDEX GROUND WATER SYSTEM</u> is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. <u>OPTIONAL</u>: If you do so, you may wish to collect the flushed water and reuse it for another beneficial purpose, such as watering plants. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline (1-800-426-4791) or at <a href="http://www.epa.gov/lead">http://www.epa.gov/lead</a>.

### **Examples of Source Water Protection Tips for Consumers**

Protection of drinking water is everyone's responsibility. You can help protect your community's drinking water source in several ways:

- Eliminate excess use of lawn and garden fertilizers and pesticides they contain hazardous chemicals that can reach your drinking water source.
- Pick up after your pets.
- If you have your own septic system, properly maintain your system to reduce leaching to water sources.
- Dispose of chemicals properly; take used motor oil to a recycling center.
- Volunteer in your community. Find a watershed or wellhead protection organization in your community and volunteer to help. If there are no active groups, consider starting one. Use U.S. EPA's Adopt Your Watershed <a href="https://nepis.epa.gov/Exe/ZyPDF.cgi/20004I2M.PDF?Dockey=20004I2M.PDF">https://nepis.epa.gov/Exe/ZyPDF.cgi/20004I2M.PDF?Dockey=20004I2M.PDF</a> or for Tools and Resources to protect watersheds visit <a href="https://www.epa.gov/hwp/tools-and-resources-protect-watersheds">https://www.epa.gov/hwp/tools-and-resources-protect-watersheds</a>.
- Organize a storm drain stenciling project with your local government. Stencil a message next to the street drain reminding people "Dump No Waste – Drains to River" or "Protect Your Water". Produce and distribute a flyer for households to remind residents that storm drains dump directly into your local water body.